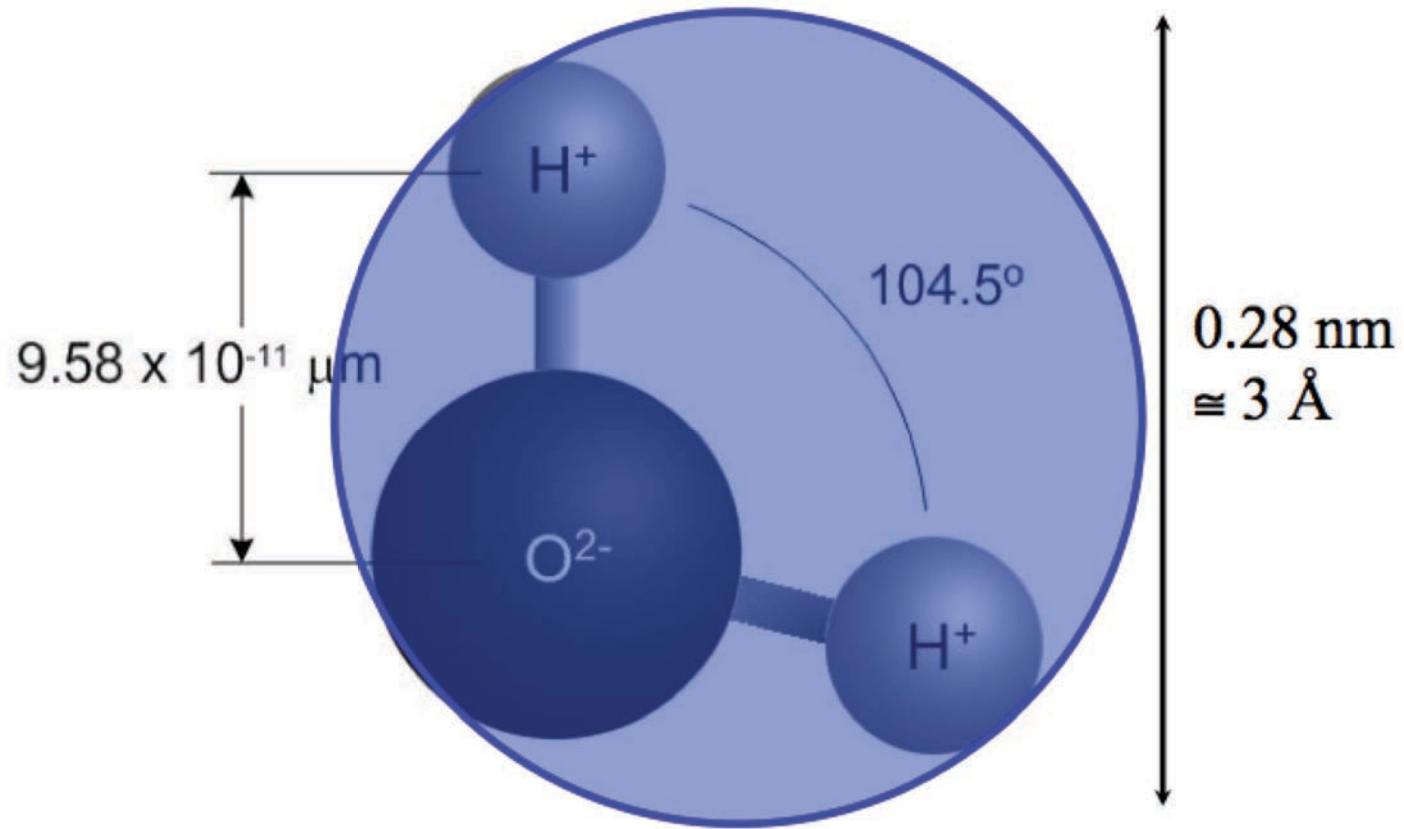


Joseph Lstiburek, Ph.D., P.Eng, ASHRAE Fellow

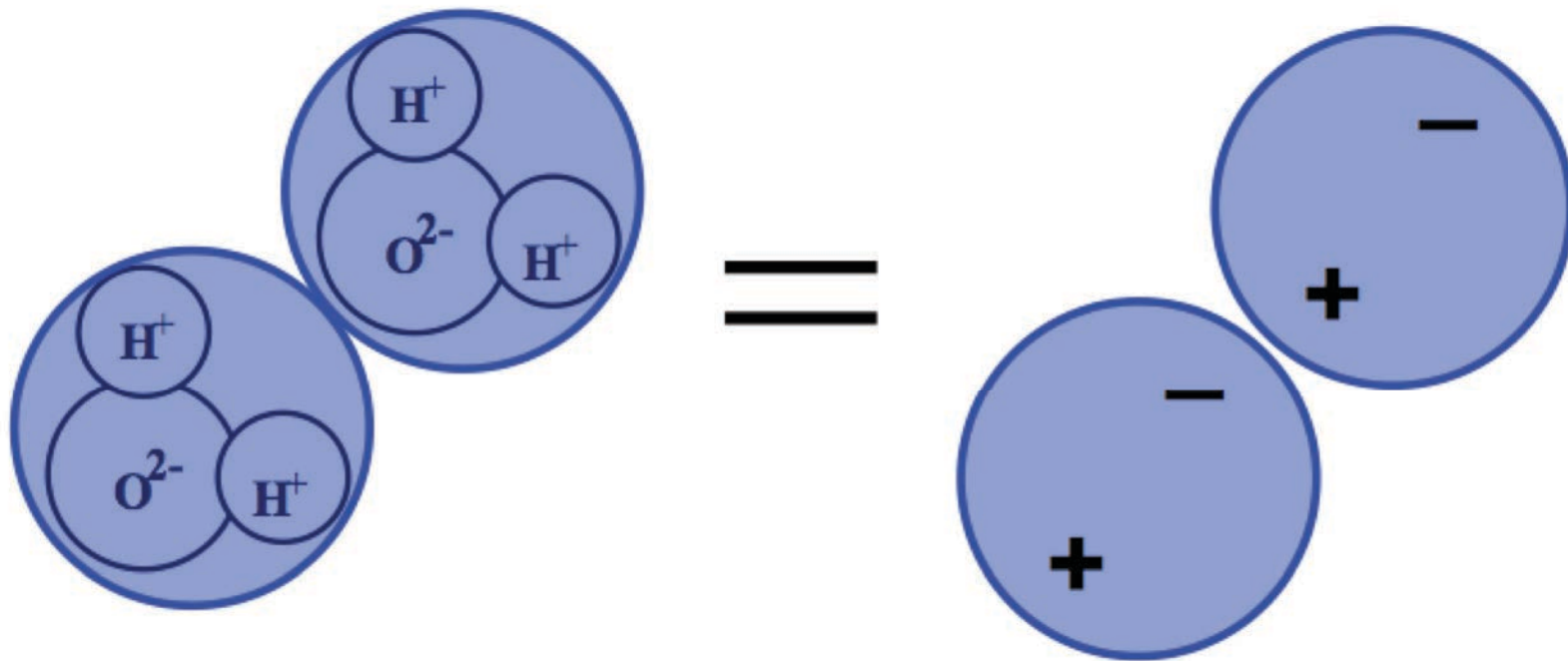
Building Science

The Water Molecule

Water Molecules



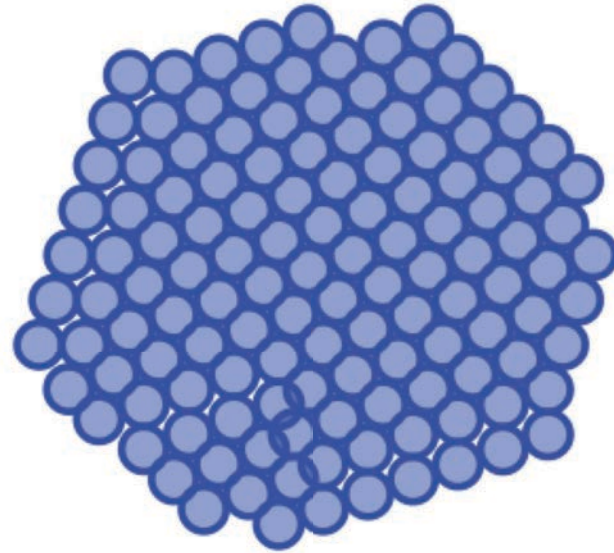
Polar Molecule



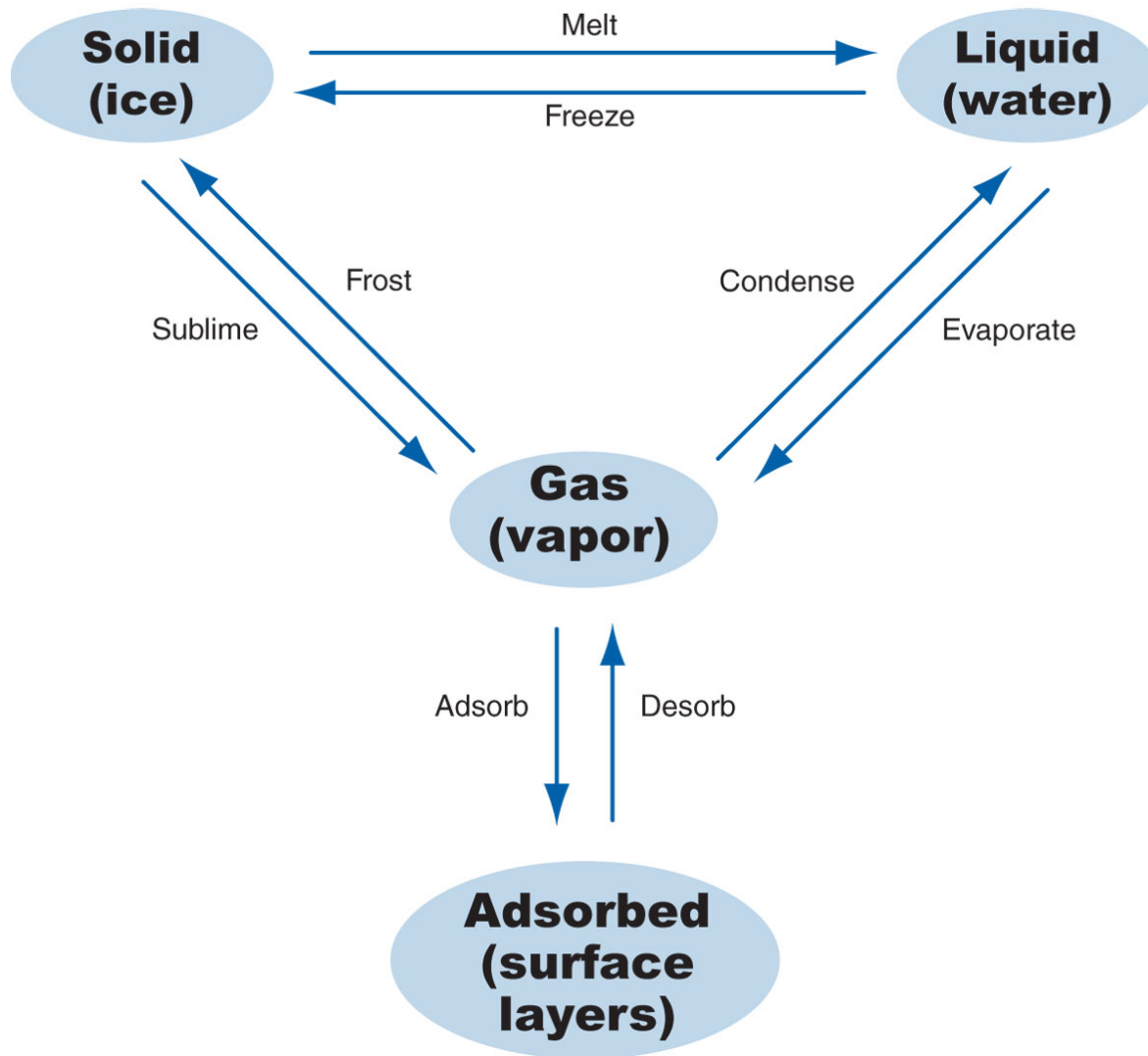
Size Matters



Vapor



Liquid



Moisture Transport in Porous Media

Phase	Transport Process	Driving Potential
Vapor	Diffusion	Vapor Concentration
Adsorbate	Surface Diffusion	Concentration
Liquid	Capillary Flow	Suction Pressure
	Osmosis	Solute Concentration

Moisture Transport in Assemblies

Phase	Transport Process	Driving Potential
Vapor	Diffusion	Vapor Concentration
	Convective Flow	Air Pressure

Adsorbate	Surface Diffusion	Concentration

Liquid	Capillary Flow	Suction Pressure
	Osmosis	Solute Concentration
	Gravitational Flow	Height
	Surface Tension	Surface Energy
	Momentum	Kinetic Energy
	Convective Flow	Air Pressure

Vapor

Diffusion

Convective Flow

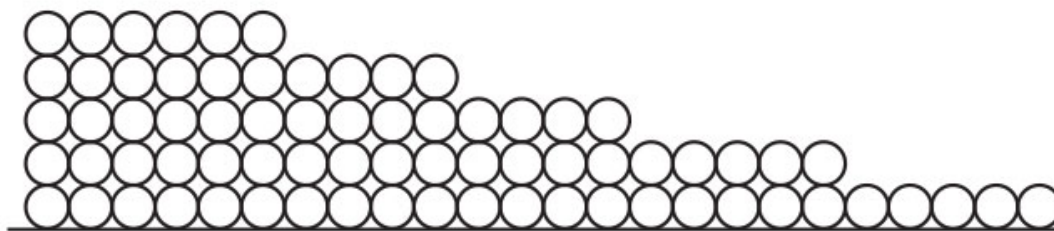
Vapor Concentration

Air Pressure

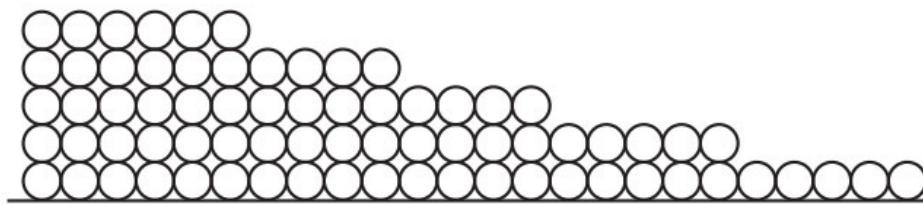
Adsorbate

Surface Diffusion

Concentration



↑
Monolayers of
adsorbed water
increase with
increasing RH



Monolayers
flow along surface
following concentration gradient



Vapor

Diffusion

Convective Flow

Vapor Concentration

Air Pressure

Adsorbate

Surface Diffusion

Concentration

Liquid

Capillary Flow

Suction Pressure

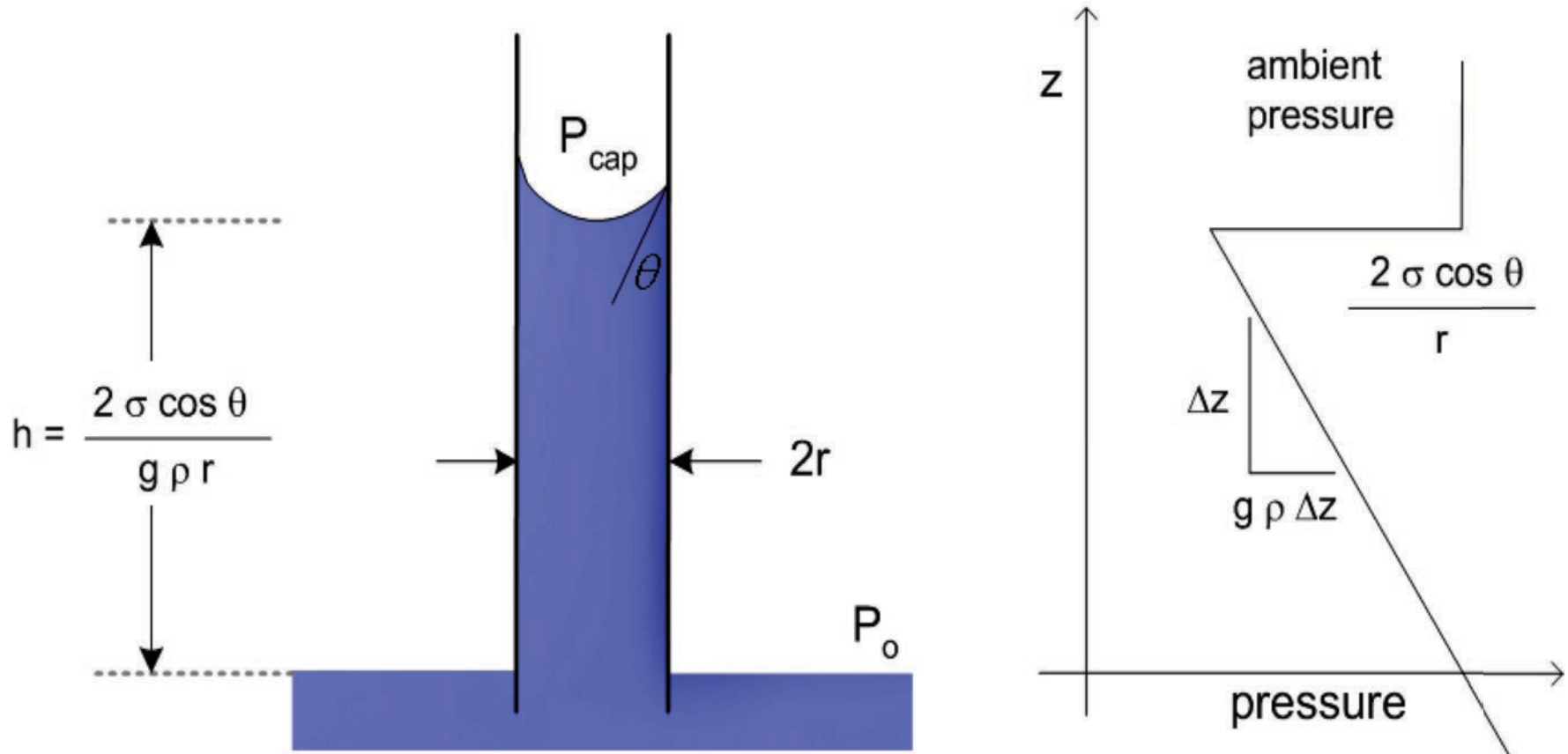
William Thomson

William Thomson – Lord Kelvin

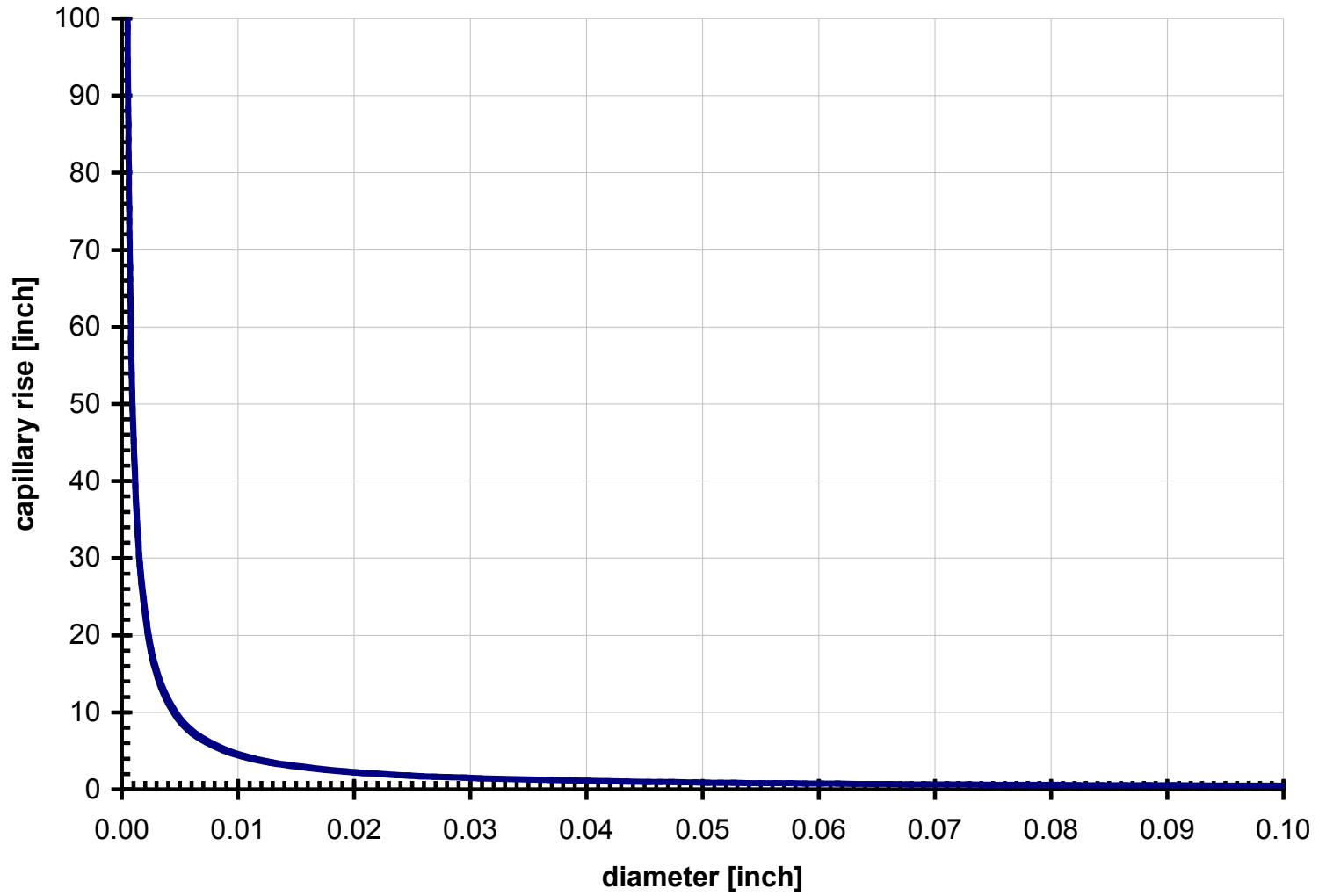
Kelvin Equation

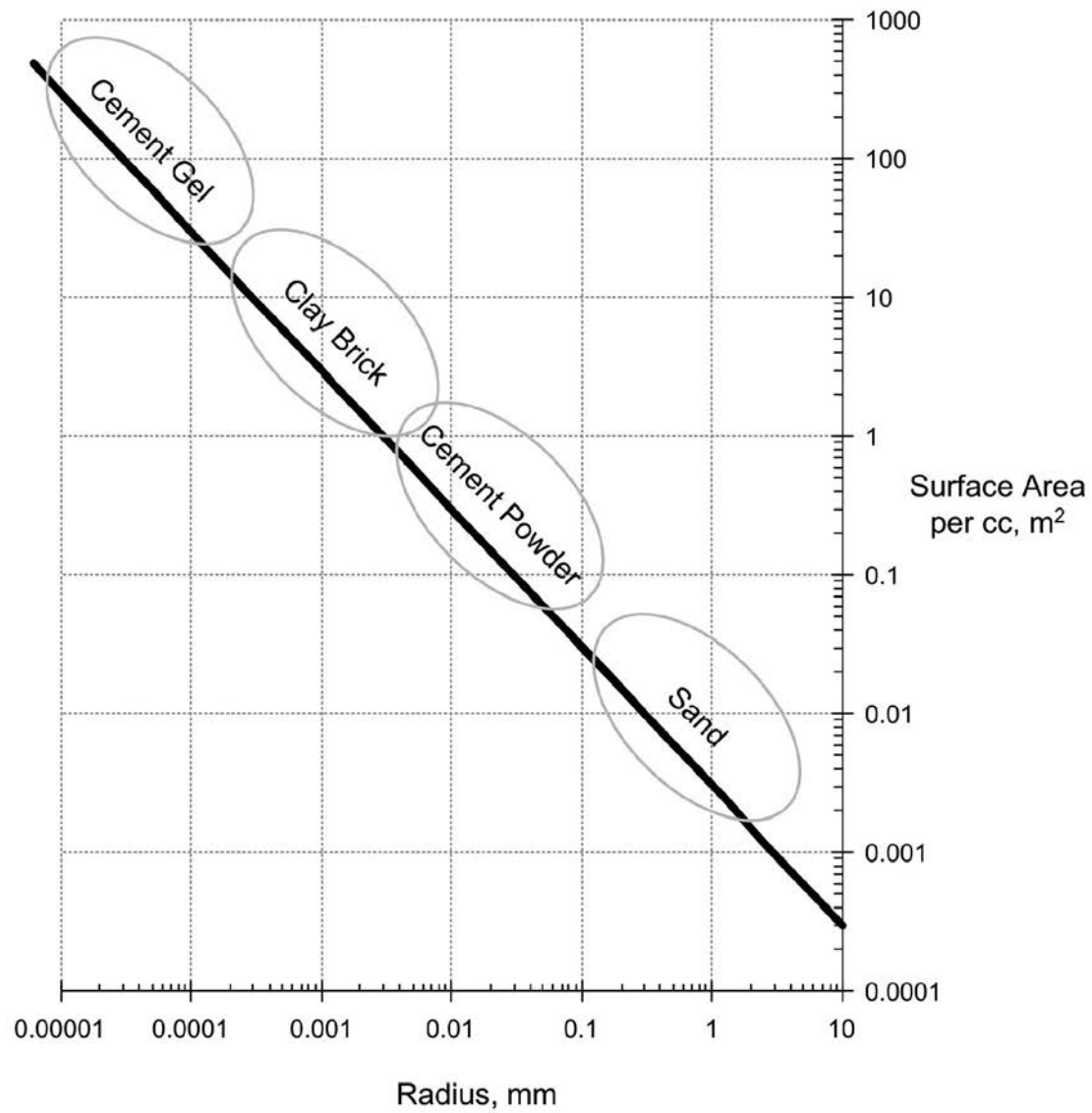
$$\ln \frac{p}{p_0} = \frac{2\gamma V_m}{rRT}$$

