Joseph Lstiburek, Ph.D., P.Eng. ASHRAE Fellow

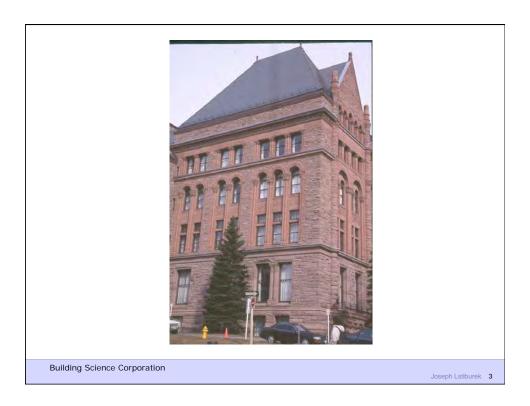
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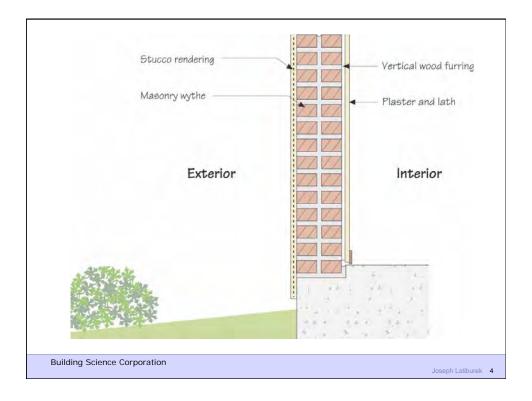
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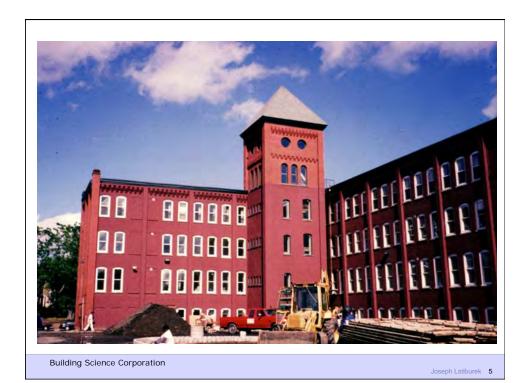
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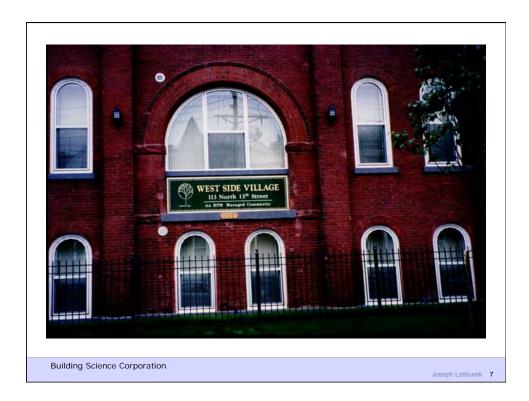
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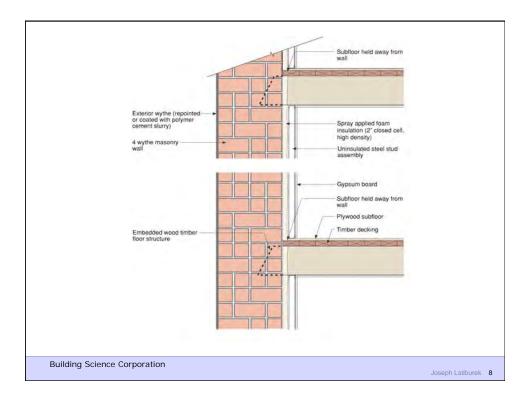


