Joseph Lstiburek, Ph.D., P.Eng, ASHRAE Fellow

Building Science

Adventures In Building Science

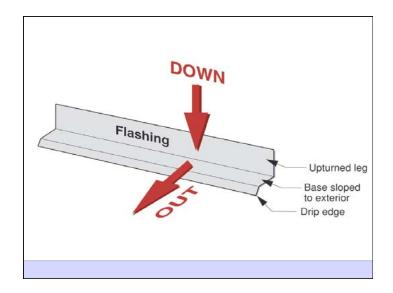
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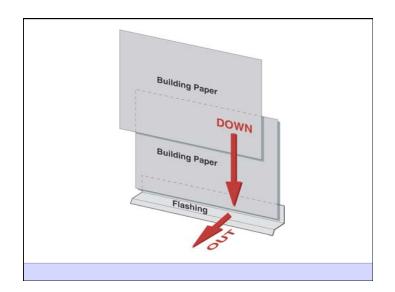


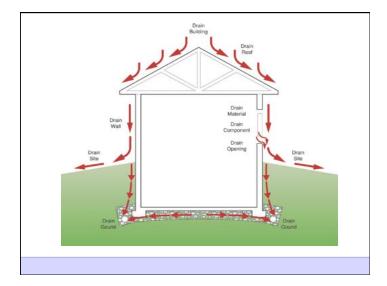


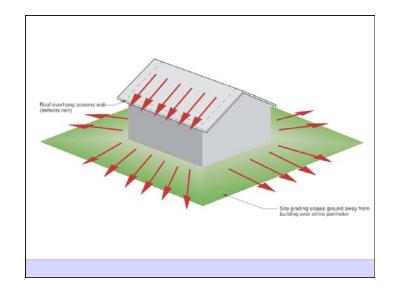


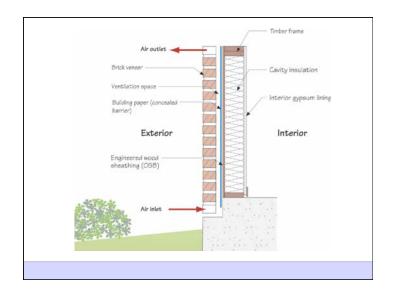










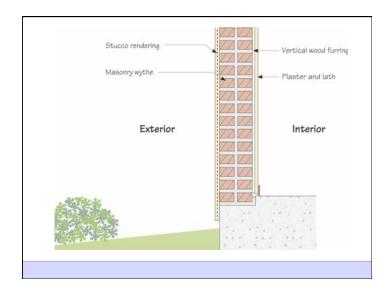








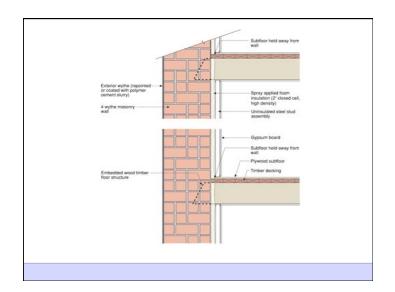


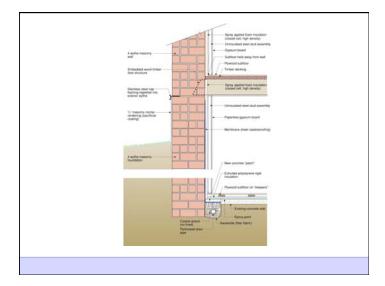


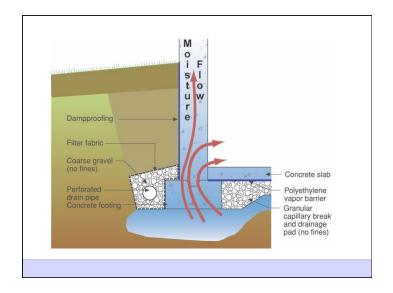


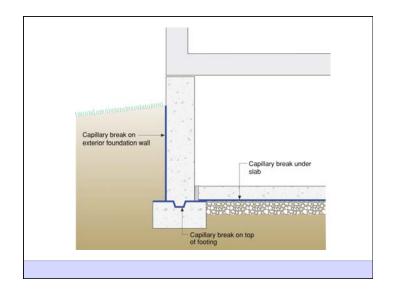


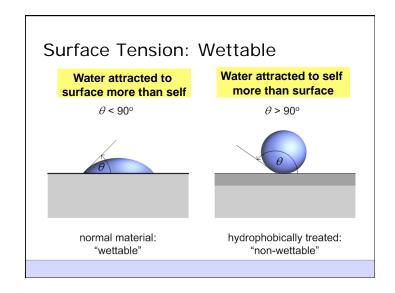


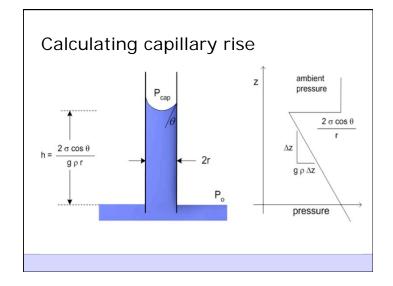


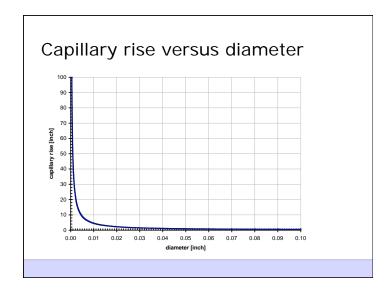


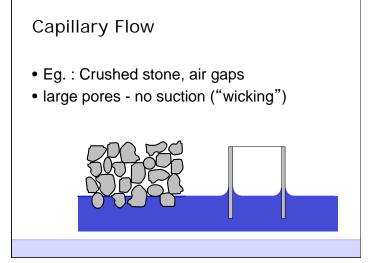


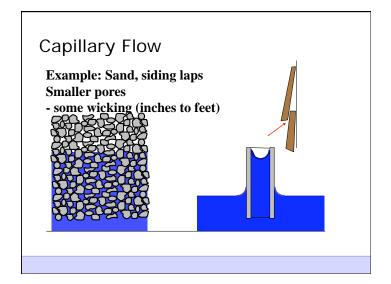






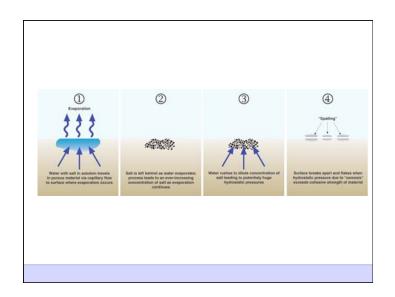






Capillarity + Salt = Osmosis

- · Mineral salts carried in solution by capillary water
- When water evaporates from a surface the salts left behind form crystals in process called efflorescence
- When water evaporated beneath a surface the salts crystallize within the pore structure of the material in called subefflorescence
- The salt crystallization causes expansive forces that can exceed the cohesive strength of the material leading to spalling



Diffusion + Capillarity + Osmosis = Problem

Diffusion Vapor Pressure 3 to 5 psi
 Capillary Pressure 300 to 500 psi
 Osmosis Pressure 3,000 to 5,000 psi