Calculating capillary rise



Capillary rise versus diameter





Surface area vs. particle size
From Straube & Burnett, 2005

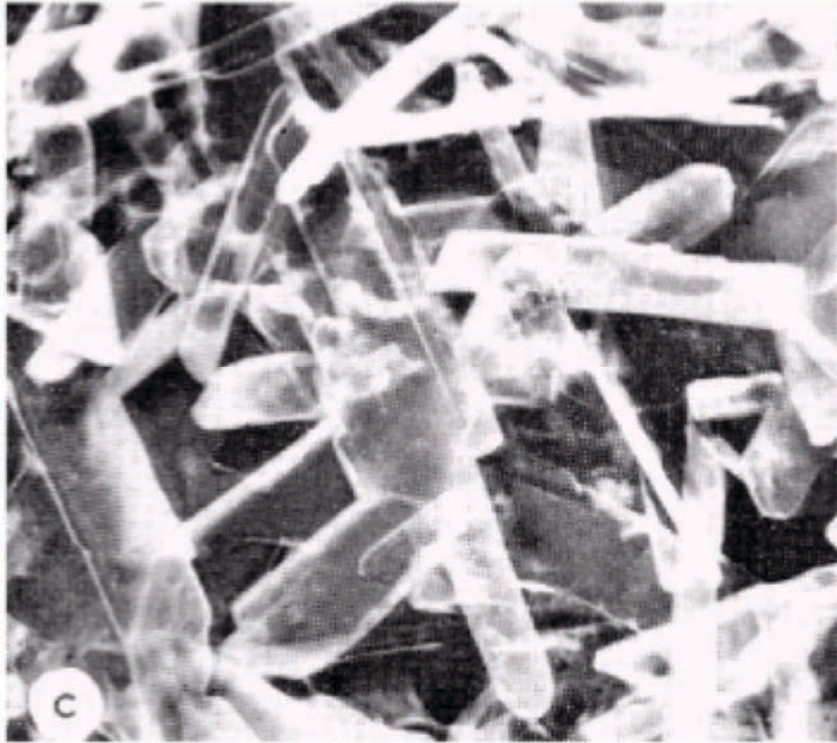


Figure 1c. Gypsum, hydrated from plaster of paris and water, porosity 30 per cent.

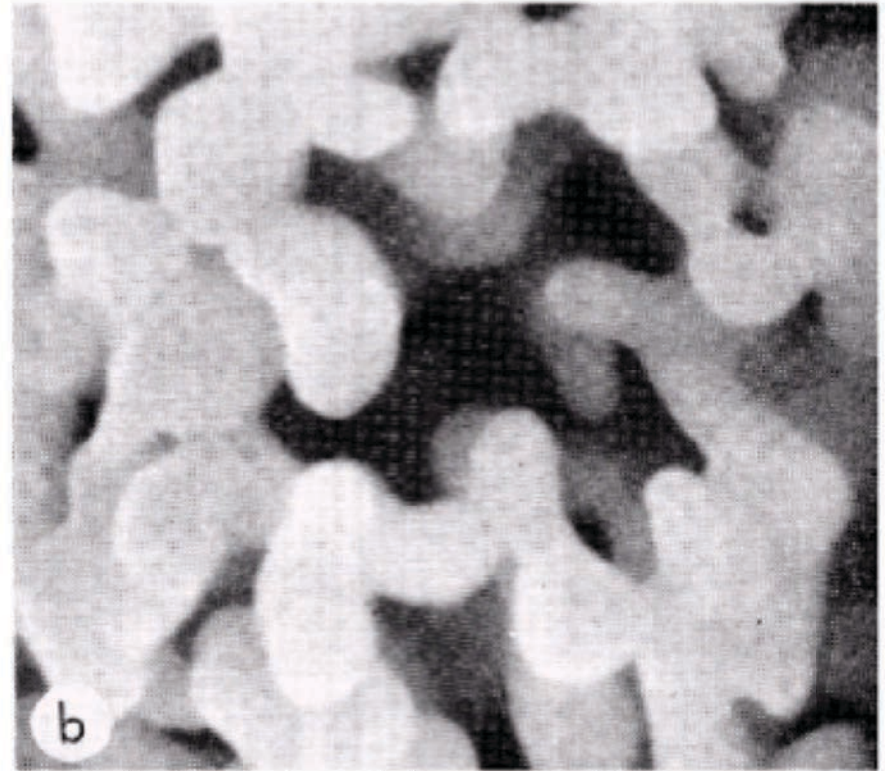
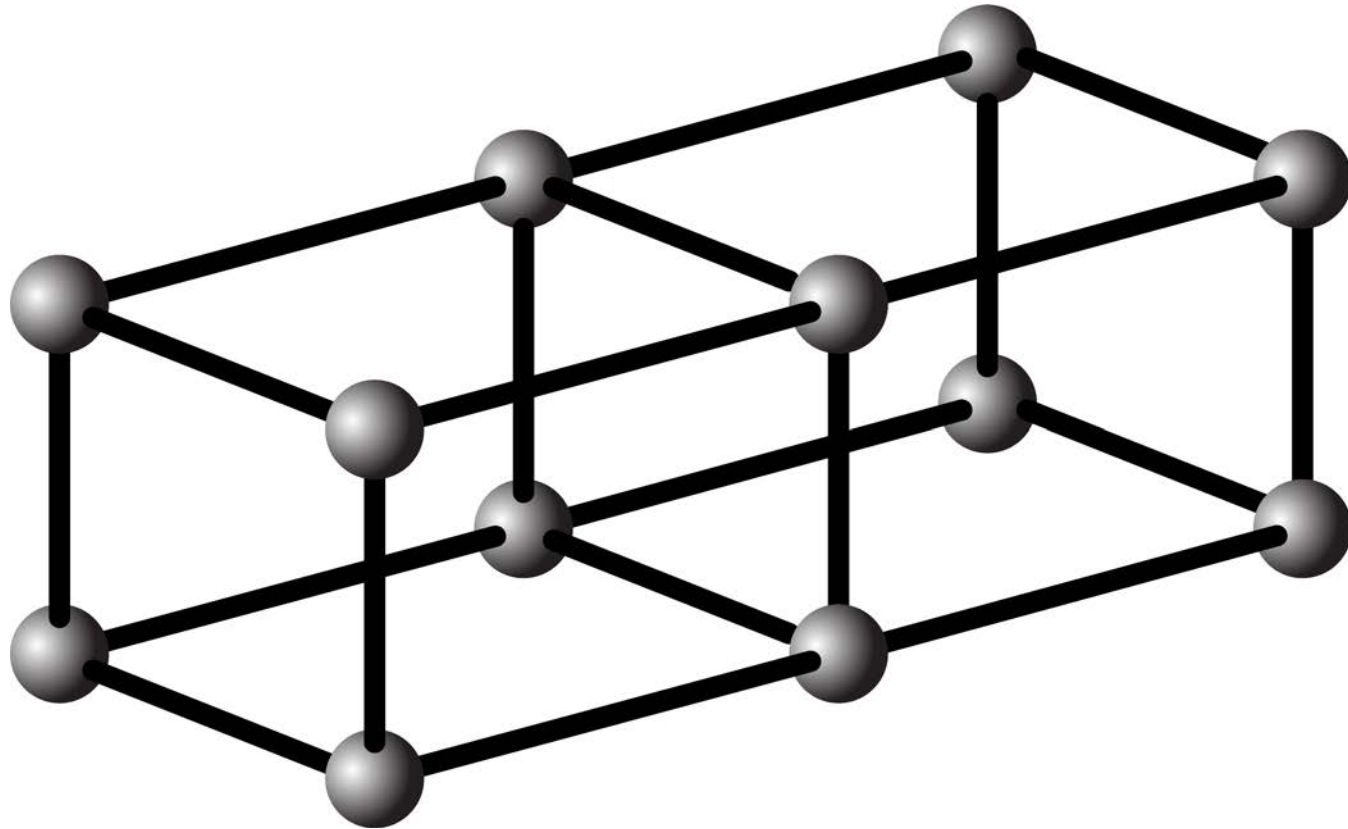
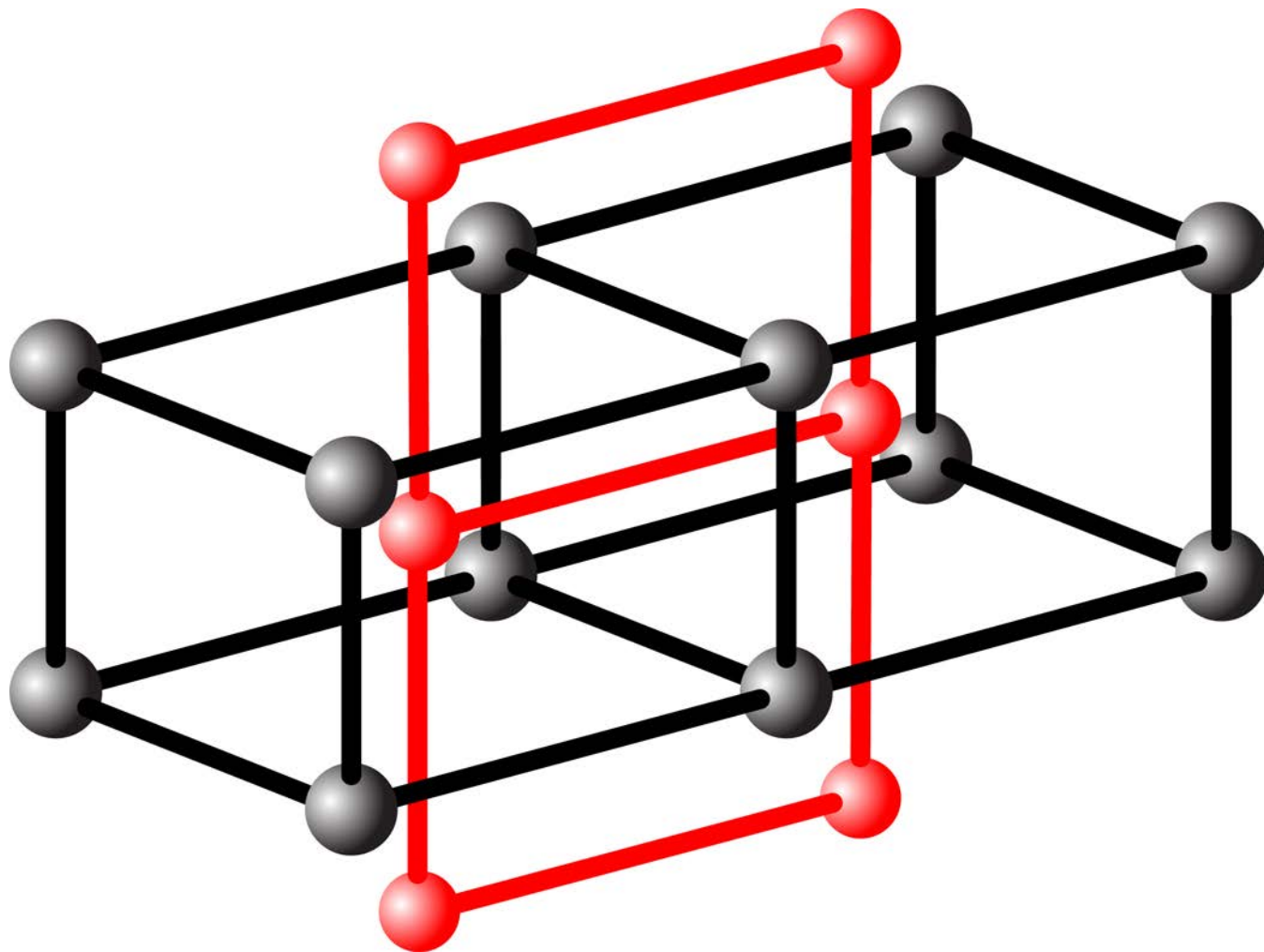
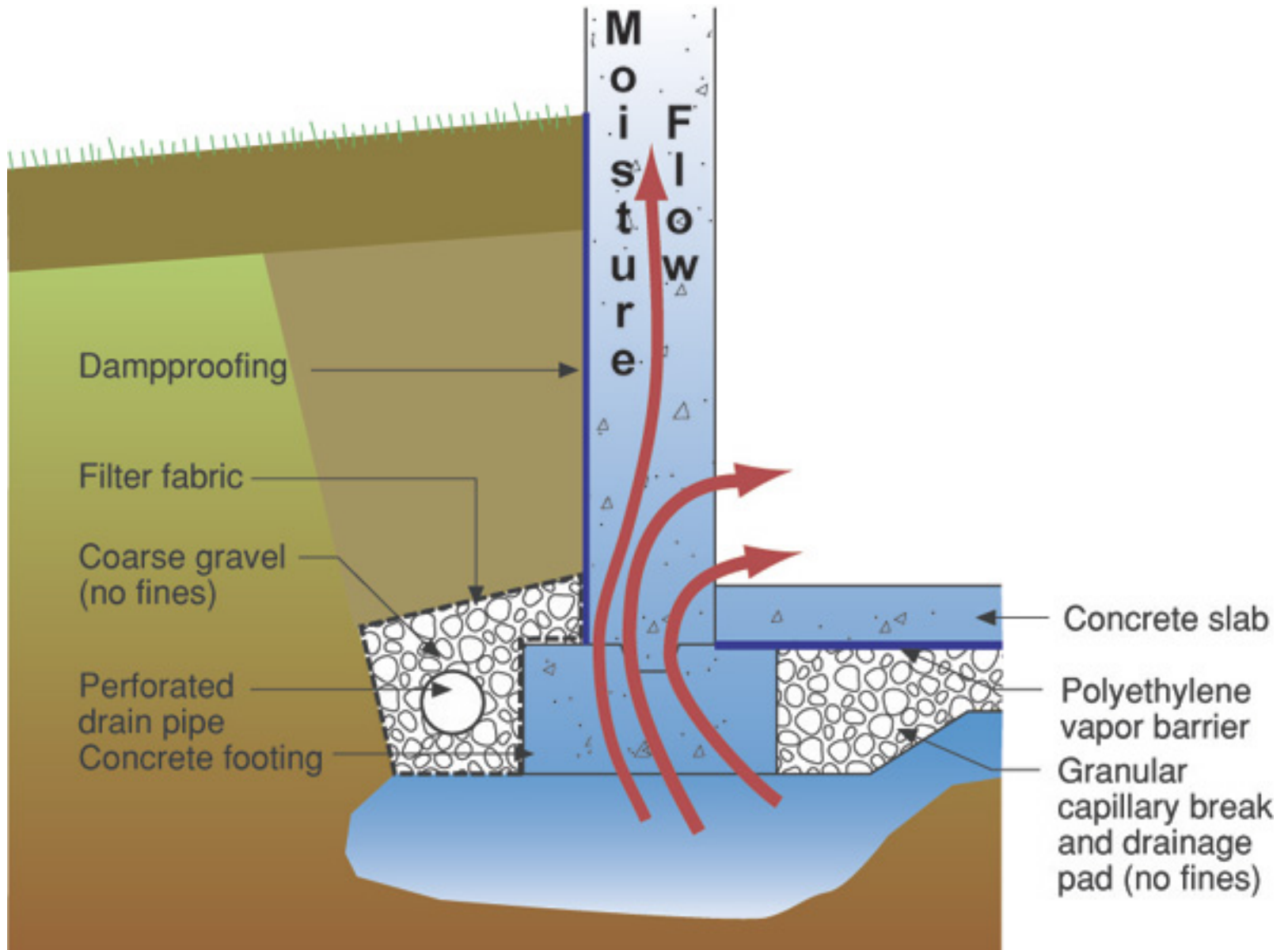
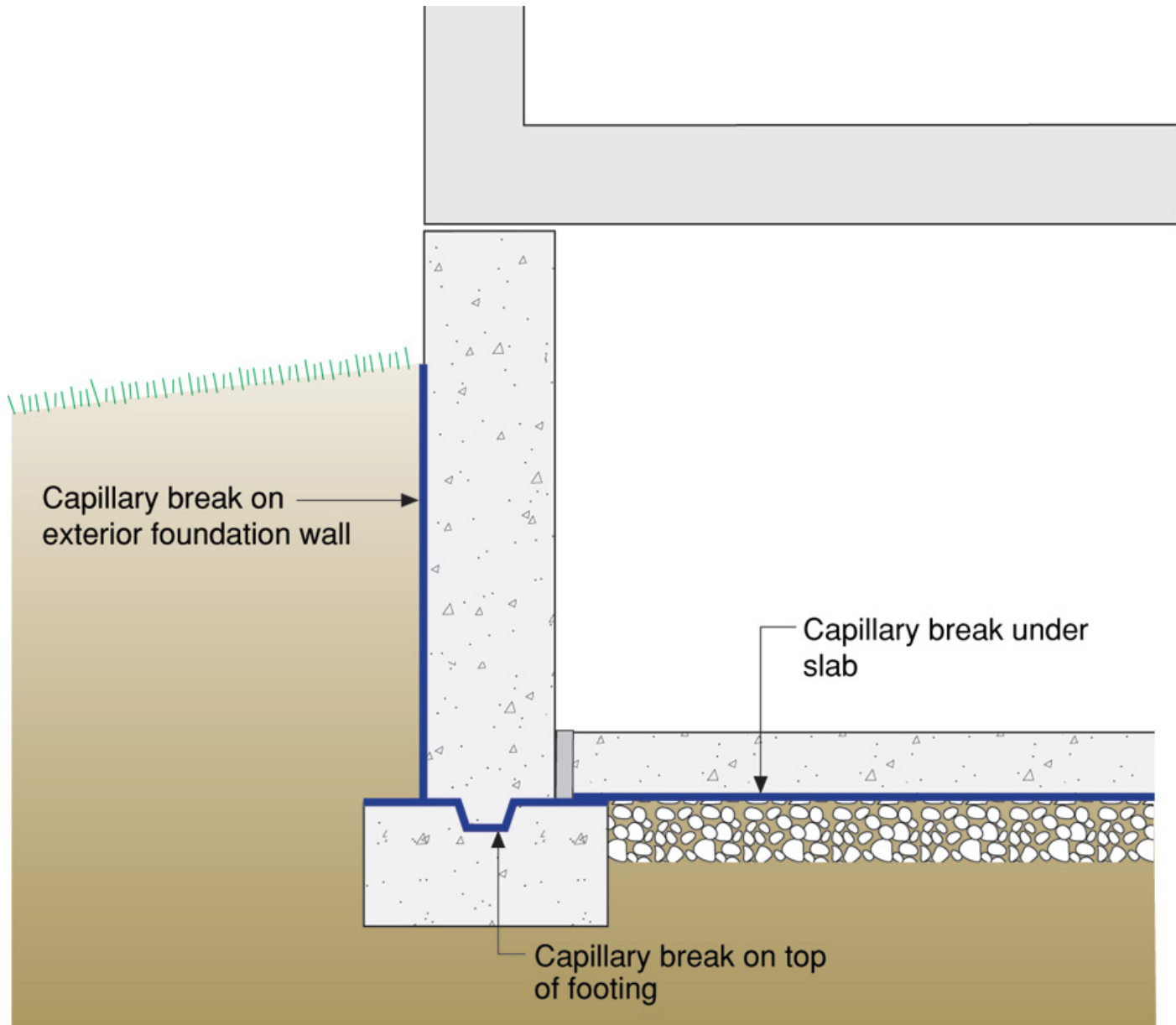


Figure 1b. Brick, sintered clay, porosity 40 per cent.