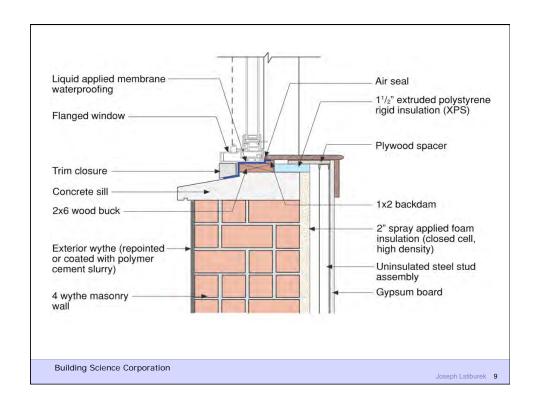


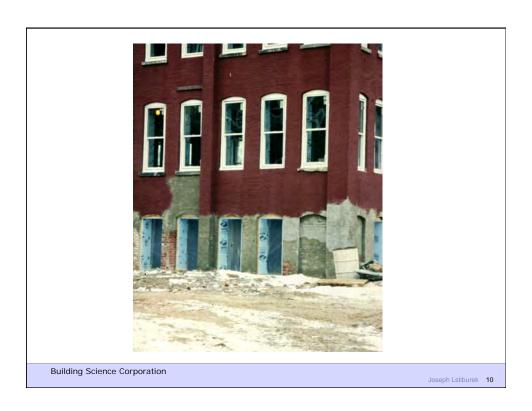






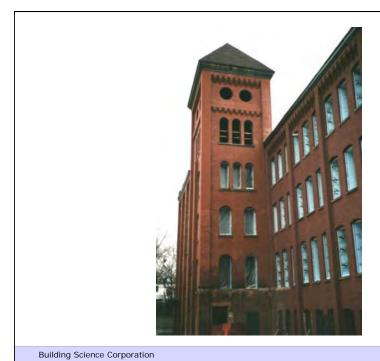




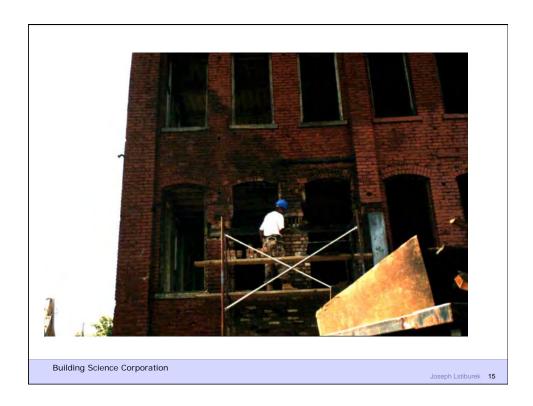








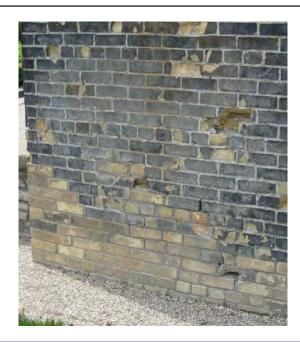




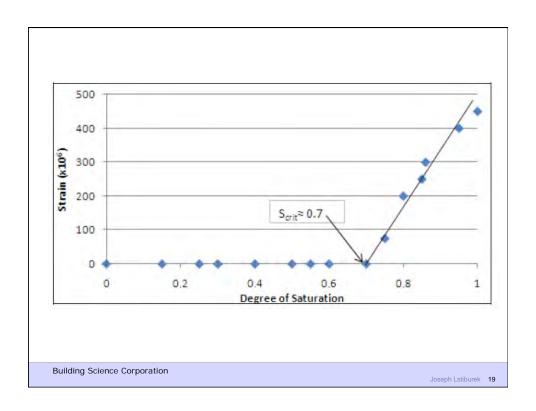


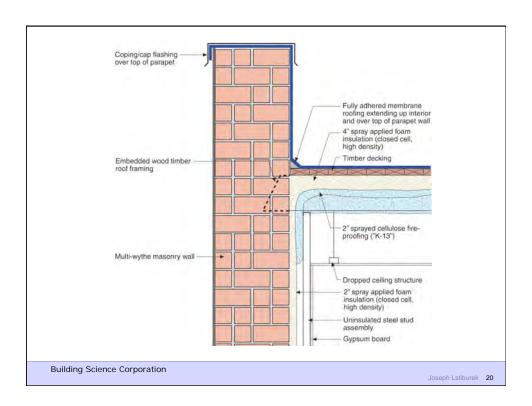