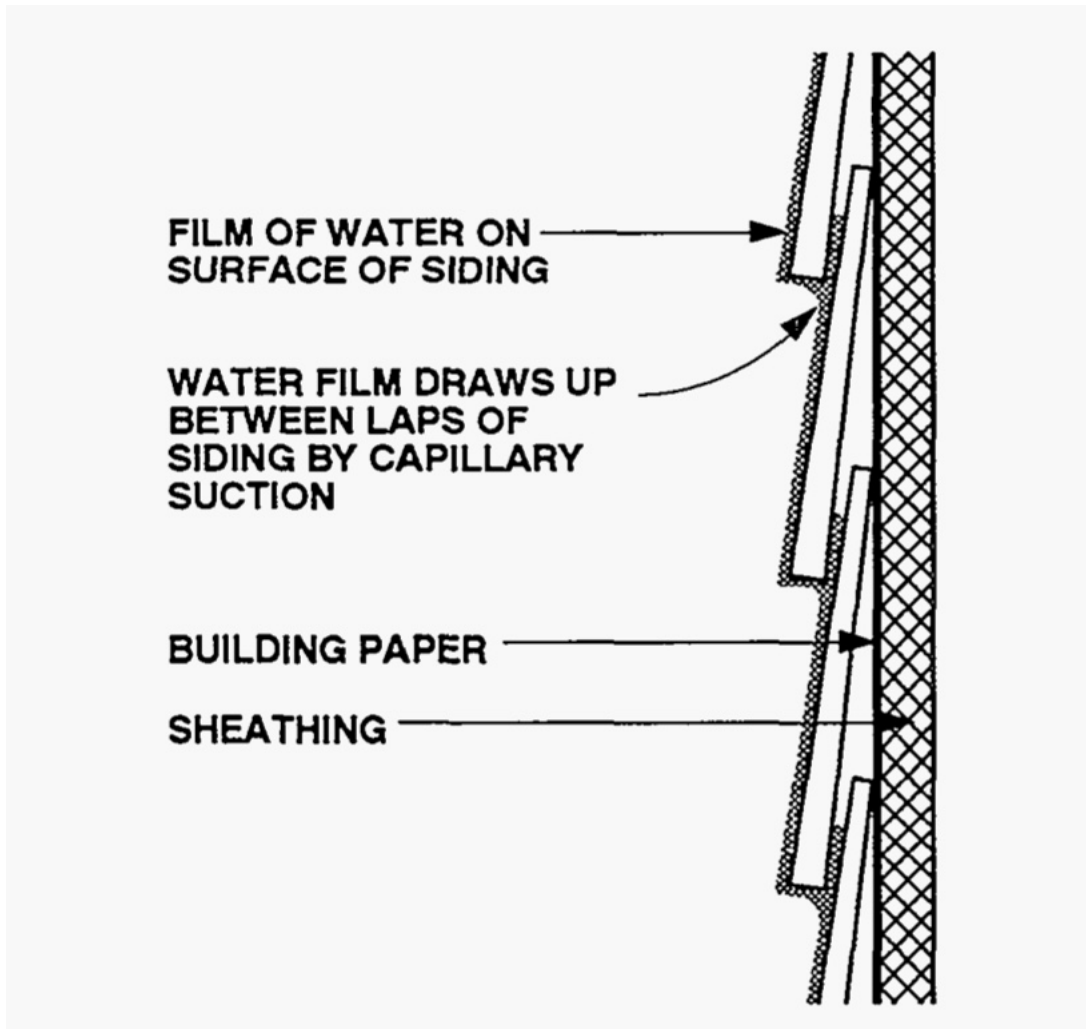




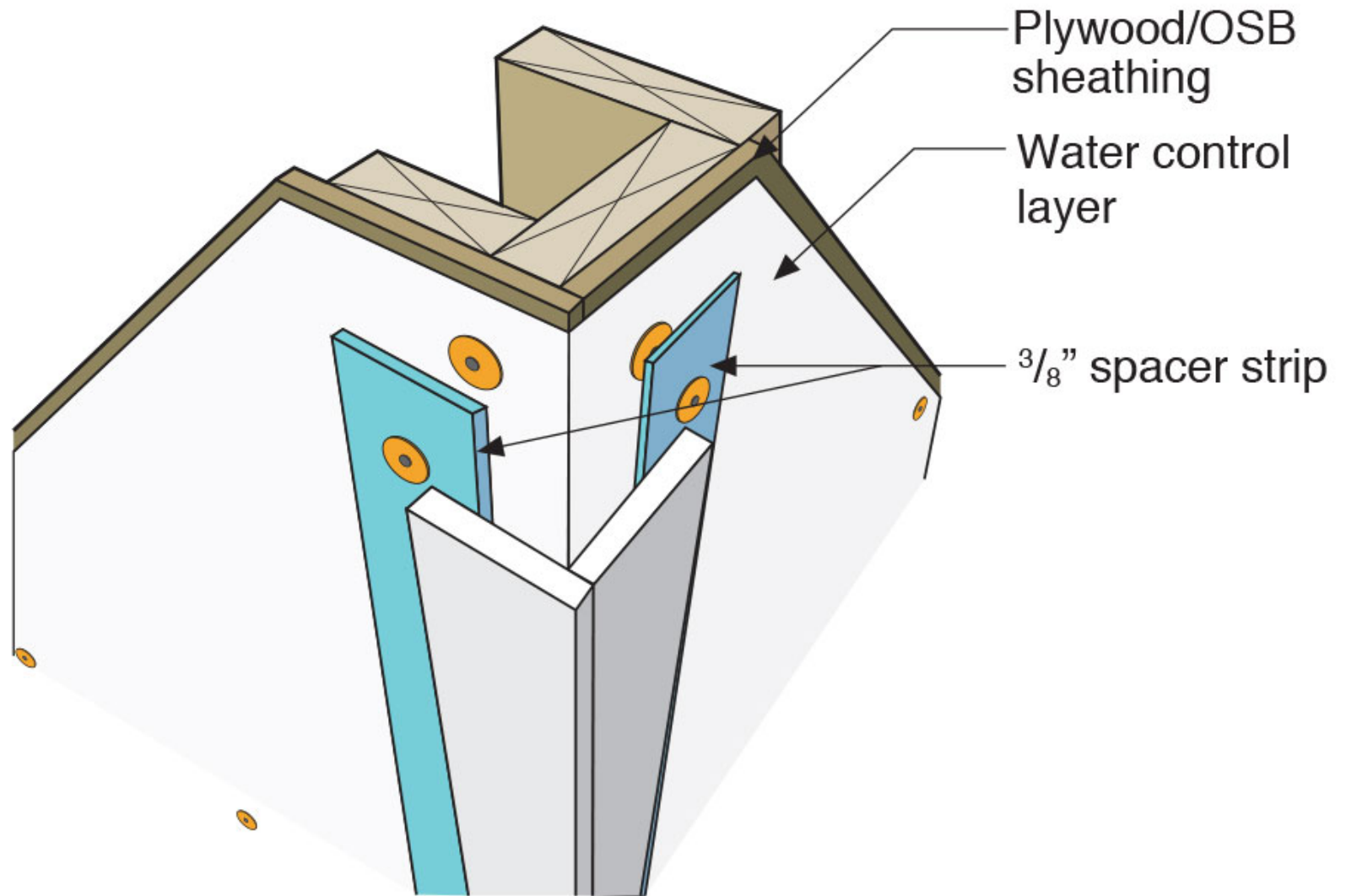


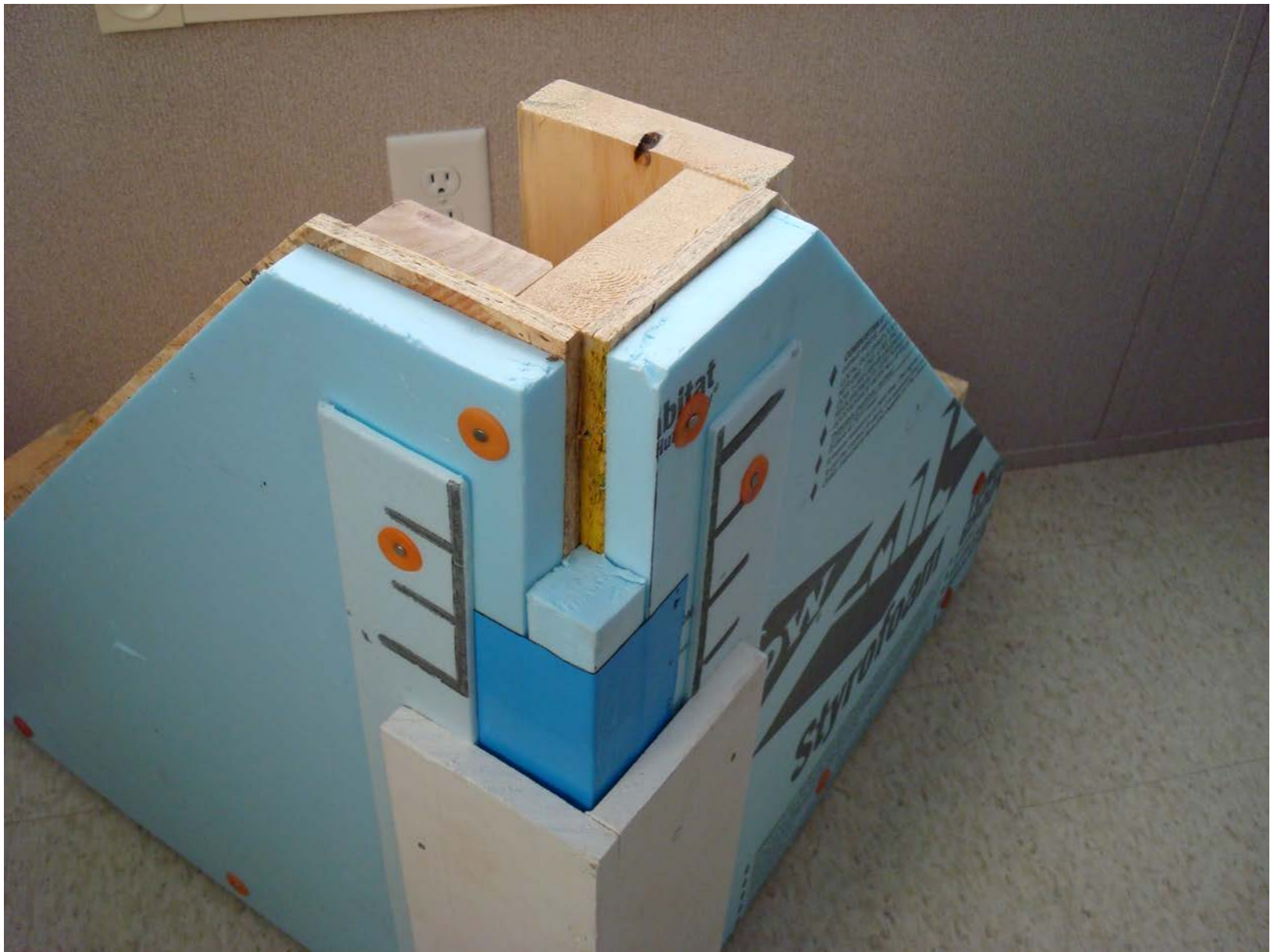
Siding Laps









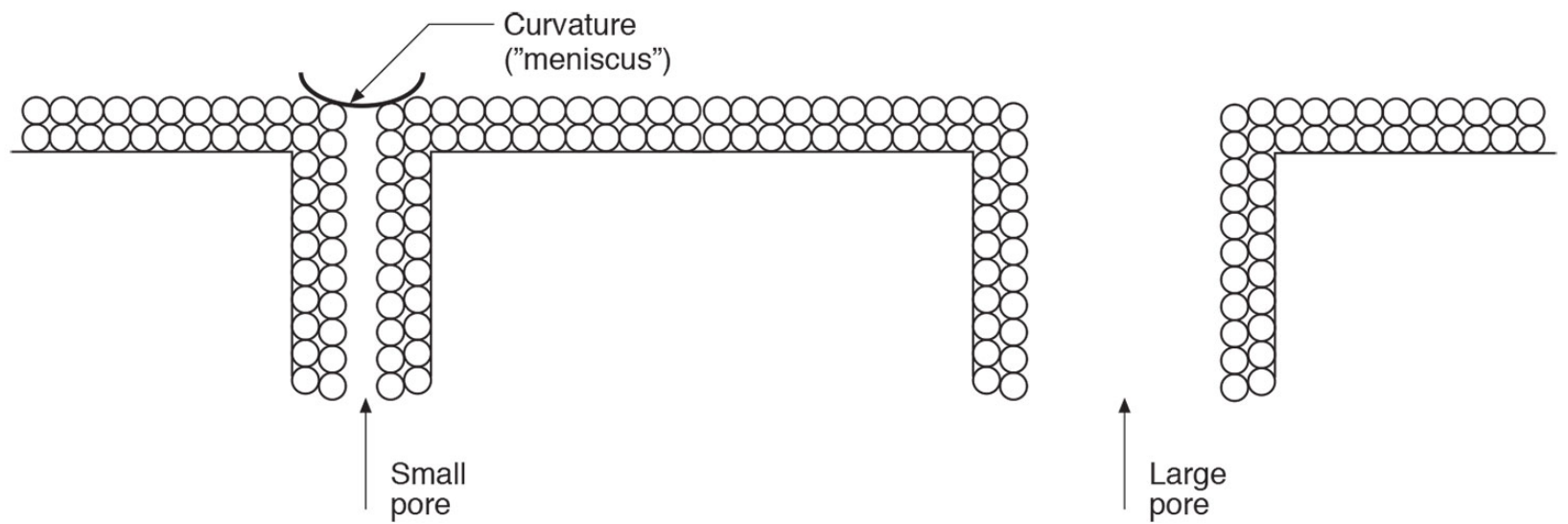




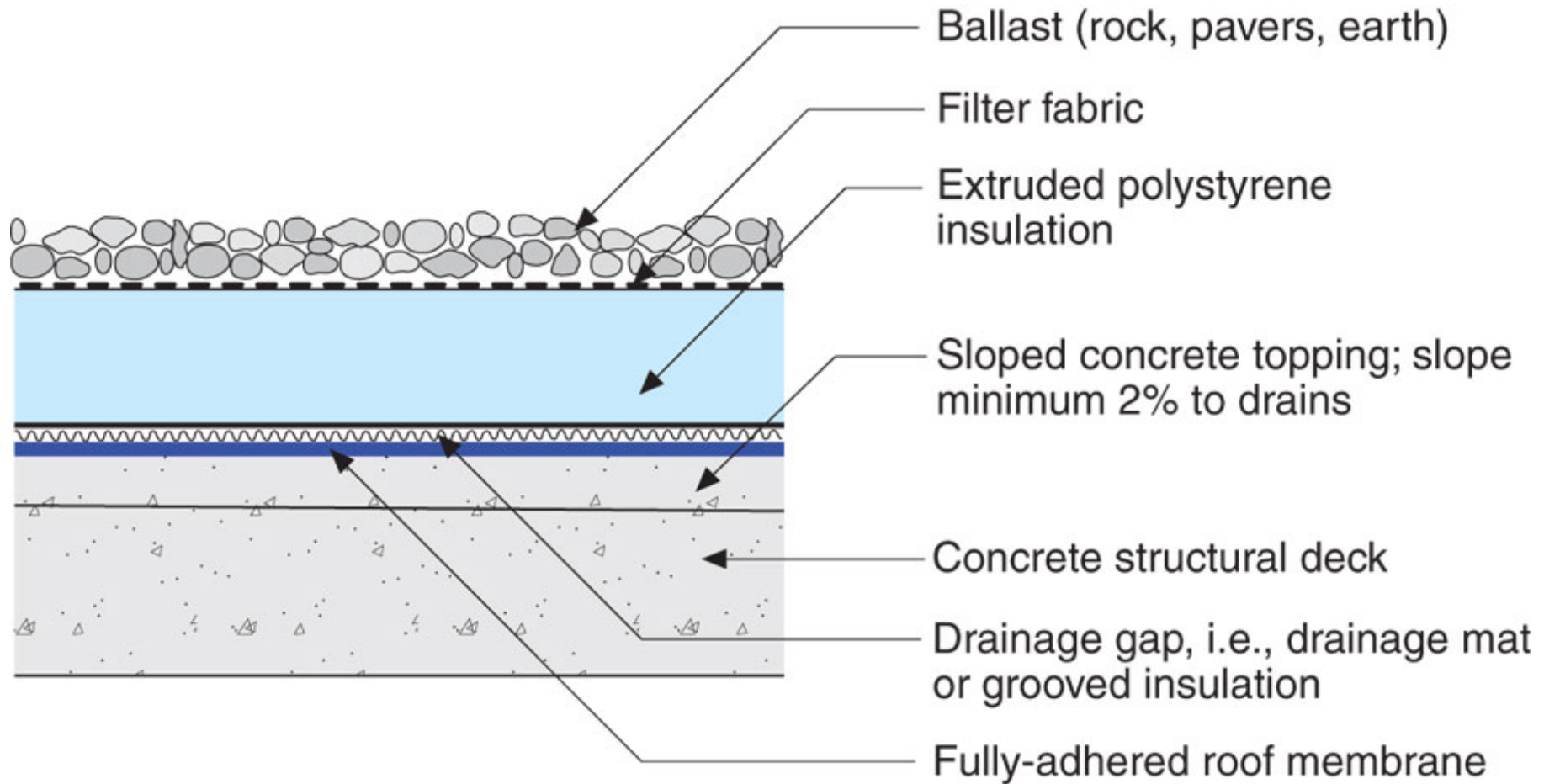


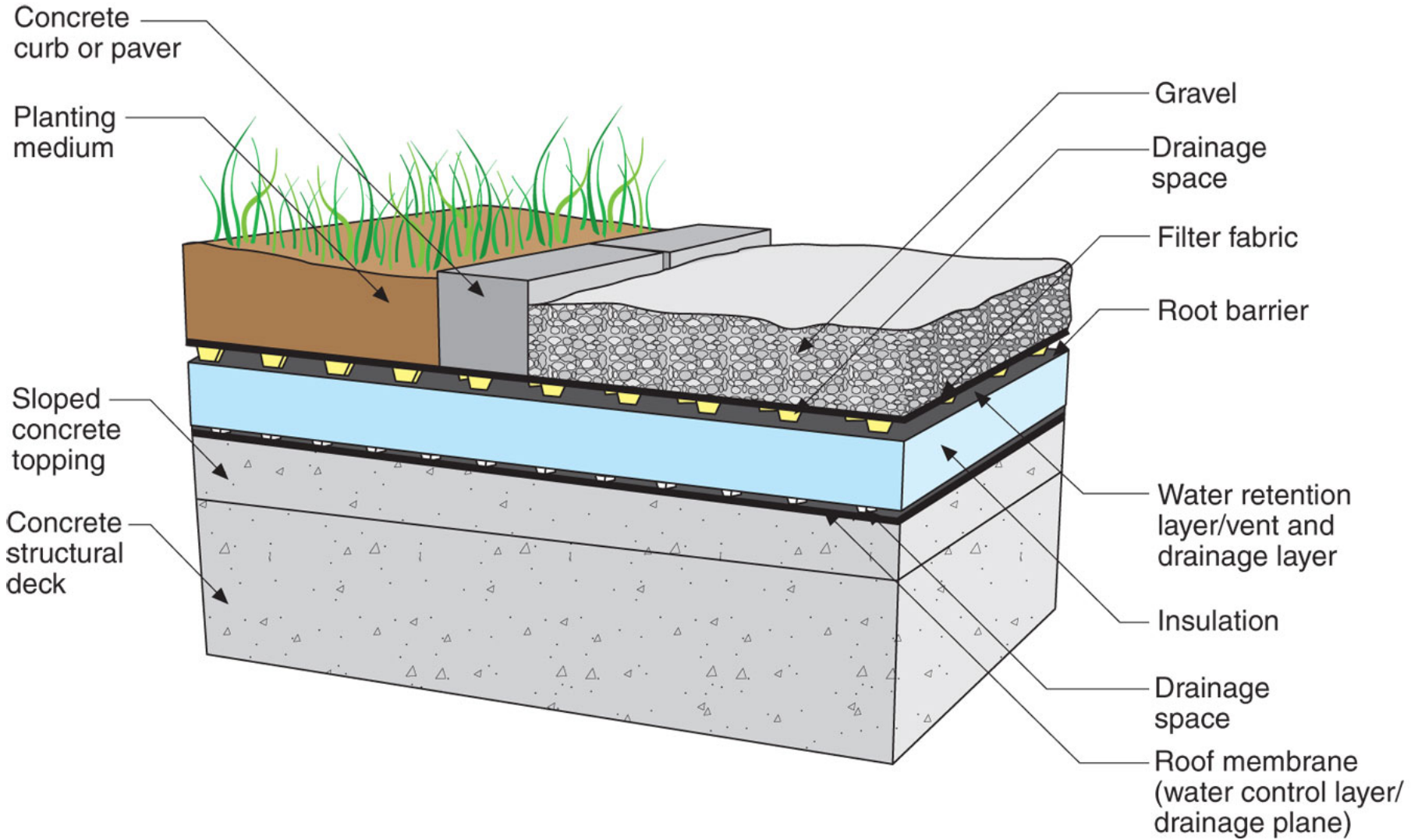


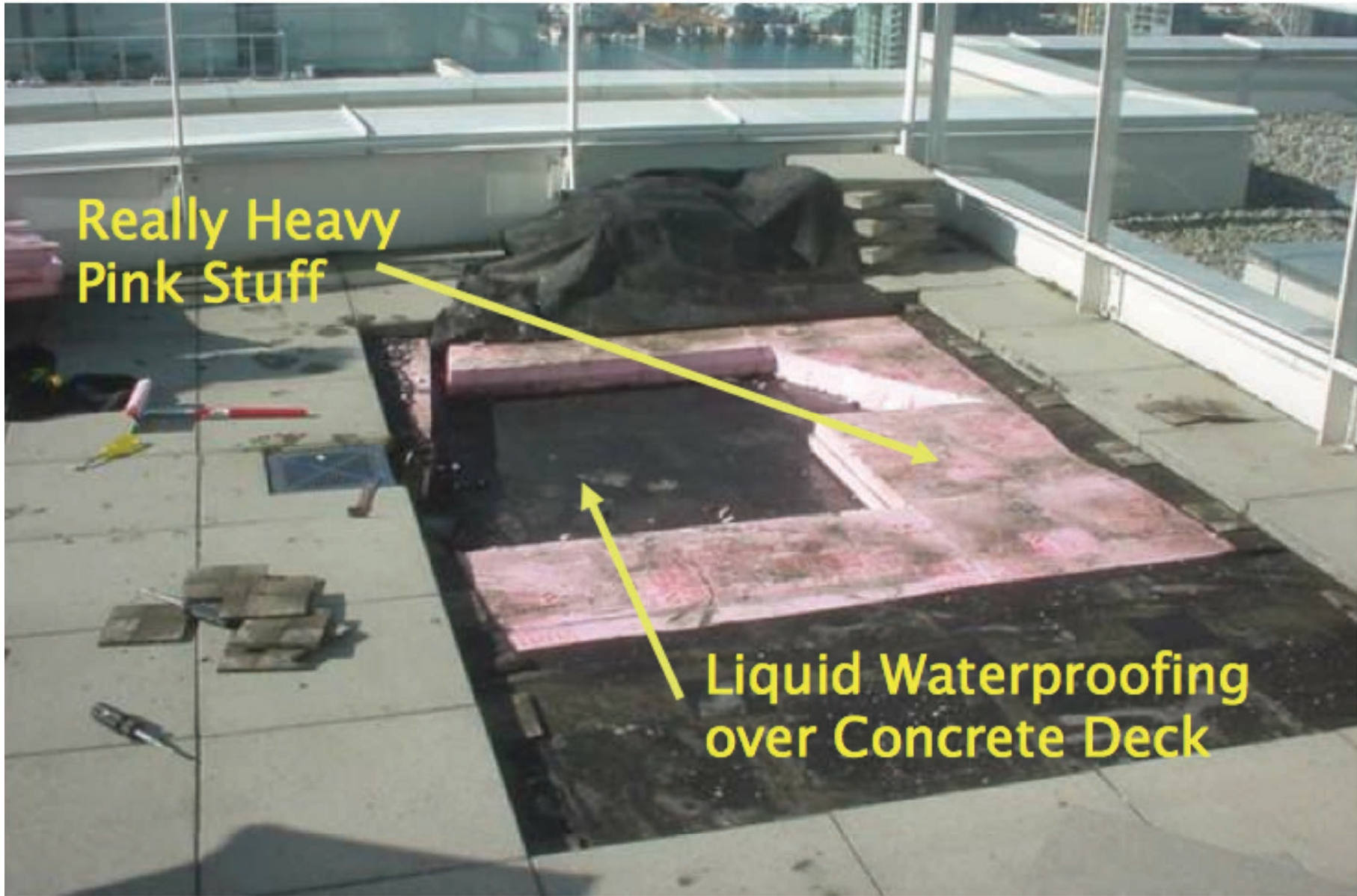












Really Heavy
Pink Stuff

Liquid Waterproofing
over Concrete Deck

Vapor

Diffusion

Convective Flow

Vapor Concentration

Air Pressure

Adsorbate

Surface Diffusion

Concentration

Liquid

Capillary Flow

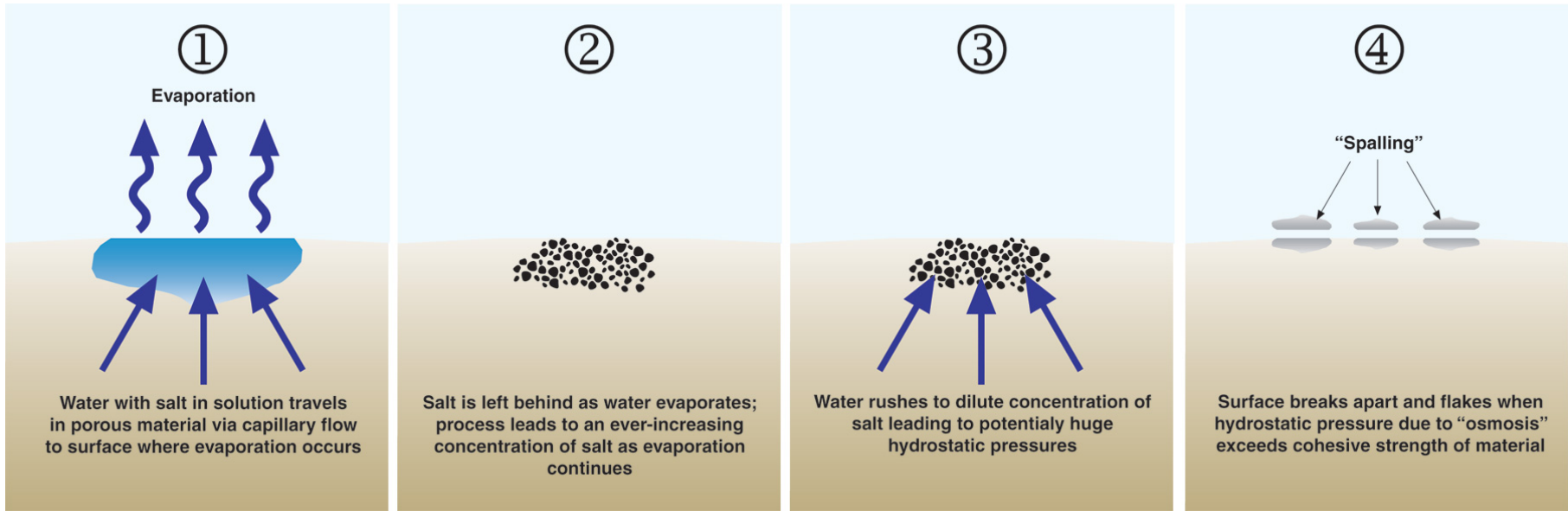
Osmosis

Suction Pressure

Solute Concentration

Capillarity + Salt = Osmosis

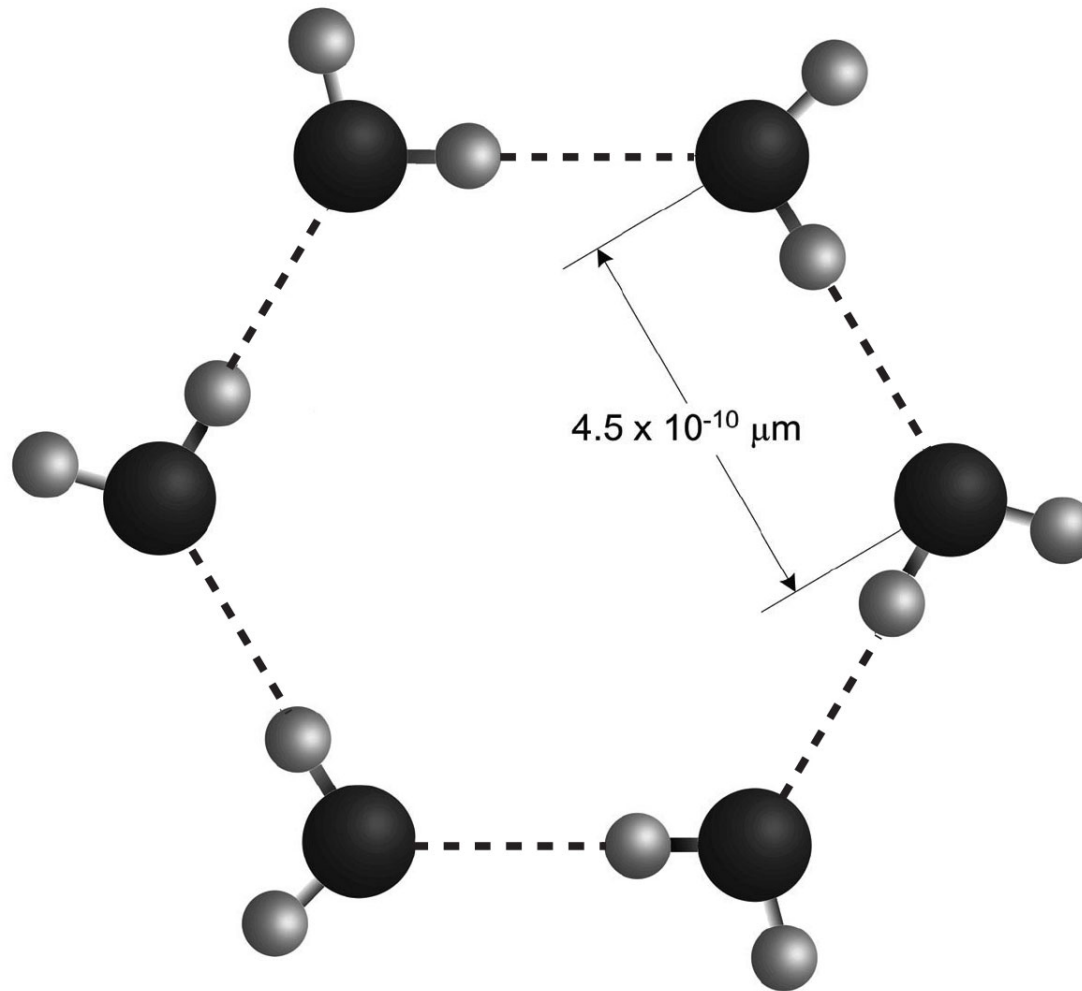
- Mineral salts carried in solution by capillary water
- When water evaporates from a surface the salts left behind form crystals in process called efflorescence
- When water evaporated beneath a surface the salts crystallize within the pore structure of the material in called sub-efflorescence
- The salt crystallization causes expansive forces that can exceed the cohesive strength of the material leading to spalling



Pressures

- Diffusion Vapor Pressure 3 to 5 psi
- Capillary Pressure 300 to 500 psi
- Osmosis Pressure 3,000 to 5,000 psi



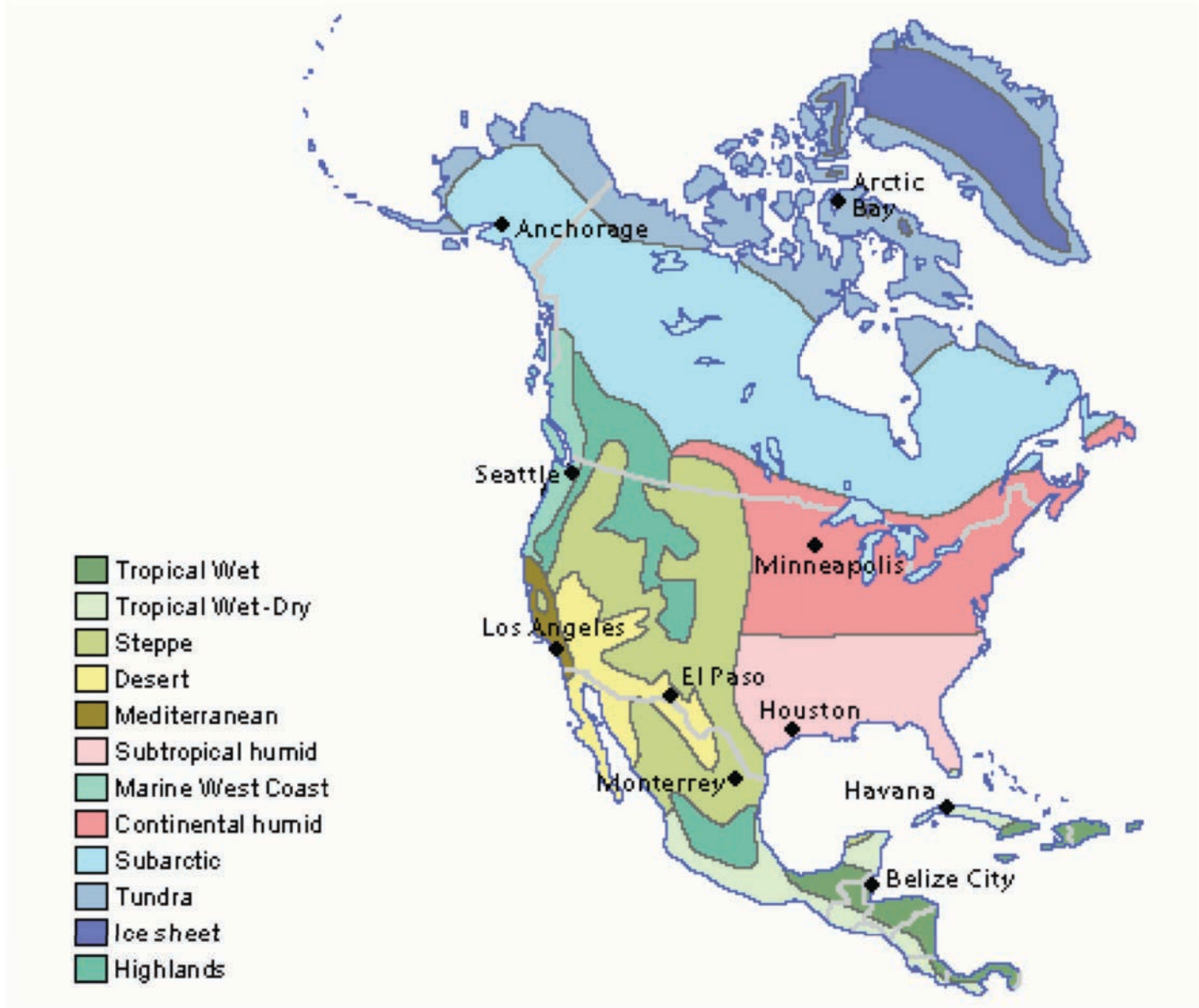


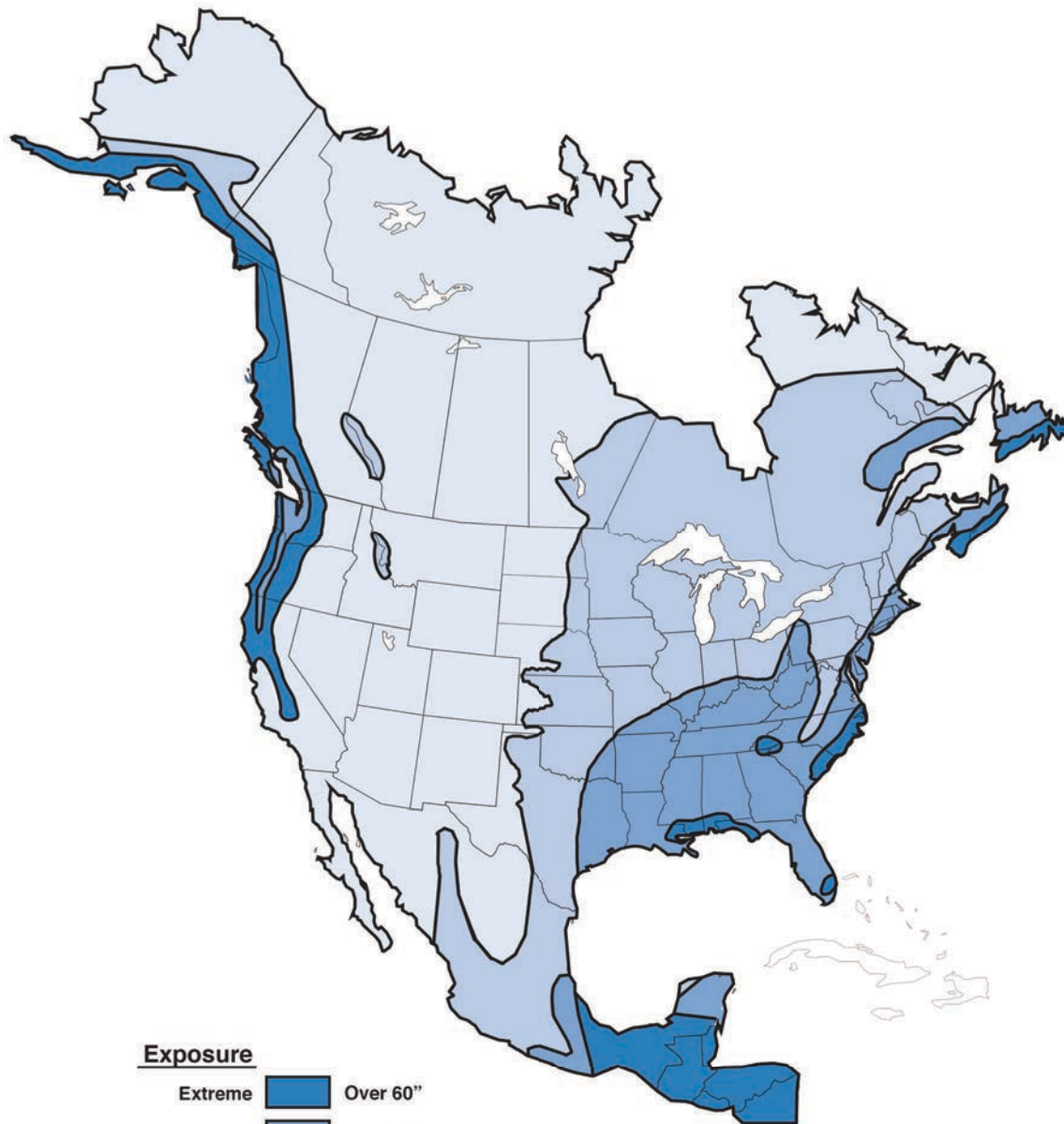


Freeze-Thaw Damage





Freeze-Thaw Damage
Freezing Temperatures
Water
Susceptible Brick

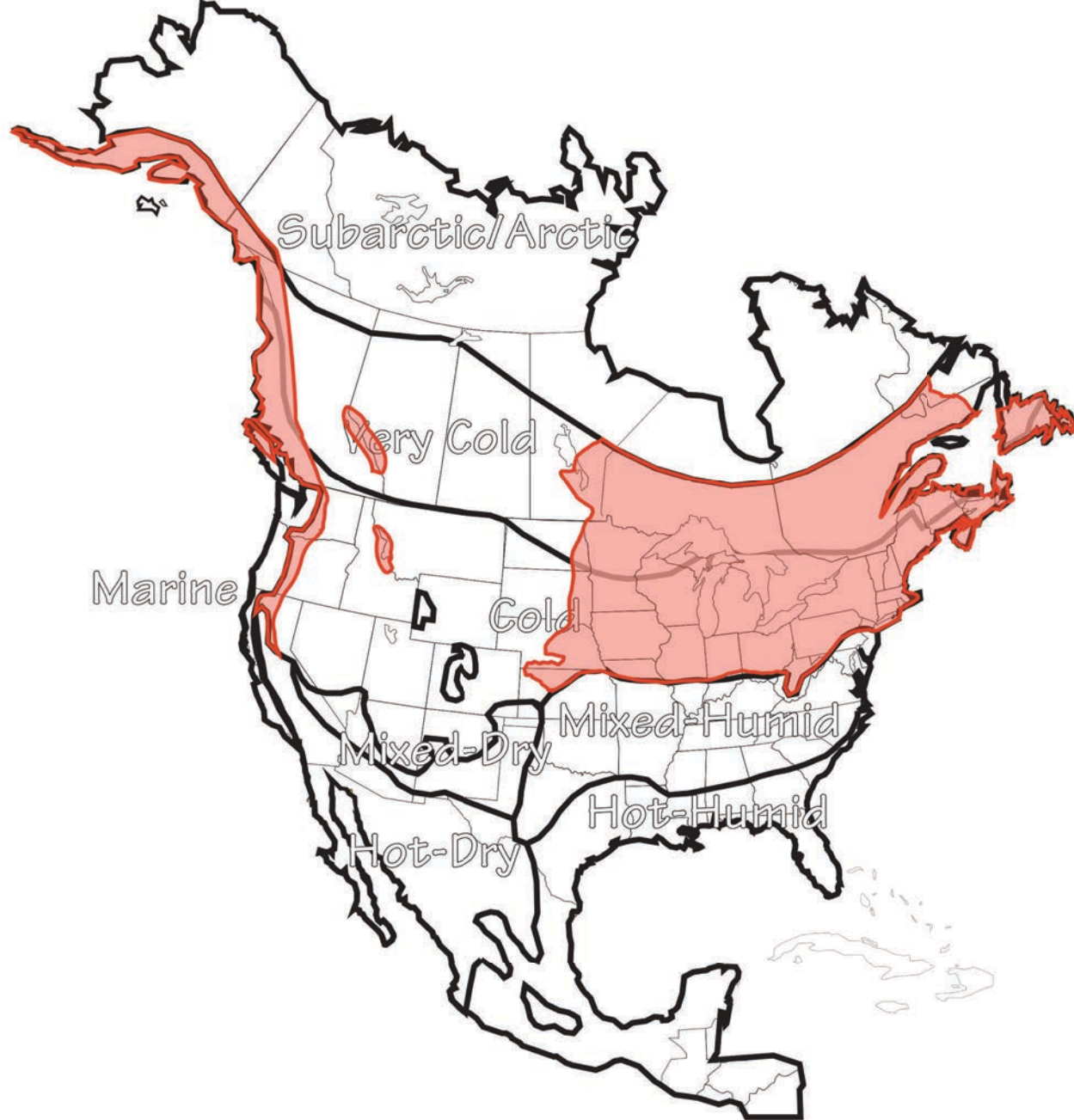




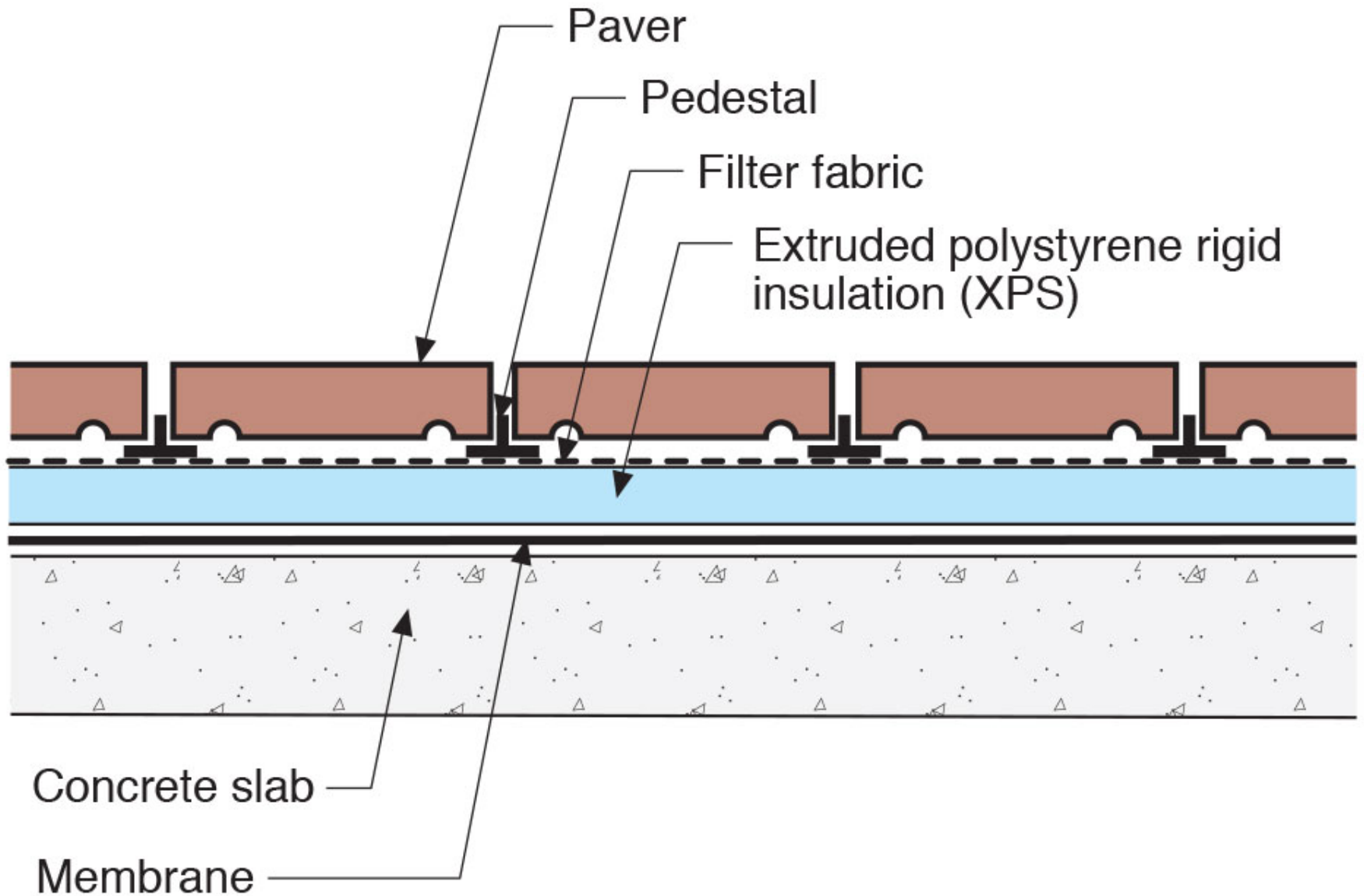


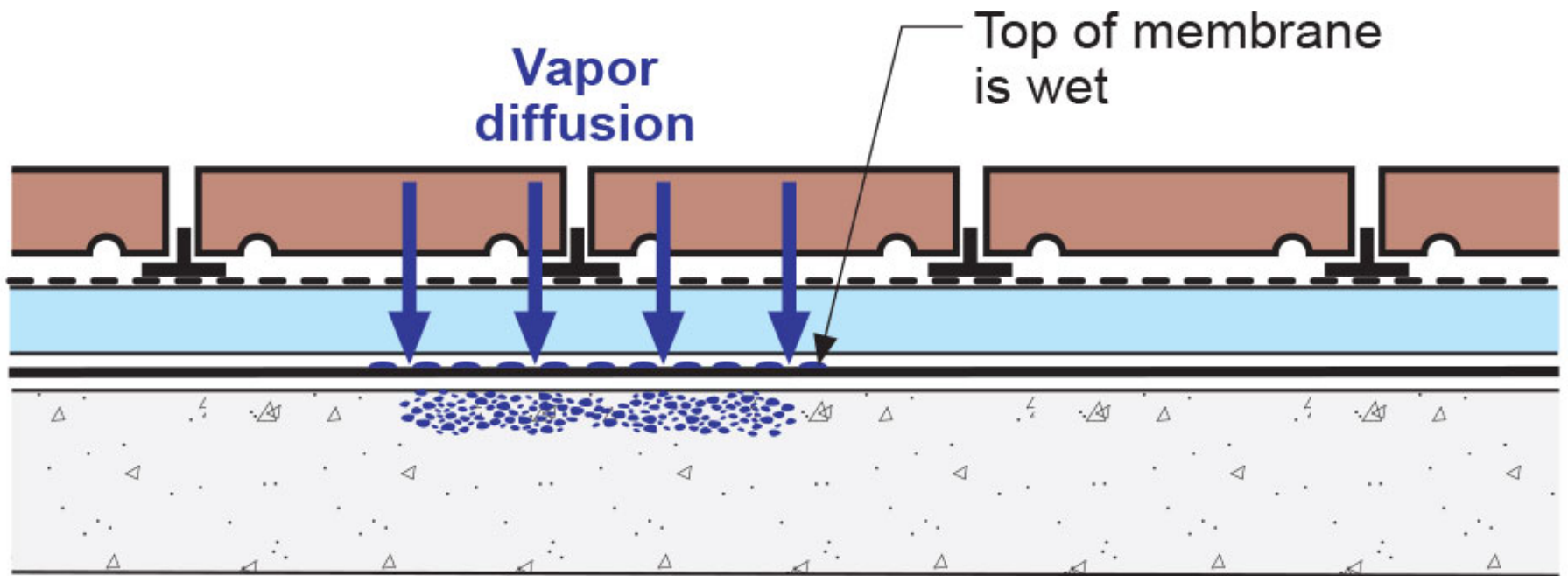
Exposure

Extreme		Over 60"
High		40" - 60"
Moderate		20" - 40"
Low		Under 20"