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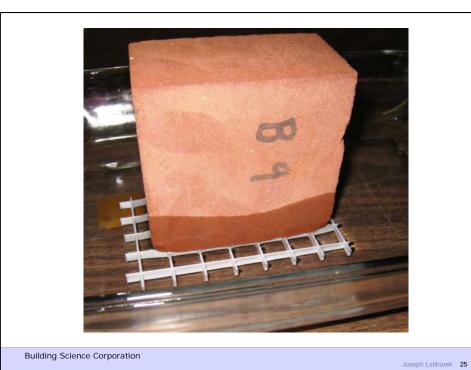


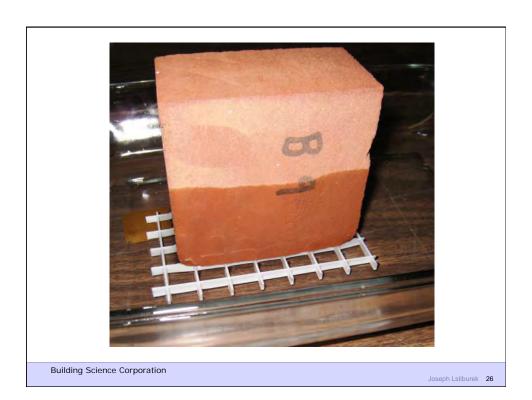


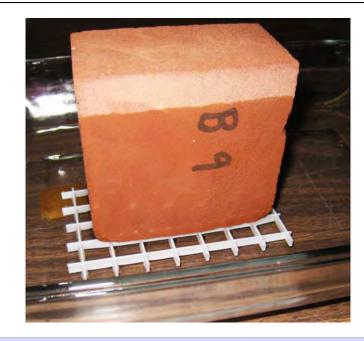
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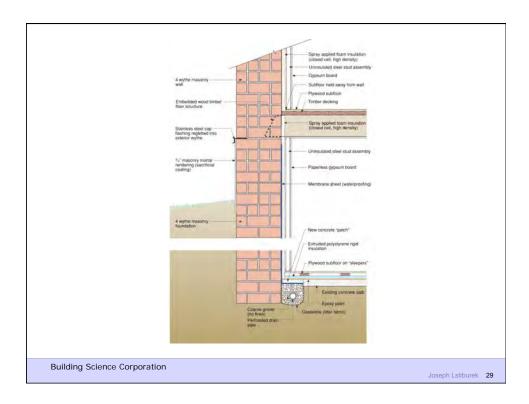
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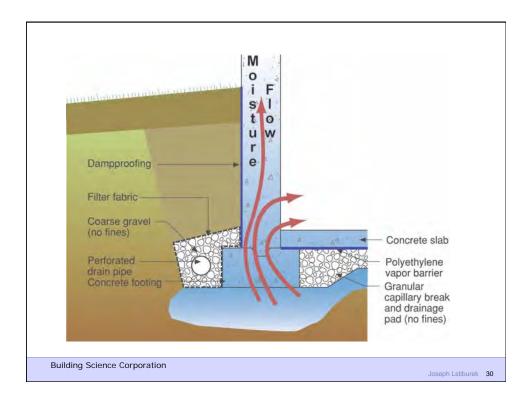