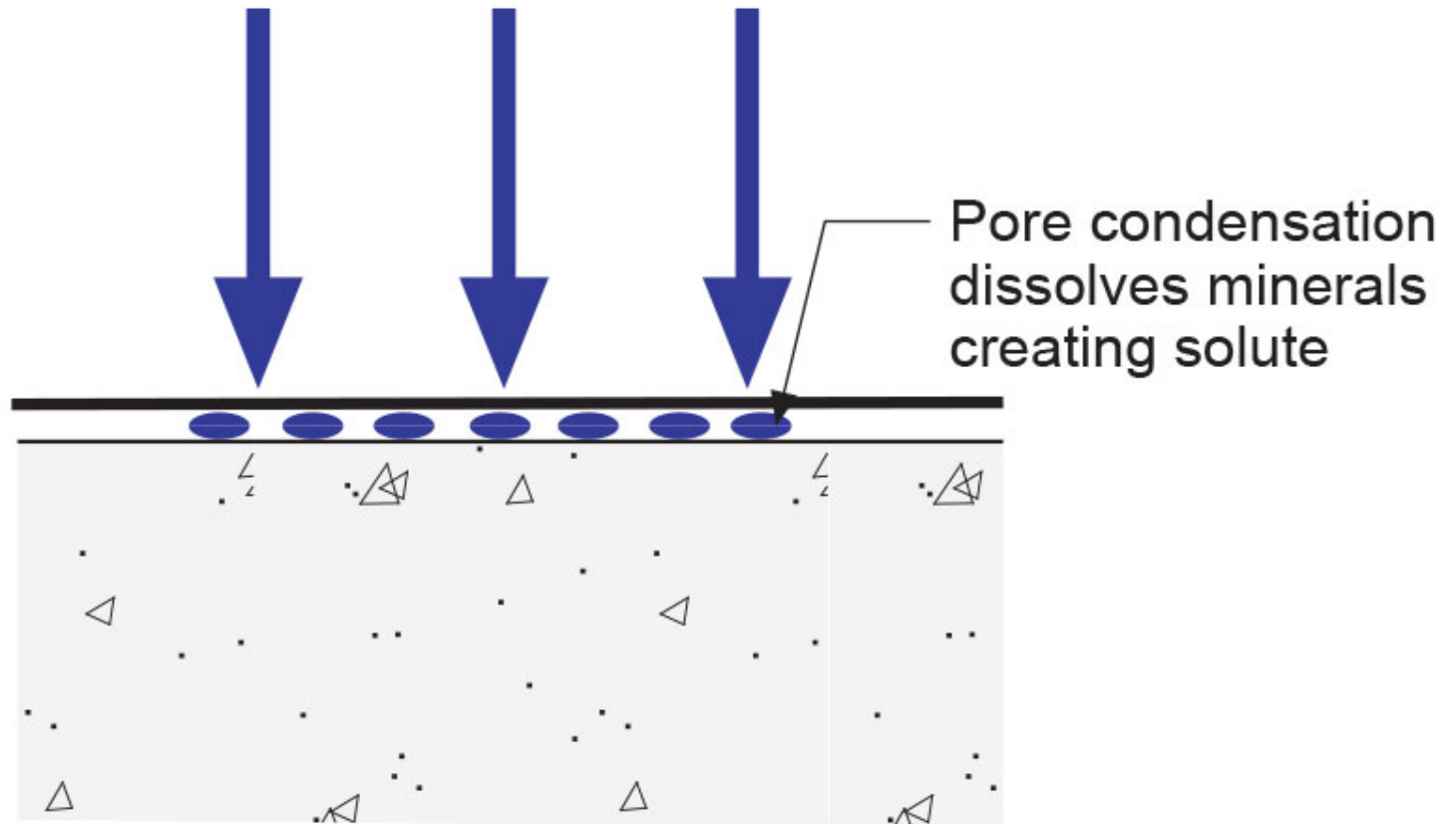


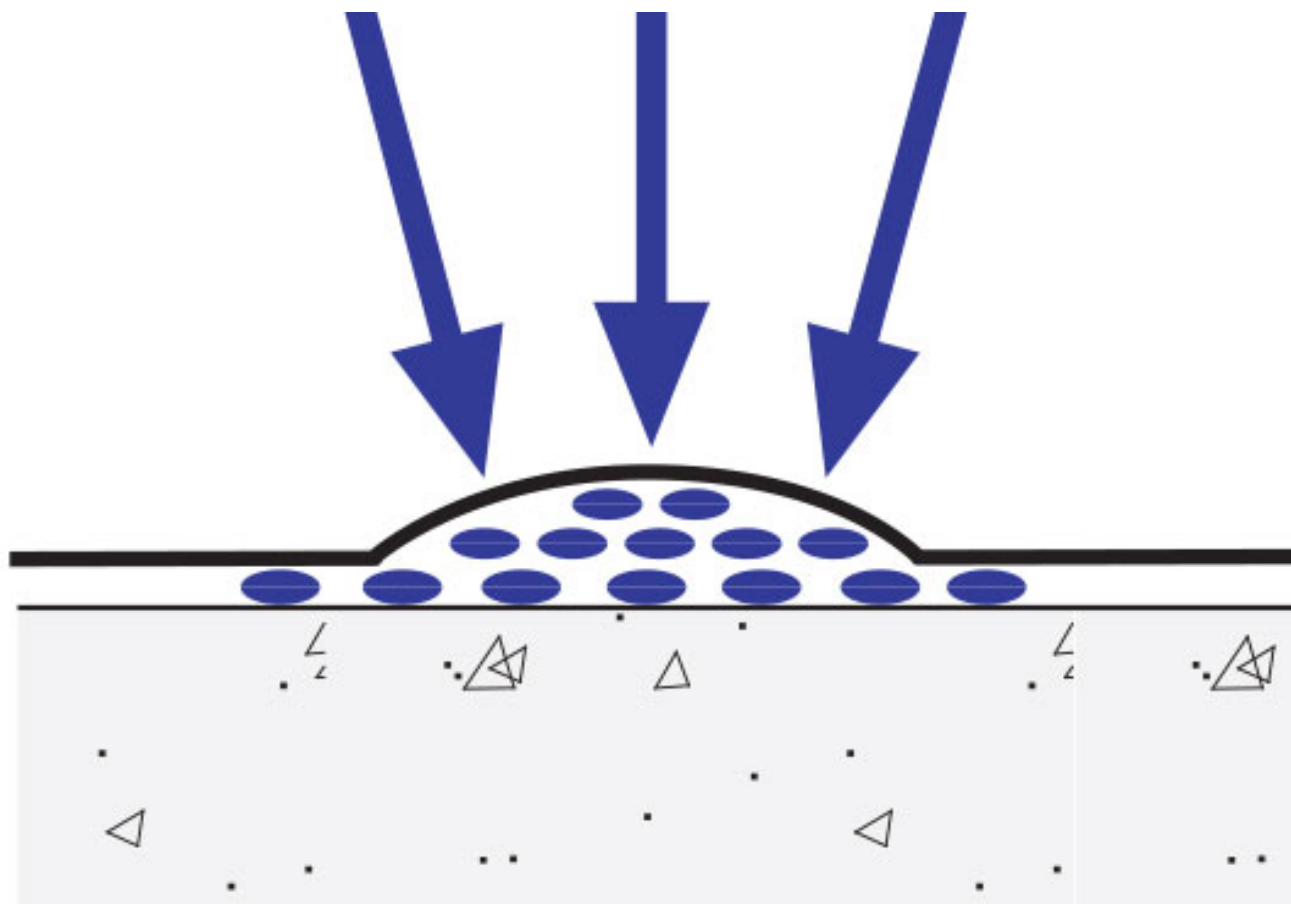
More Osmosis





Vapor diffusion

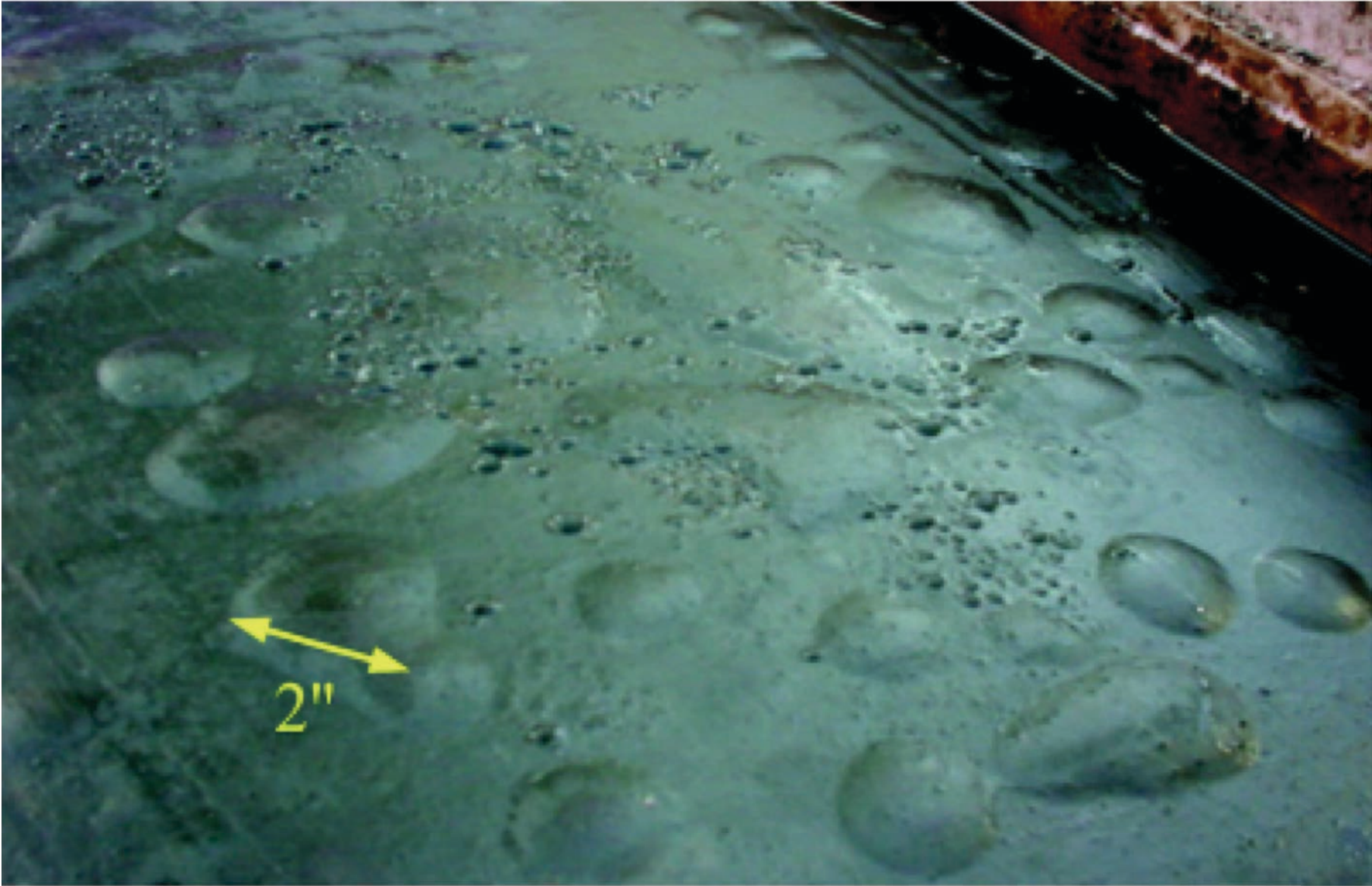






Paver Water Beds!

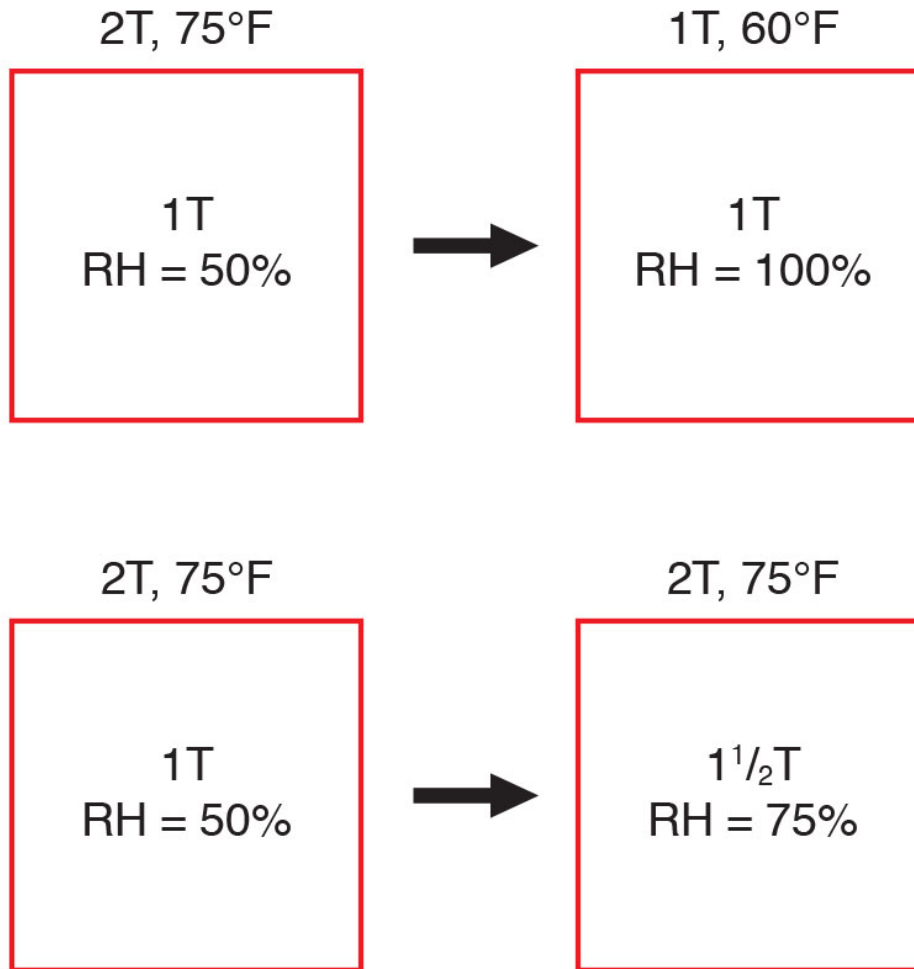


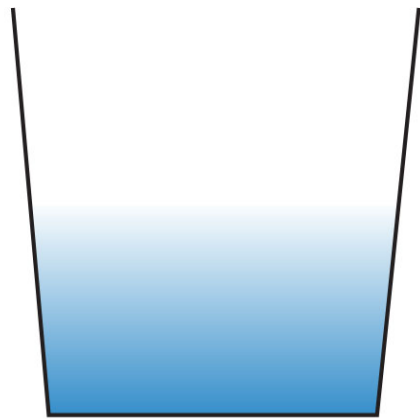




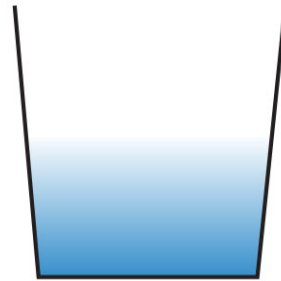
Need To Do An Aside...Necessary For A Segway

Relative Humidity Vapor Pressure

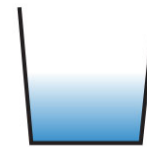




90°F
50% RH



75°F
50% RH



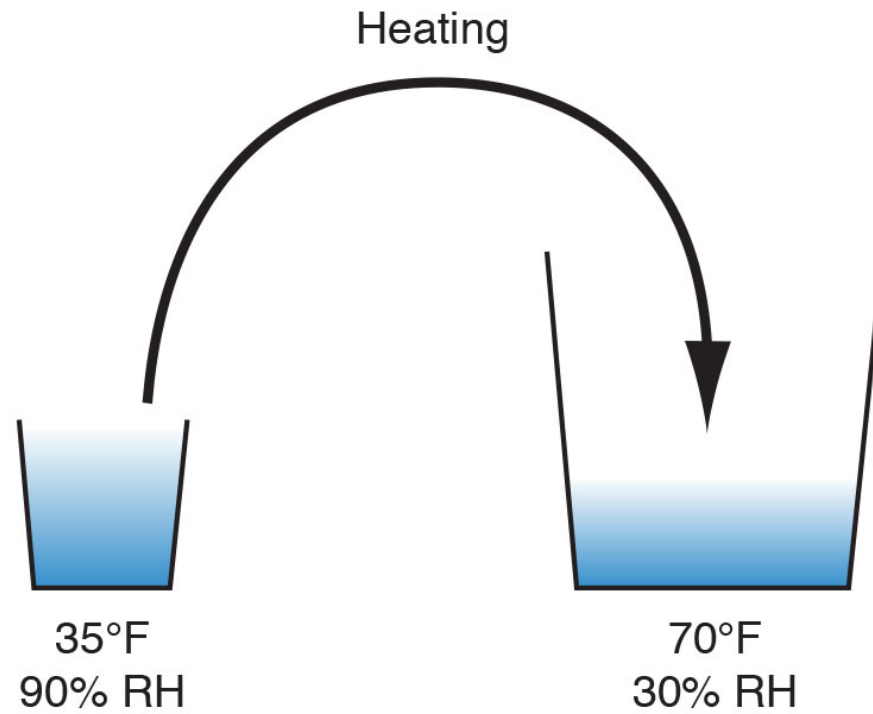
60°F
50% RH

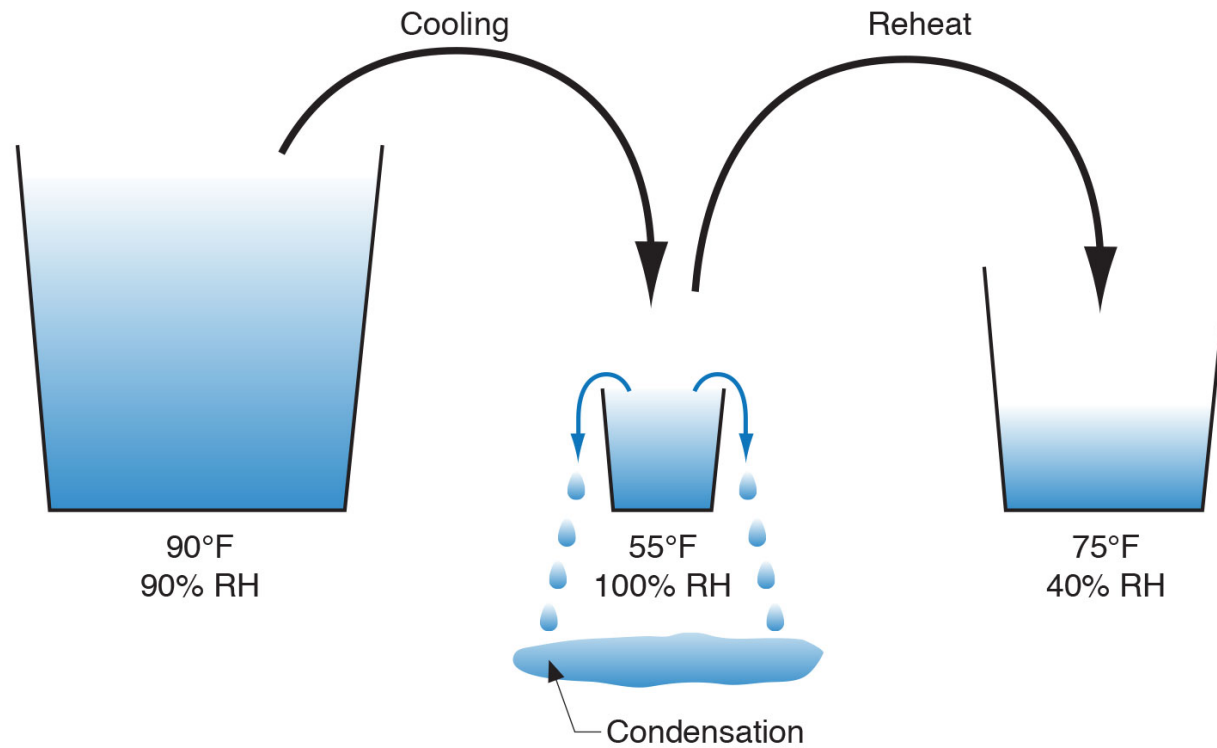


45°F
50% RH

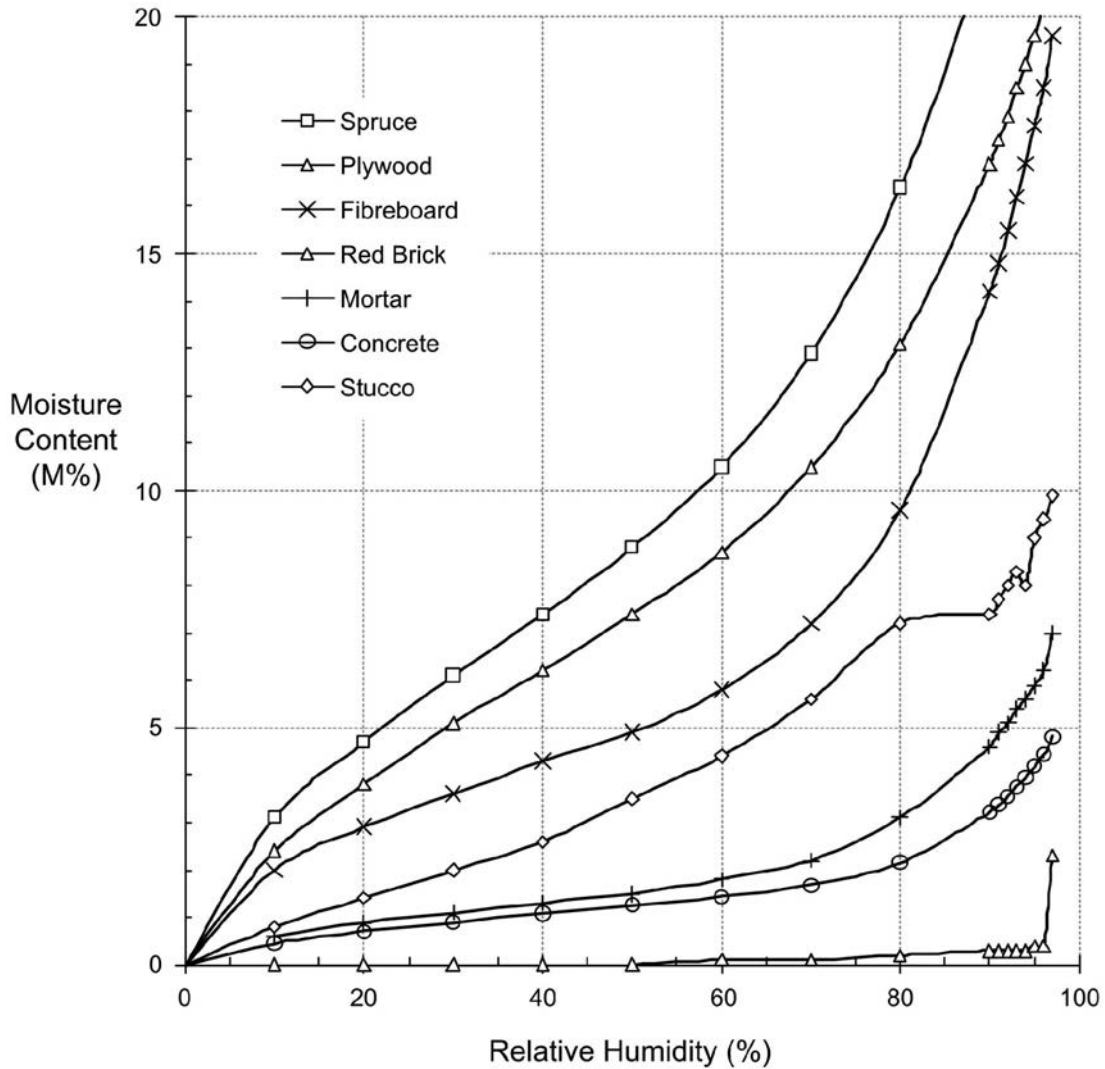


30°F
50% RH





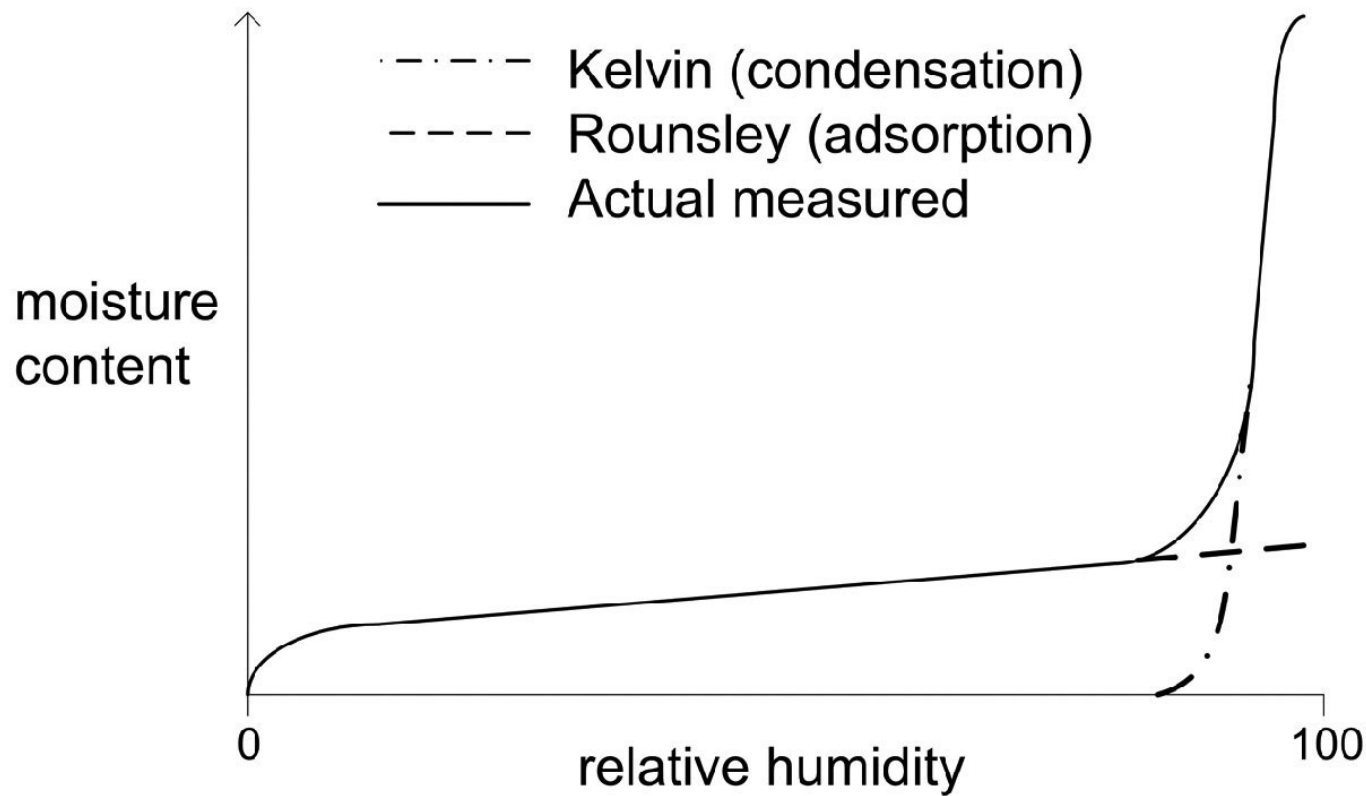
Sorption



Sorption isotherm for several building materials [Kumaran 2002]
 From Straube & Burnett, 2005

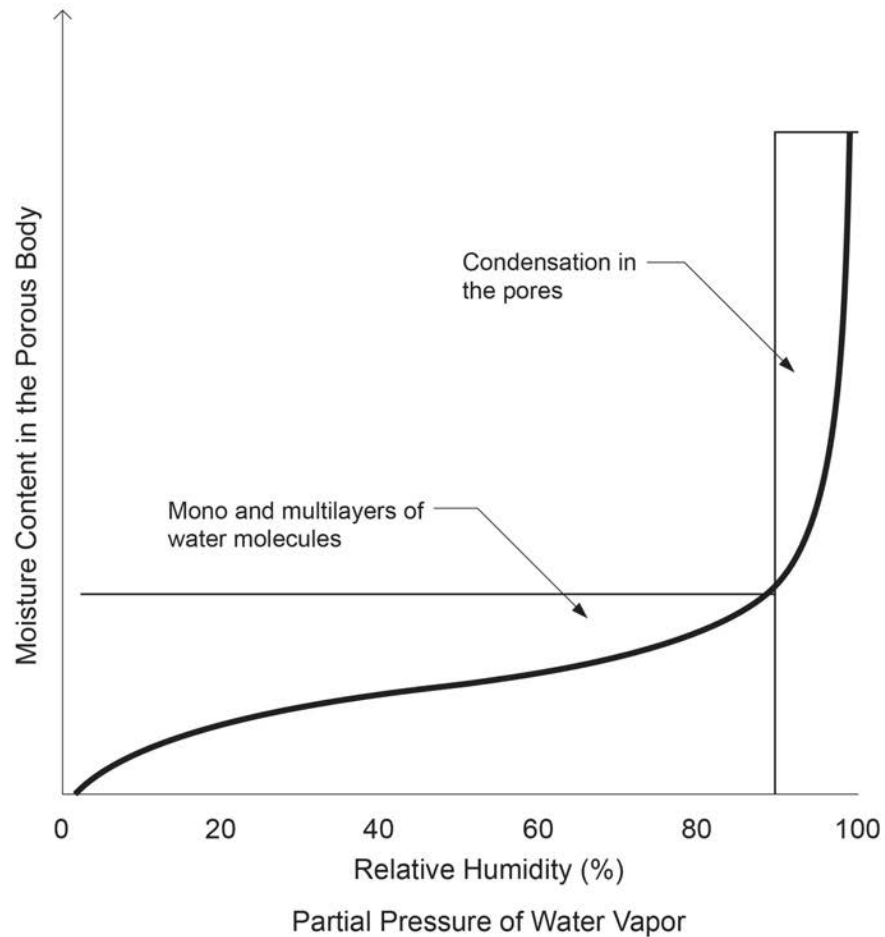
BET Theory

BET Theory
Stephen Brunauer
Paul Emmett
Edward Teller



**Typical predicted sorption isotherm according to Kelvin equation
and modified BET theory**

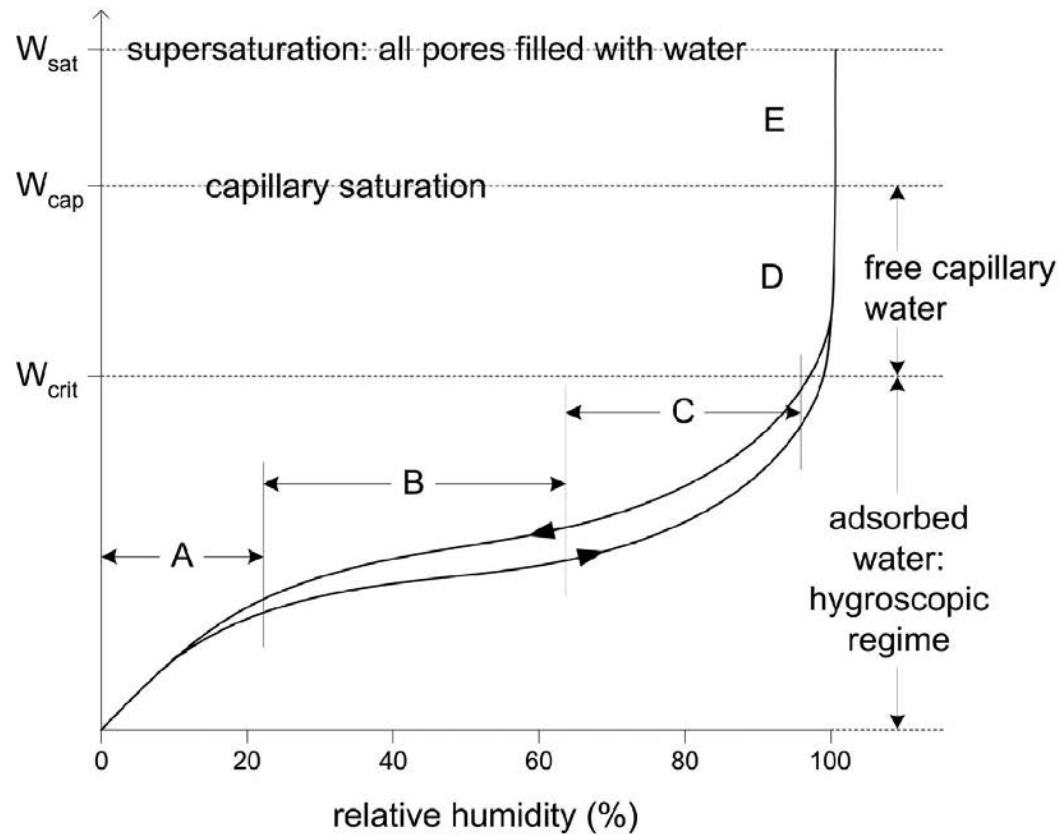
From Straube & Burnett, 2005



Change in the storage of moisture in a porous building material as the partial pressure of water vapor in the ambient air increases from zero to full saturation value at a given temperature.

Sorption Curve

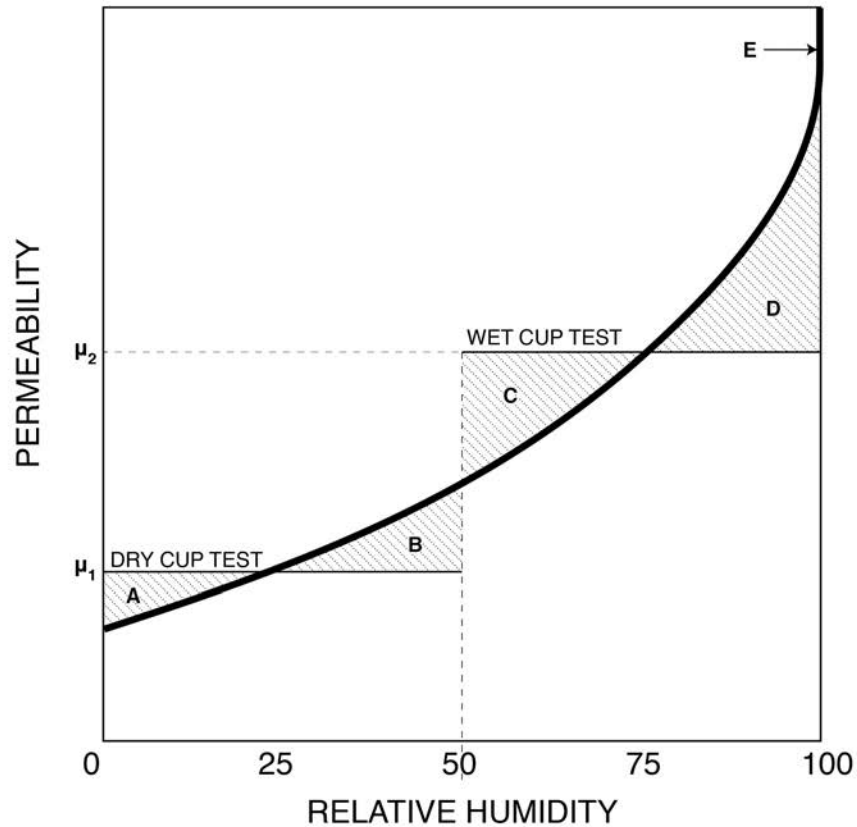
From M.K. Kumaran, ASTM MNL 18-2nd Edition,
Moisture Control in Buildings, 2009



- A: Single-layer of adsorbed molecules
- B: Multiple layers of adsorbed molecules
- C: Interconnected layers (internal capillary condensation)
- D: Free water in Pores, capillary suction
- E: Supersaturated Regime

Regimes of moisture storage in a hygroscopic porous material

From Straube & Burnett, 2005

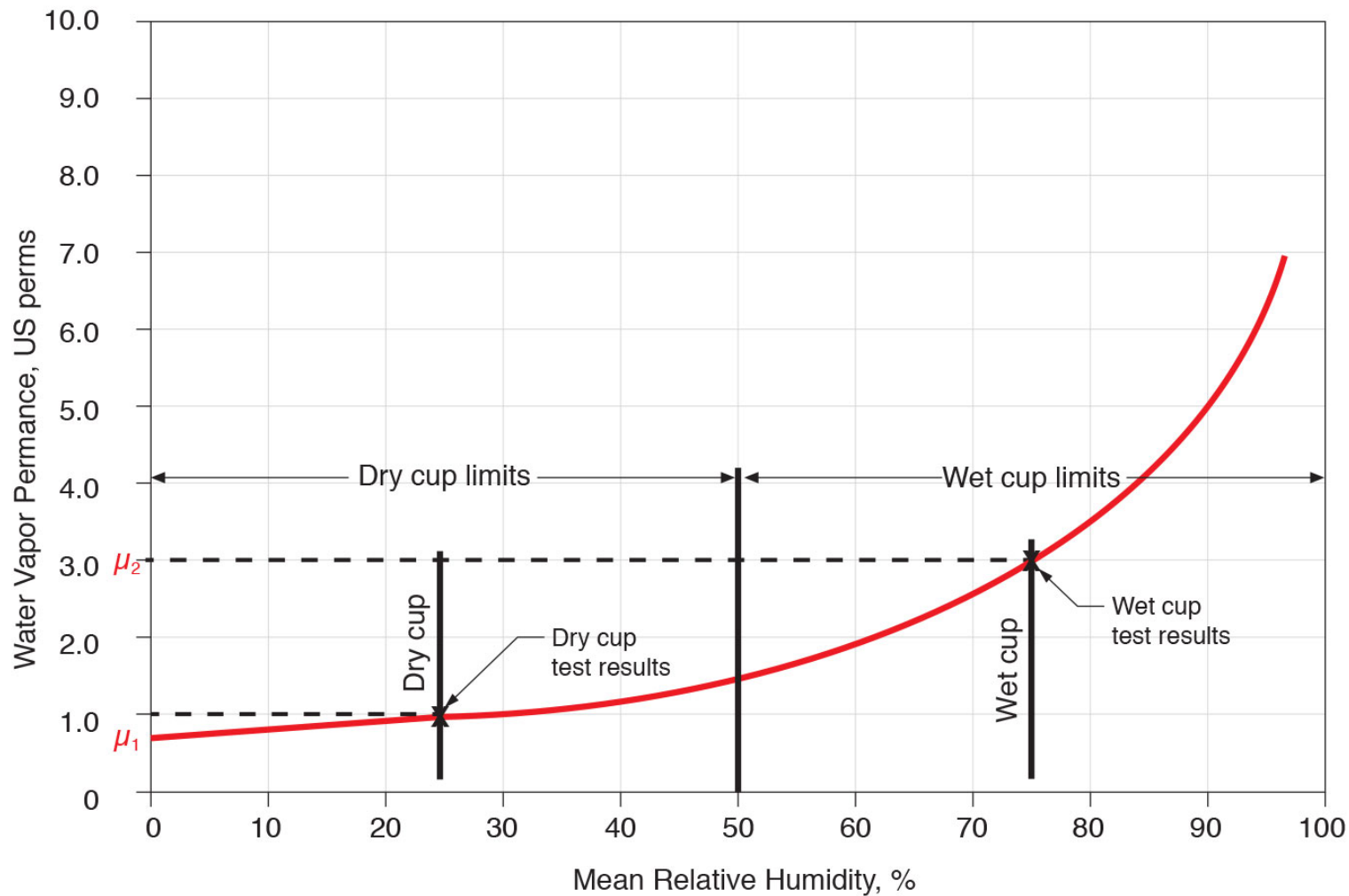


- A - Single-layer of absorbed molecules
- B - Multiple layers of absorbed molecules
- C - Interconnected layers (internal capillary condensation)
- D - Free water in pores, capillary suction
- E - Supersaturated regime