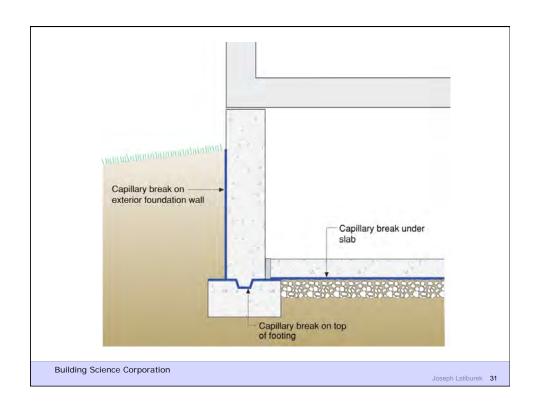










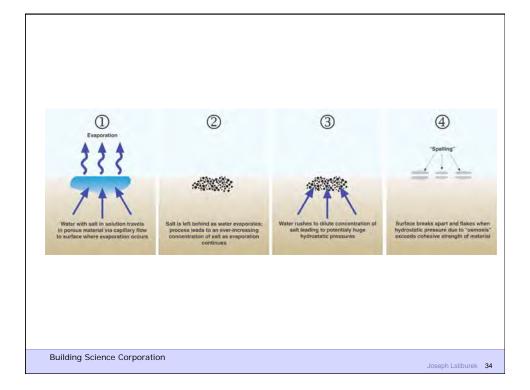




## Capillarity + Salt = Osmosis

- · Mineral salts carried in solution by capillary water
- When water evaporates from a surface the salts left behind form crystals in process called efflorescence
- When water evaporated beneath a surface the salts crystallize within the pore structure of the material in called subefflorescence
- The salt crystallization causes expansive forces that can exceed the cohesive strength of the material leading to spalling

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## Diffusion + Capillarity + Osmosis = Problem

• Diffusion Vapor Pressure

• Capillary Pressure

· Osmosis Pressure

3 to 5 psi

300 to 500 psi

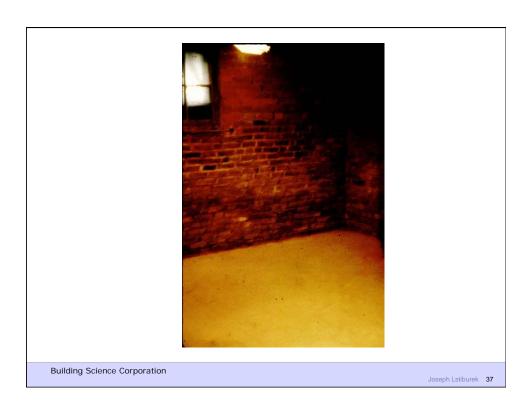
3,000 to 5,000 psi

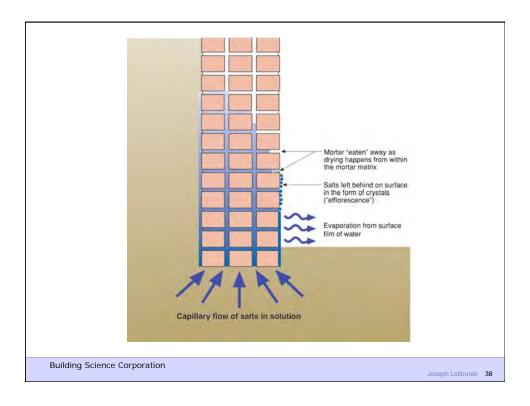
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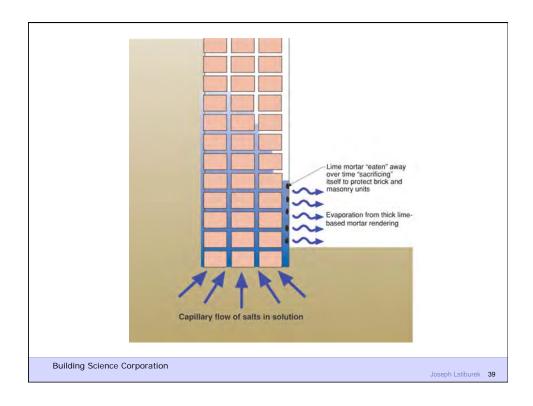
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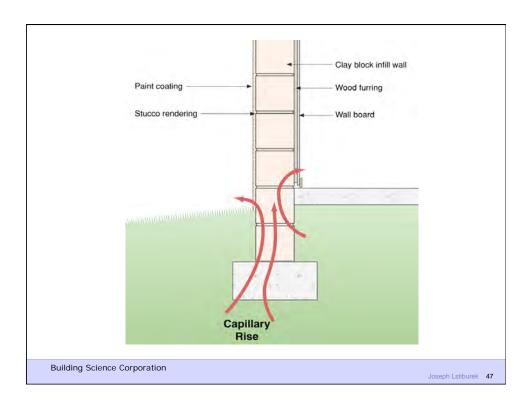


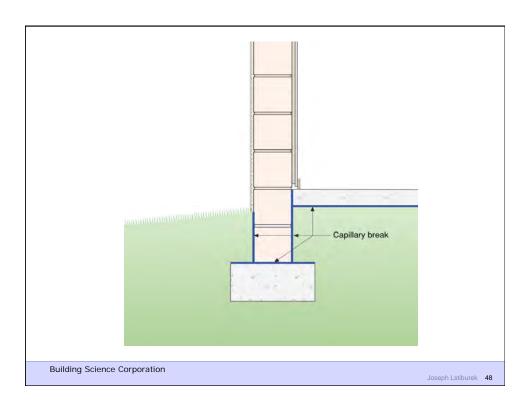


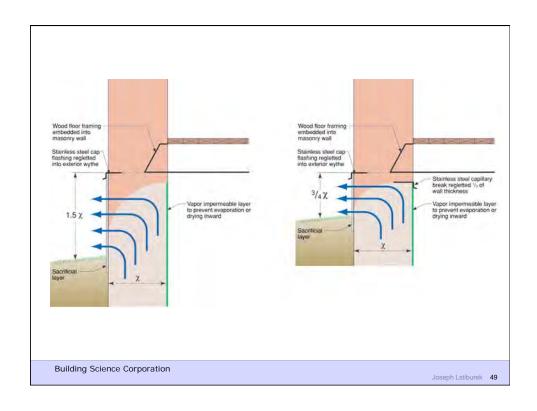


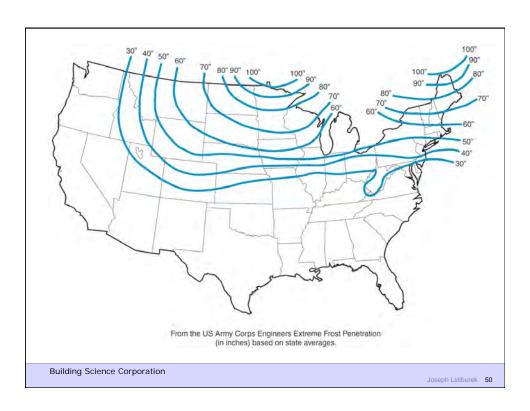


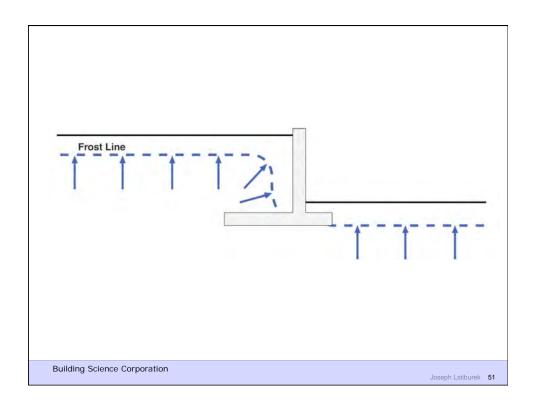


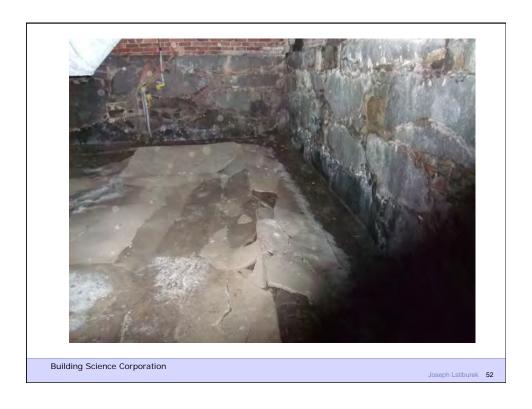


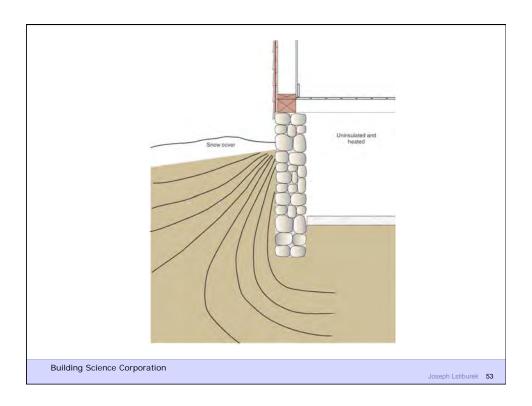