Relationship between Dry Cup and Wet Cup
Adapted from Joy & Wilson, 1963



Water Vapor Permeance vs. Relative Humidity



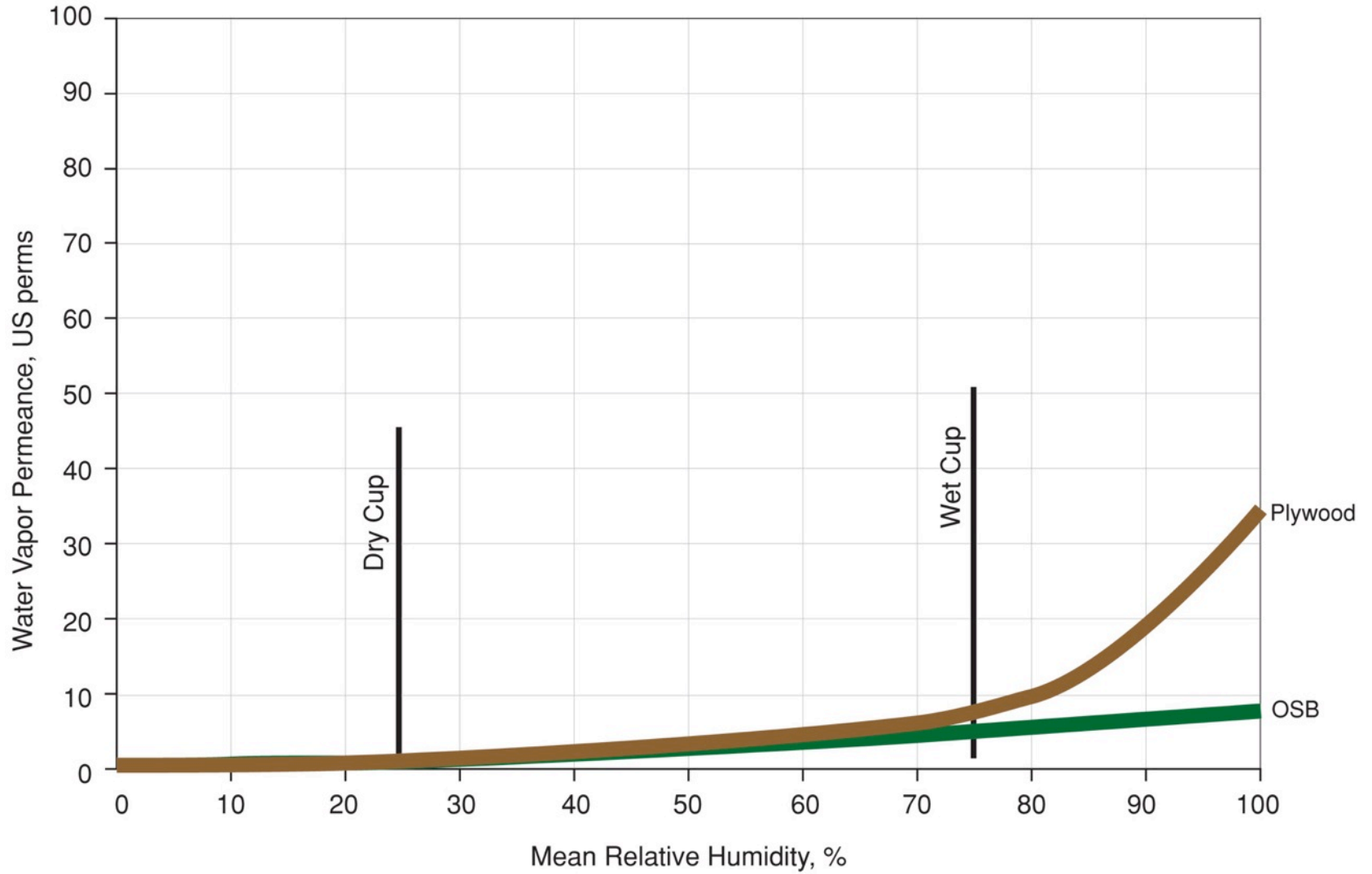
μ_1 = Dry cup permeance

μ_2 = Wet cup permeance

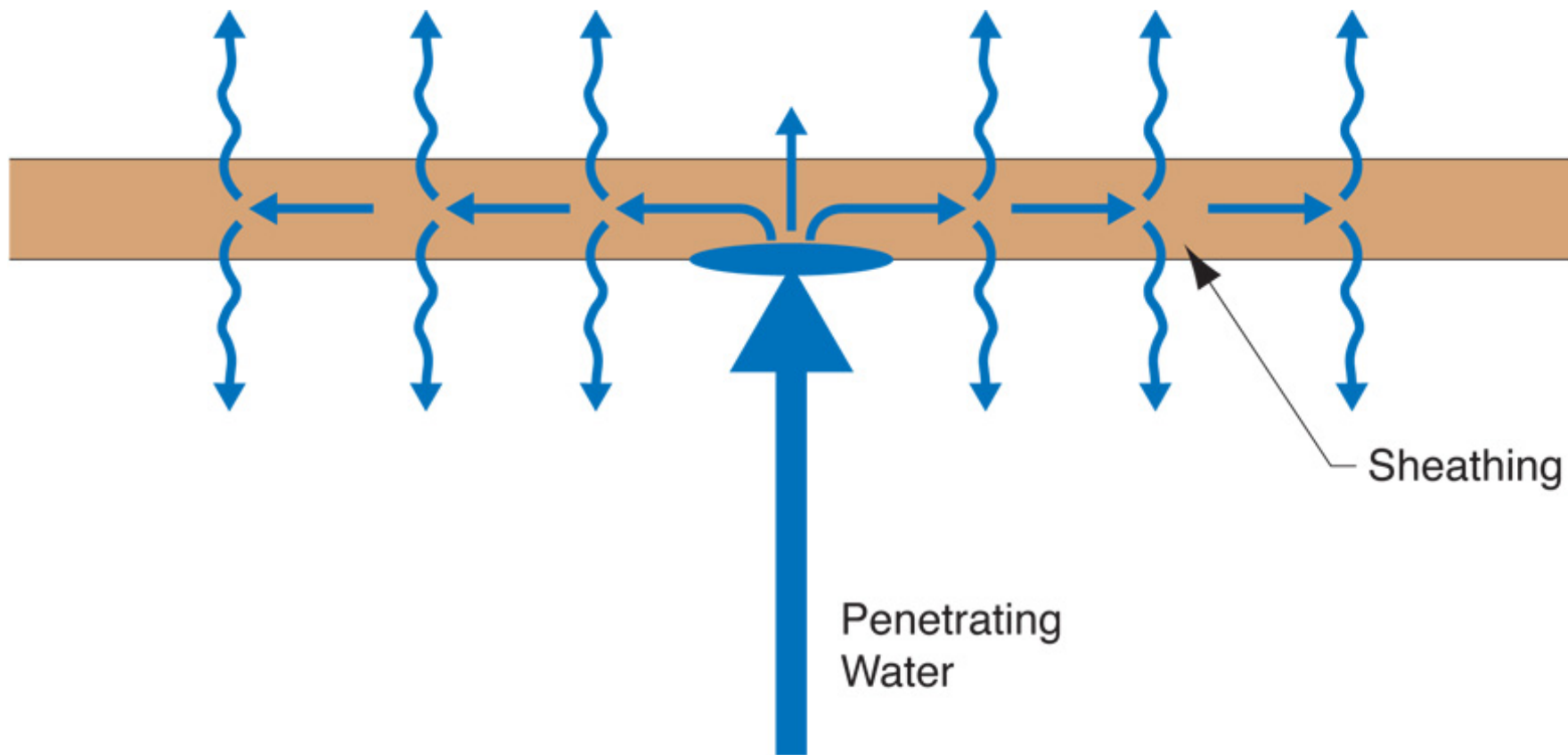


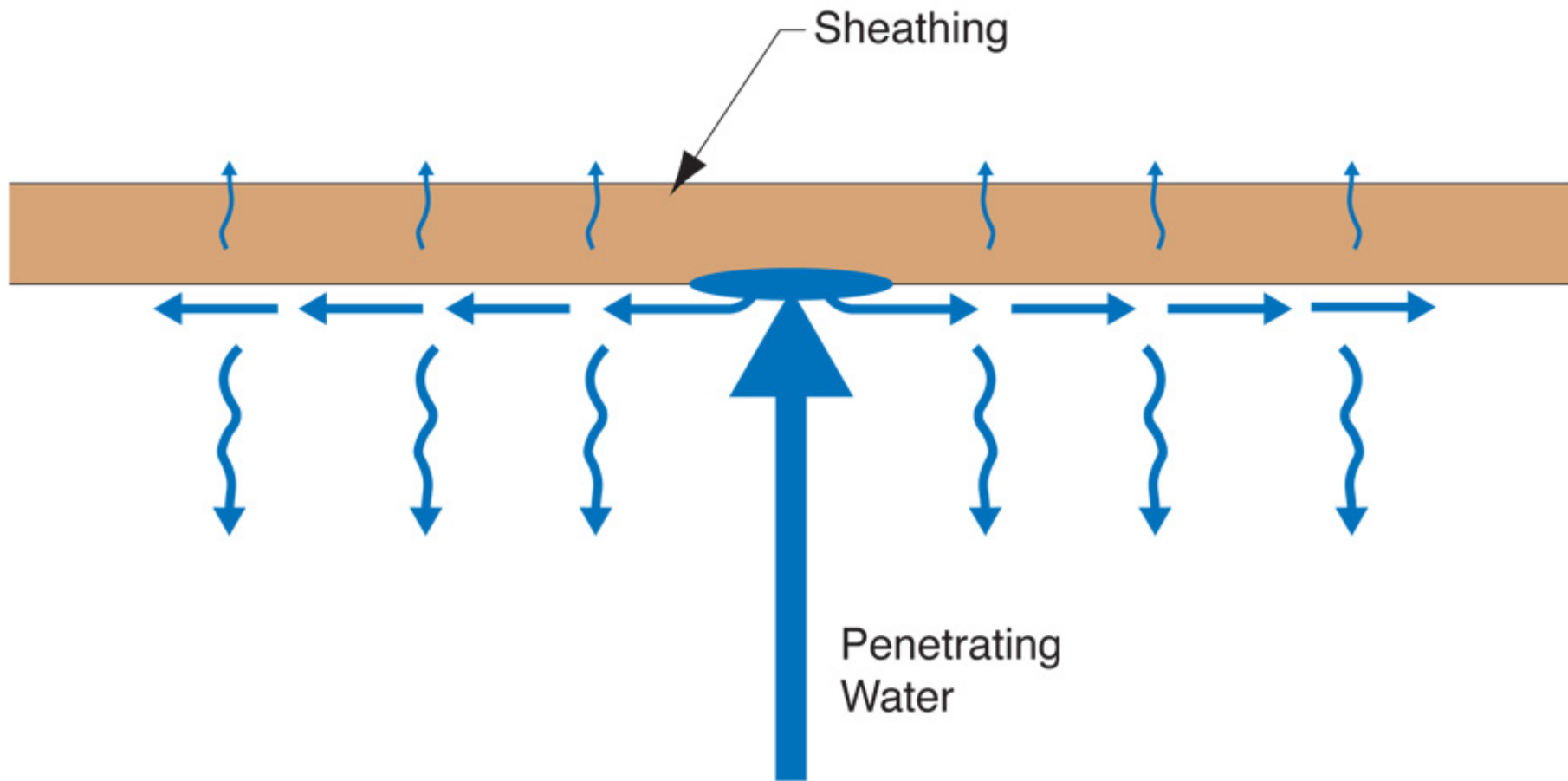


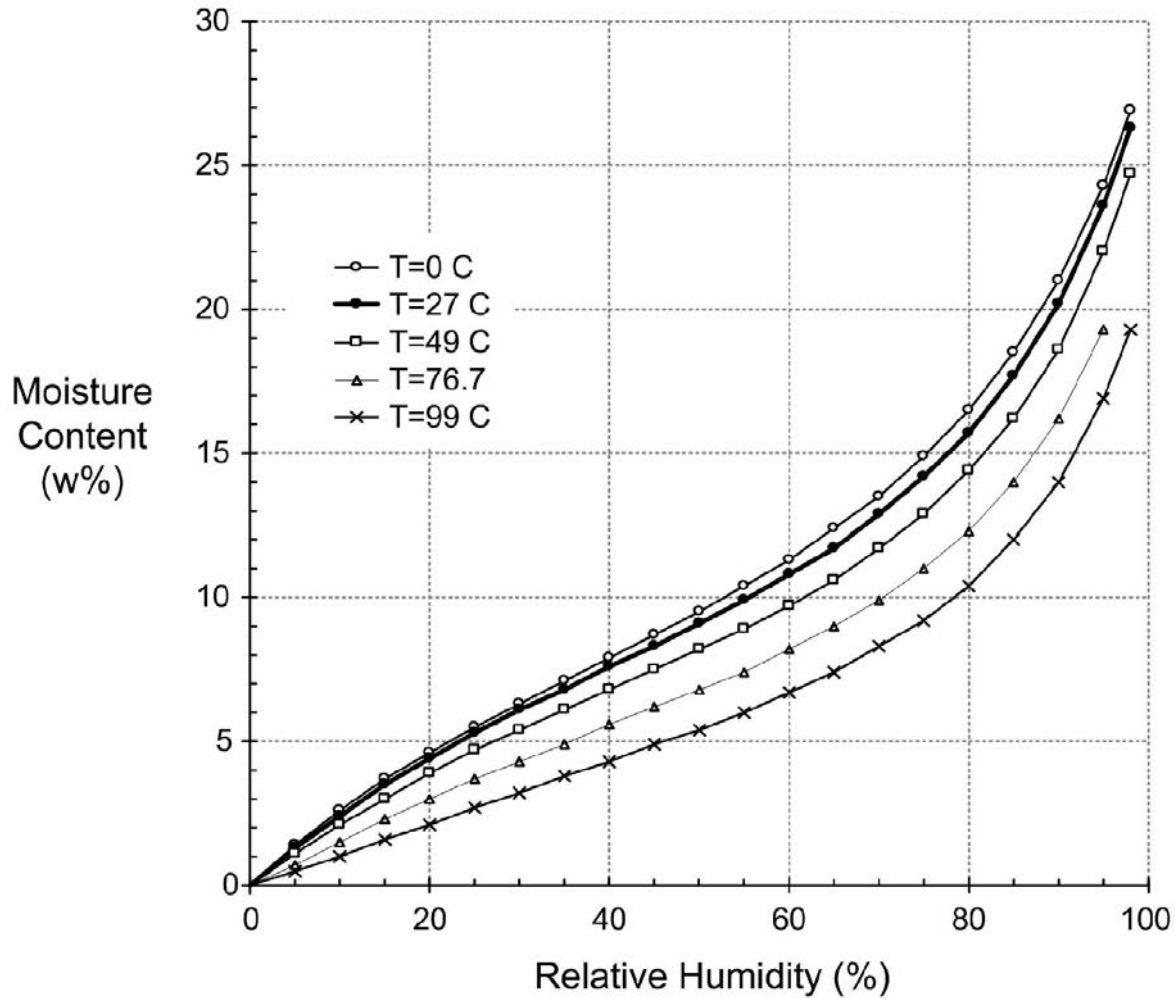
Water Vapor Permeance of Sheathing Materials





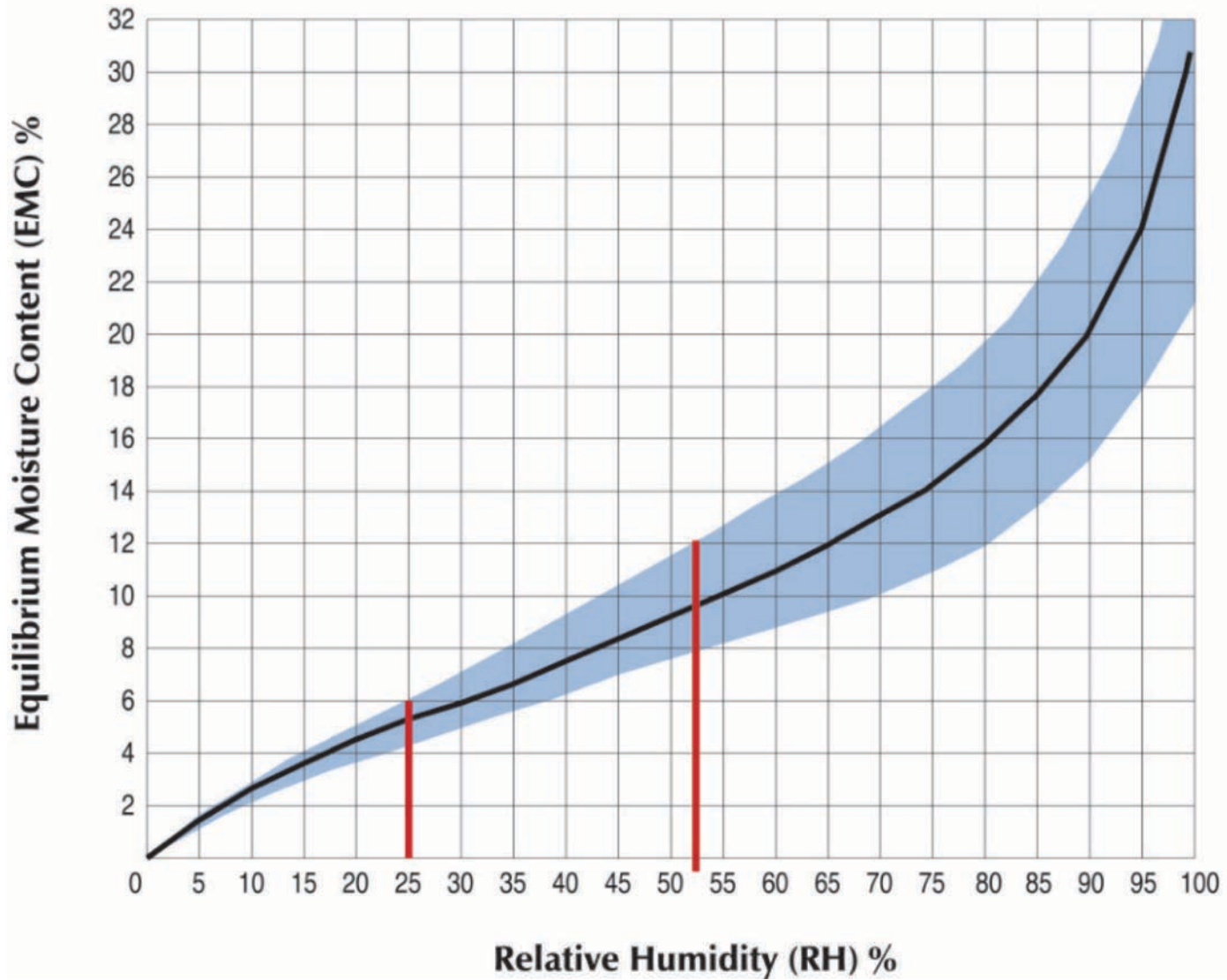






Average sorption isotherm for wood as a function of temperature
 From Straube & Burnett, 2005

Moisture Content vs. Relative Humidity

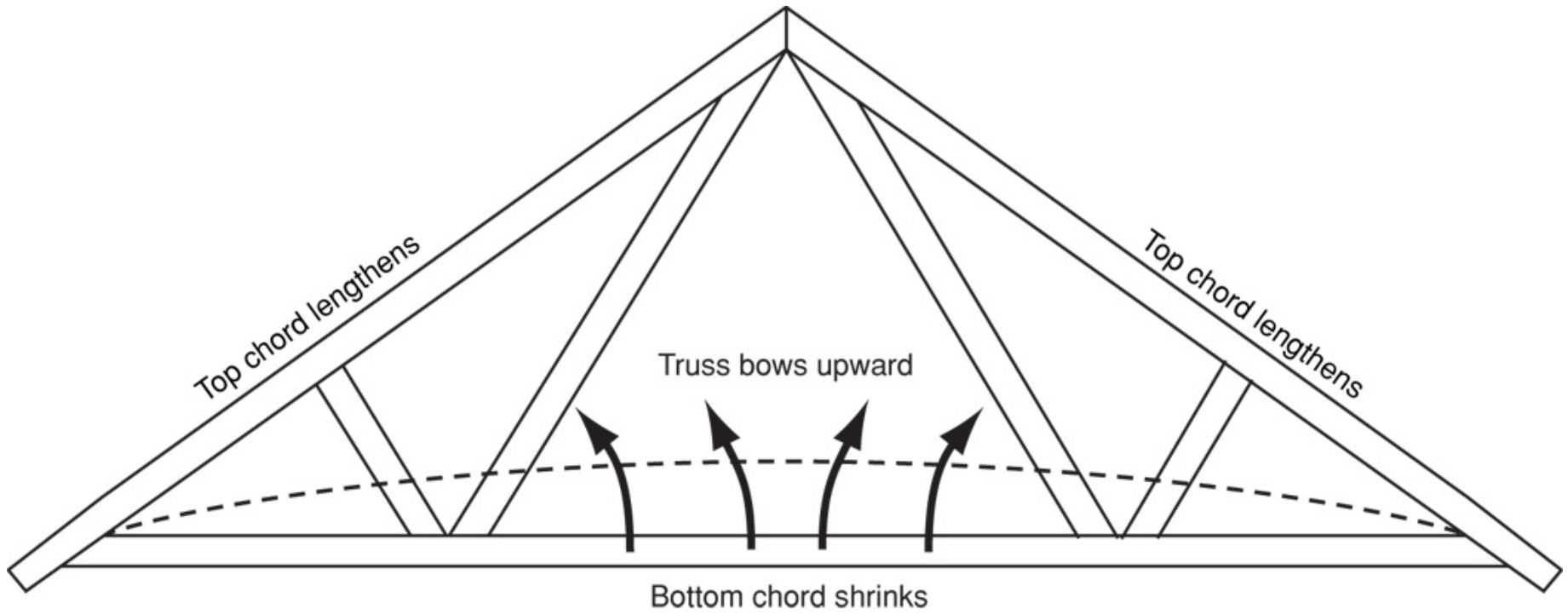


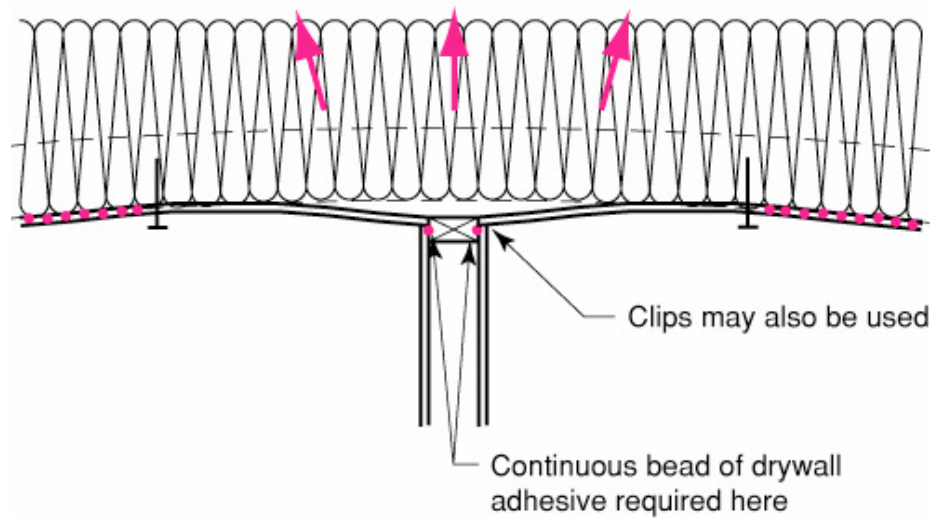
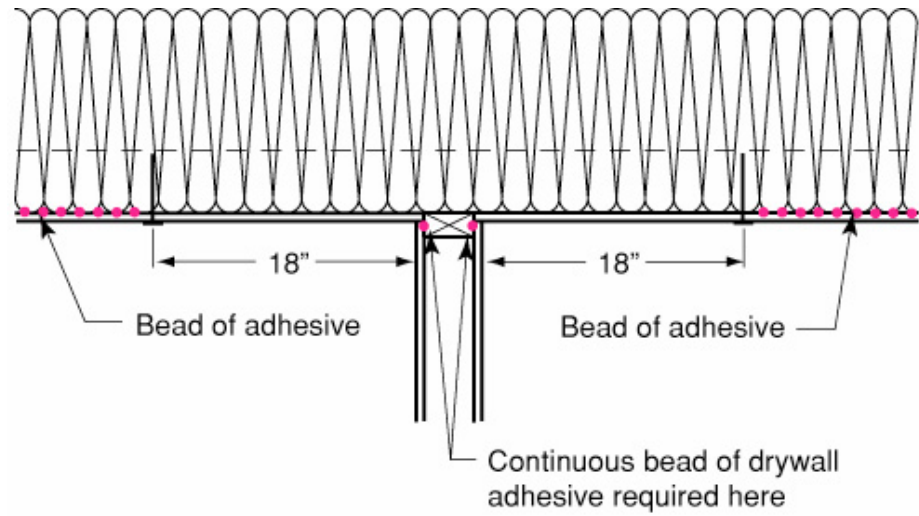