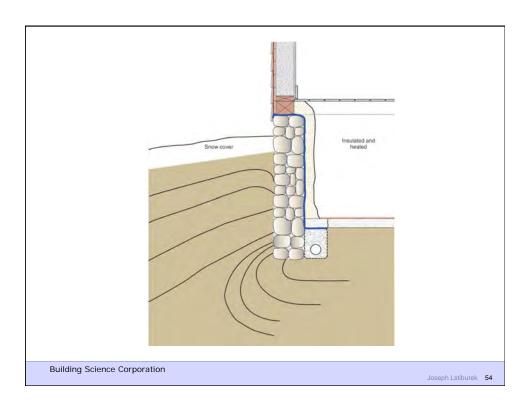


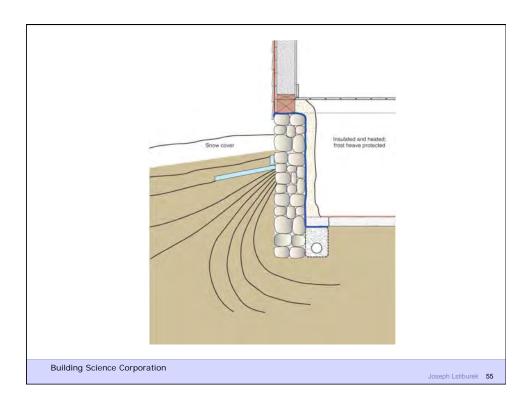


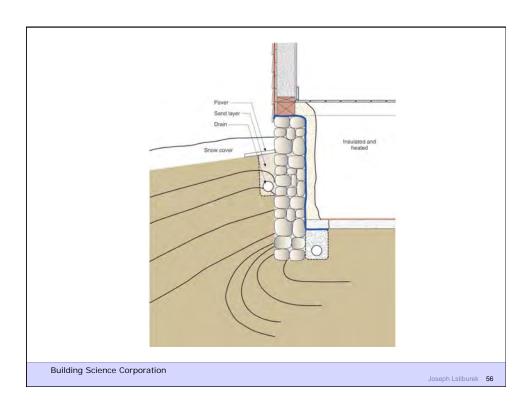


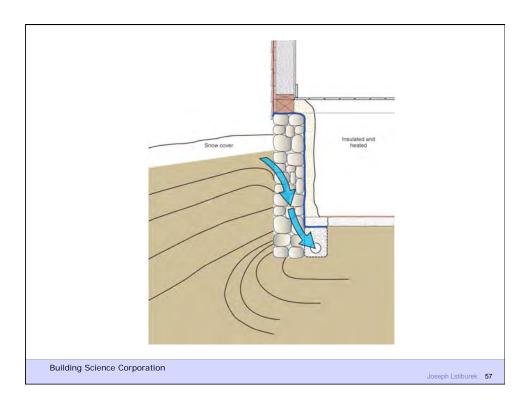


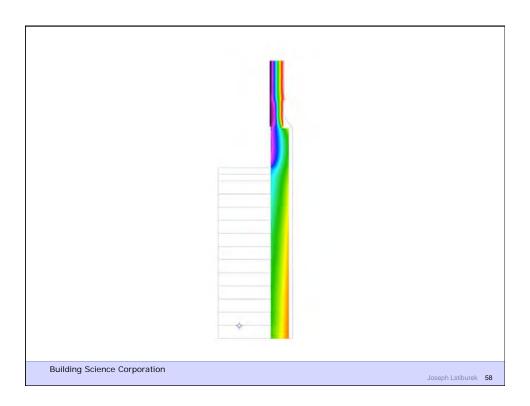


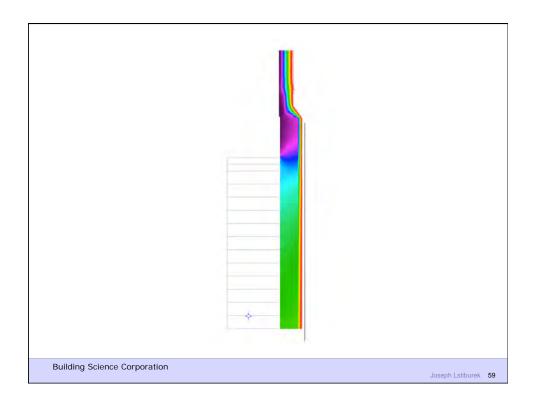


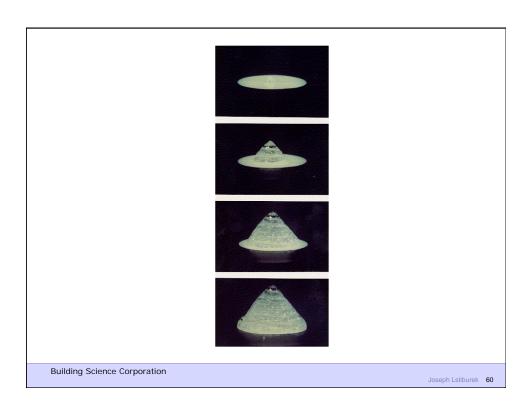






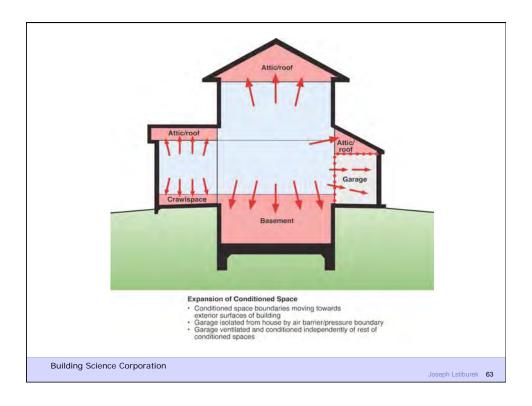


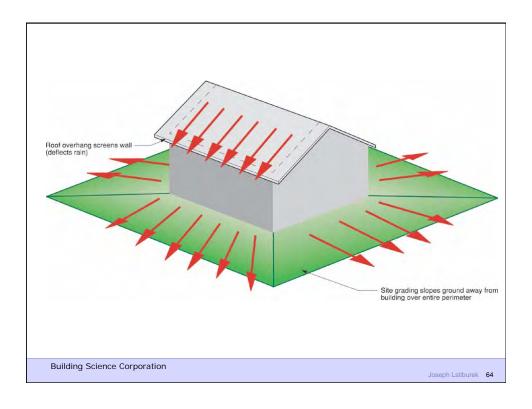


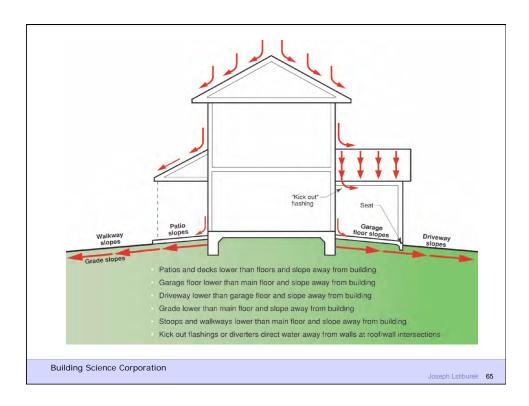


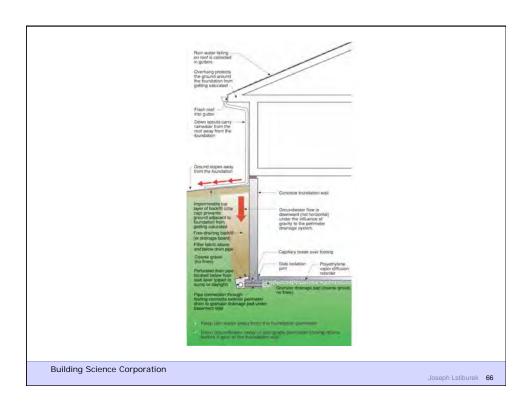


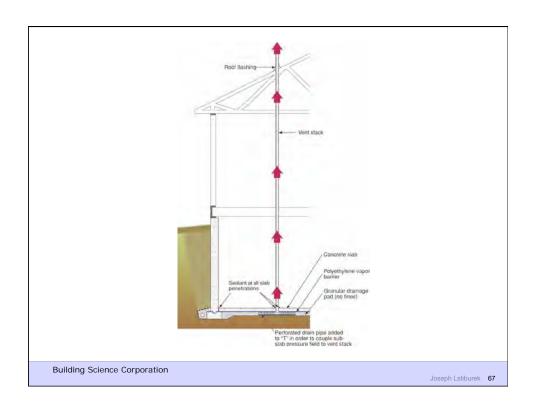


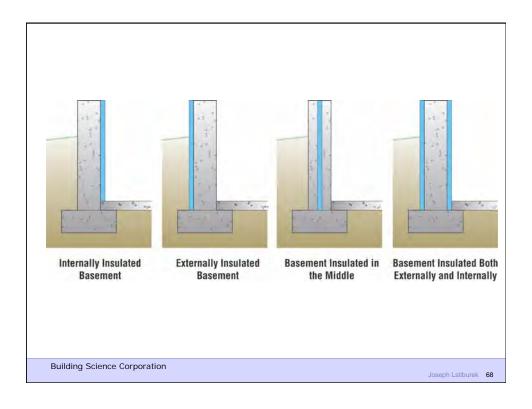


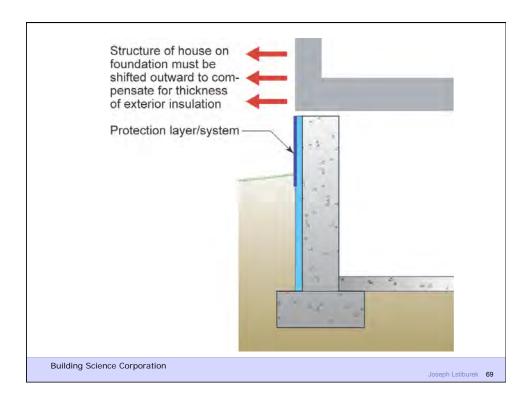














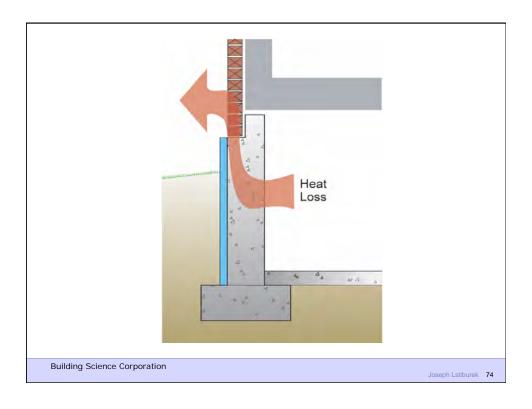


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