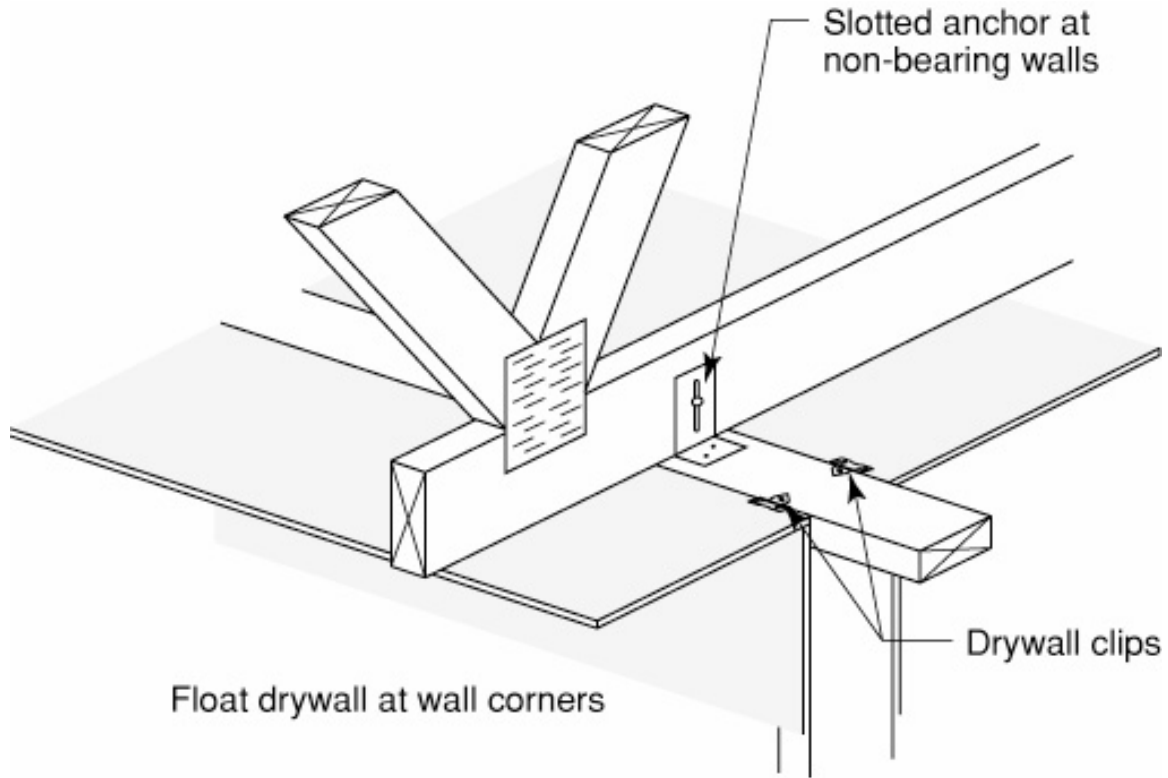








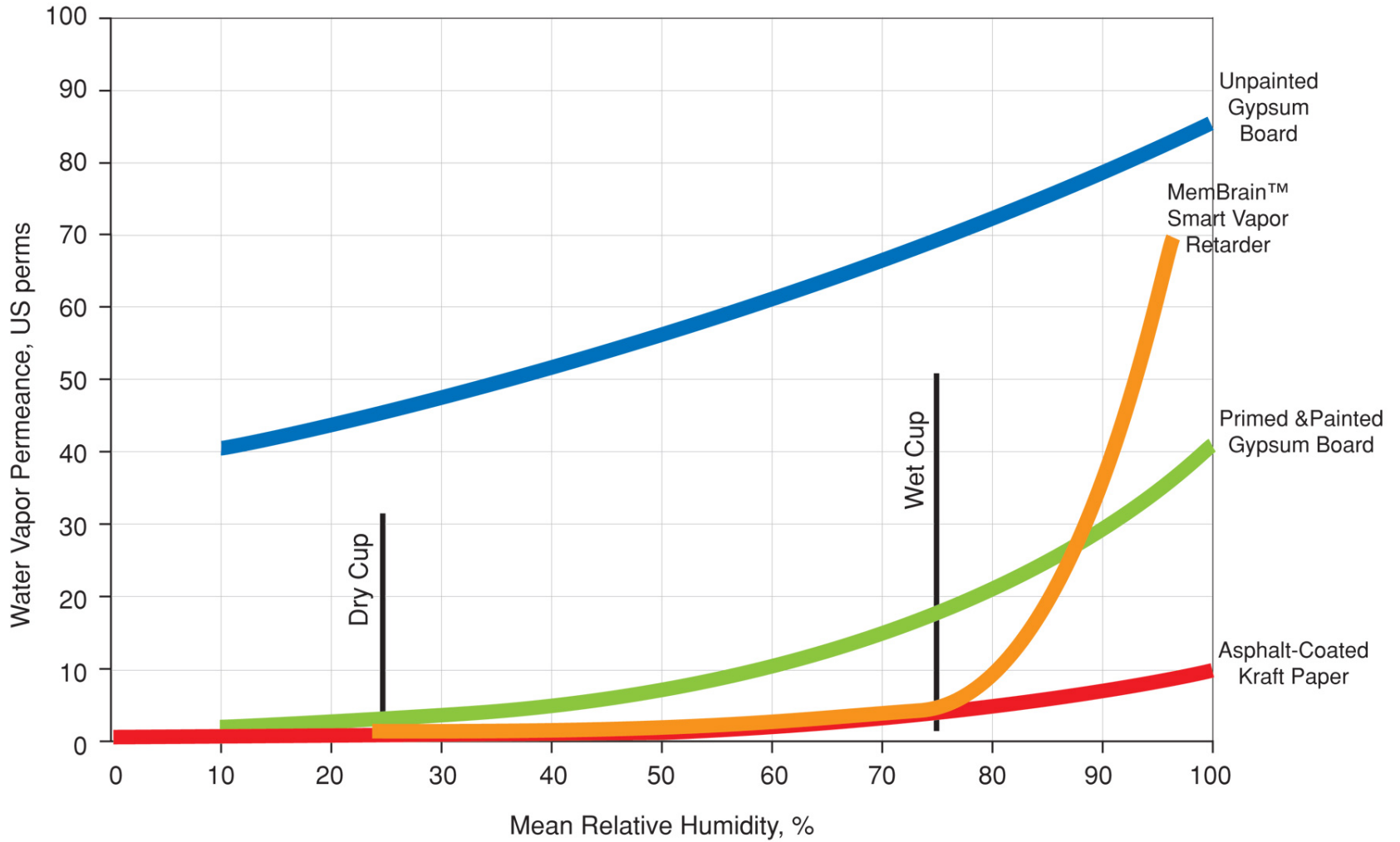




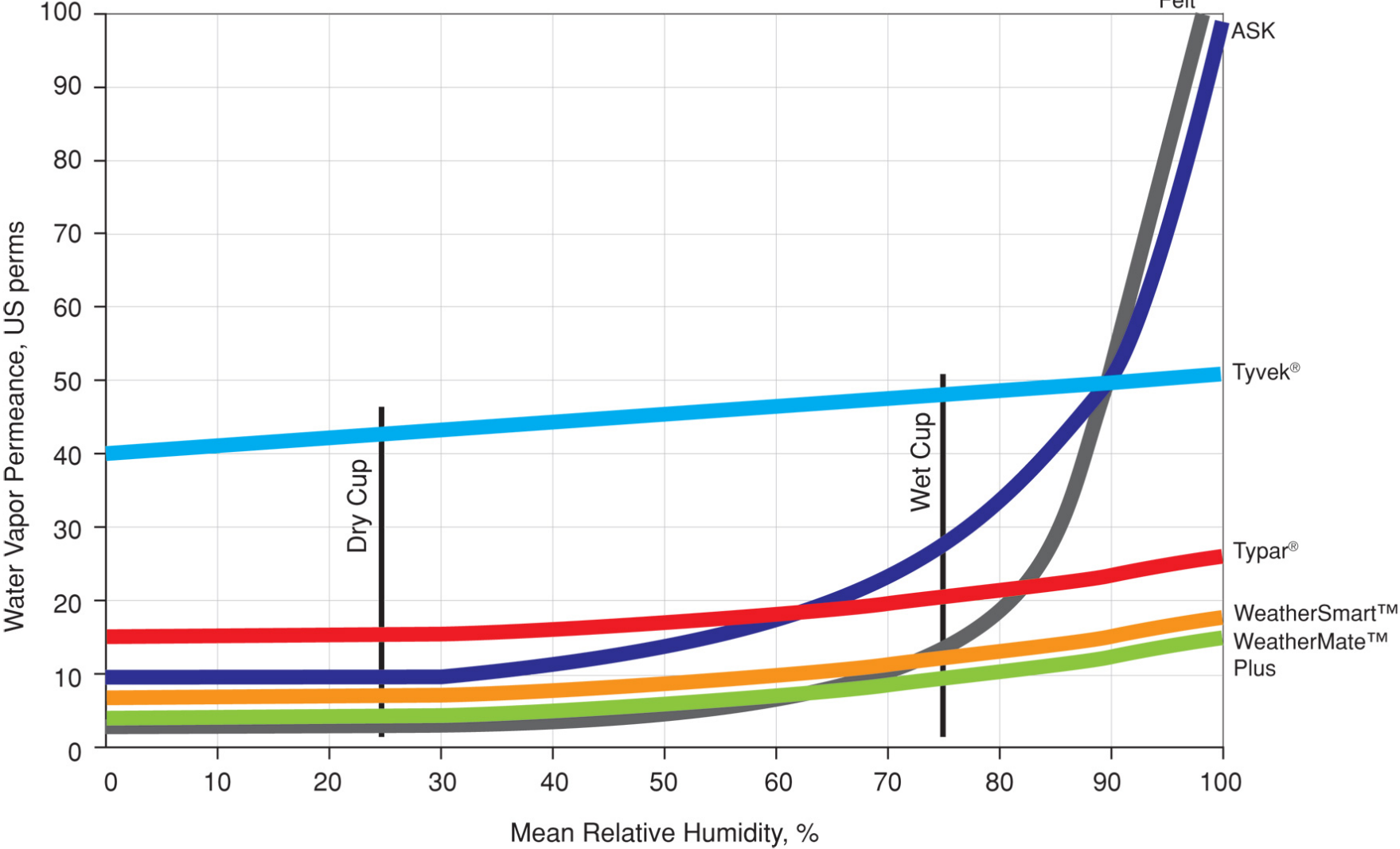




Water Vapor Permeance of MemBrain™ Smart Vapor Retarder, Primed and Painted Gypsum Board, Unpainted Gypsum Board and Asphalt-Coated Kraft Paper



Water Vapor Permeance of WRB's



Laws of Thermodynamics

Zeroth Law – Equal Systems

First Law - Conservation of Energy

Second Law - Entropy

Third Law – Absolute Zero

2nd Law of Thermodynamics

In an isolated system, a process can occur only if it increases the total entropy of the system

Rudolf Clausius

Moisture Flow Is From Warm To Cold
Moisture Flow Is From More To Less

Moisture Flow Is From Warm To Cold
Moisture Flow Is From More To Less

Thermal Gradient – Thermal Diffusion
Concentration Gradient – Molecular Diffusion

Moisture Flow Is From Warm To Cold
Moisture Flow Is From More To Less

Thermal Gradient – Thermal Diffusion
Concentration Gradient – Molecular Diffusion

Vapor Diffusion

Thermodynamic Potential



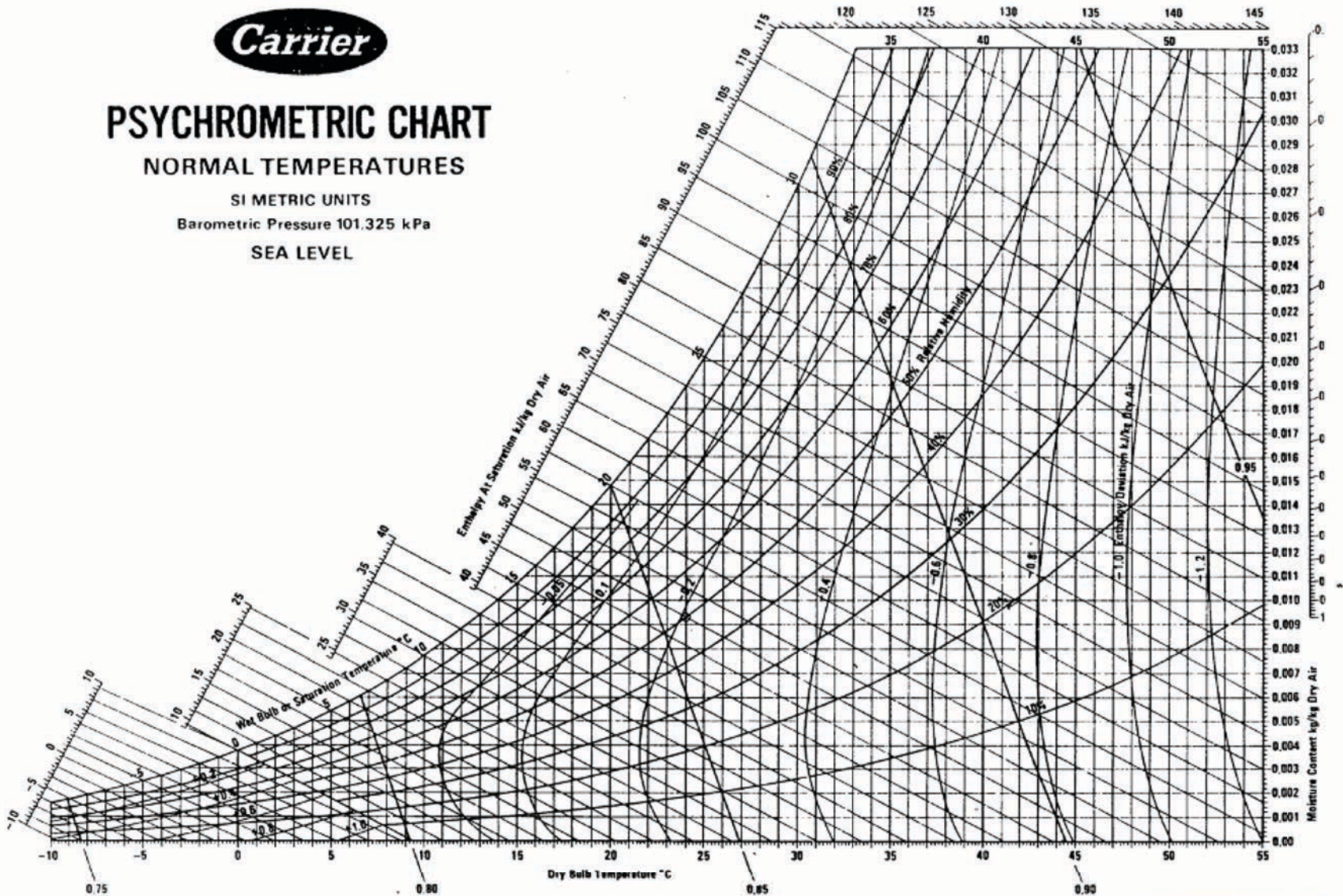
PSYCHROMETRIC CHART

NORMAL TEMPERATURES

SI METRIC UNITS

Barometric Pressure 101.325 kPa

SEA LEVEL



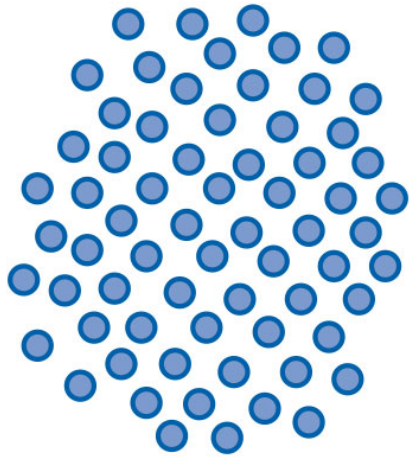
Below 0°C Properties and Enthalpy Deviation Lines Are For Ice

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Cat. No. 794 002 Printed in U.S.A.

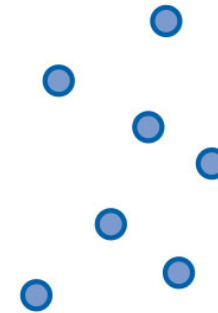
Vapor

Diffusion
Convective Flow

Vapor Concentration
Air Pressure



DIFFUSION

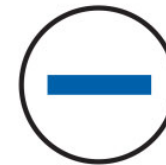


**Higher Dewpoint Temperature
Higher Water Vapor Density
or Concentration
(Higher Vapor Pressure)
on Warm Side of Assembly**

**Low Dewpoint Temperature
Lower Water Vapor Density
or Concentration
(Lower Vapor Pressure)
on Cold Side of Assembly**

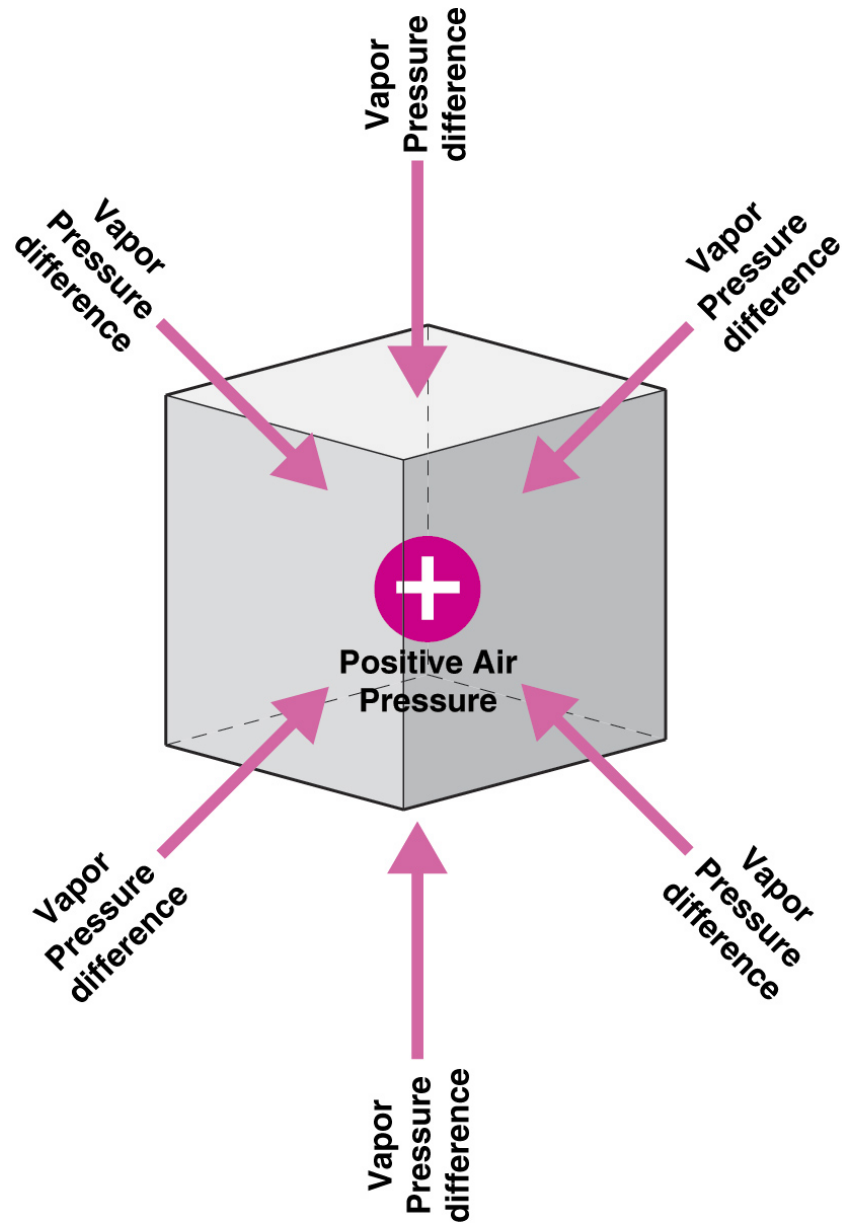


AIR TRANSPORT



**Higher Air
Pressure**

**Lower Air
Pressure**

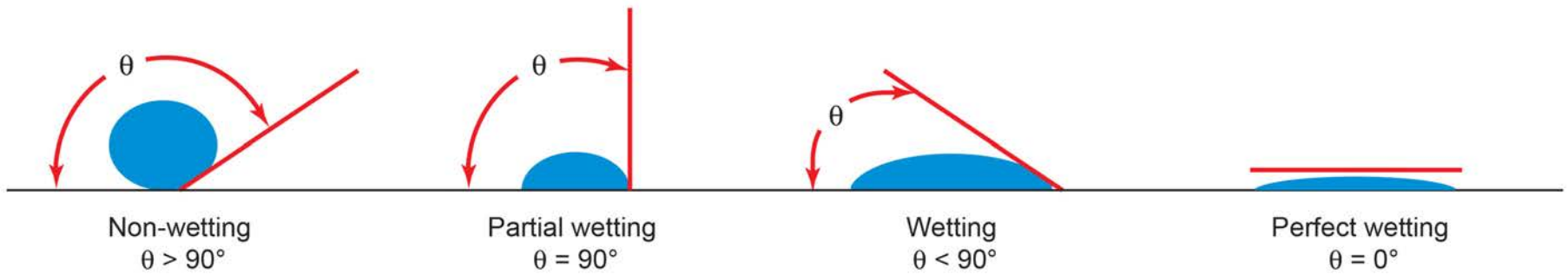


How Does Wetting Occur?



- “non-wetable” surface
- water repellent surface
- hydrophobic surface
- water more attracted to itself than to surface
- surface energy of water greater than surface energy of surface
- water “beads up”
- “greasy” surface
- high contact angle “ θ ”

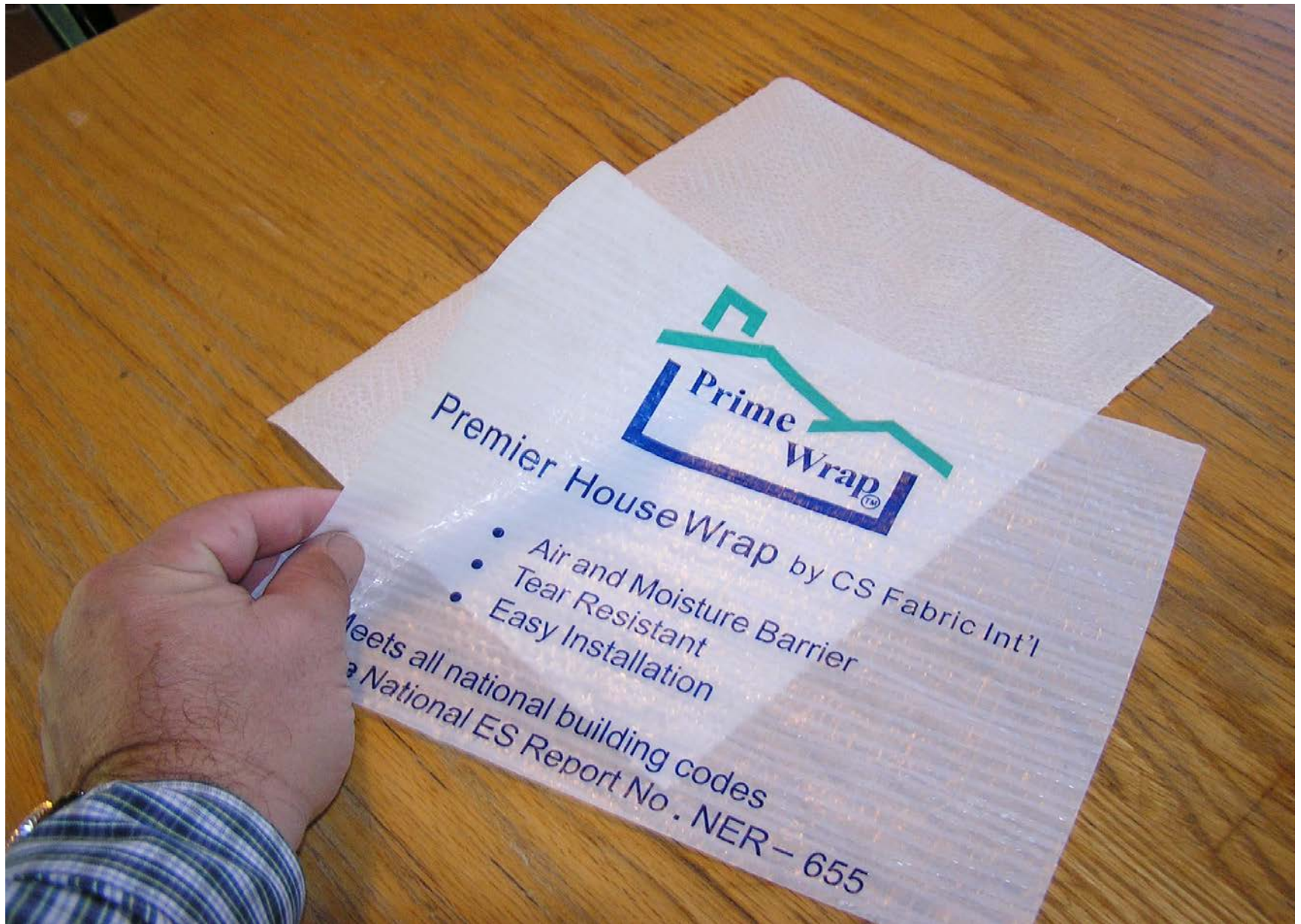
- “wetable” surface
- non-water repellent surface
- hydrophilic surface
- water more attracted to surface than itself
- surface energy of surface greater than surface energy of water
- water “spreads out”
- “non-greasy” surface
- low contact angle “ θ ”

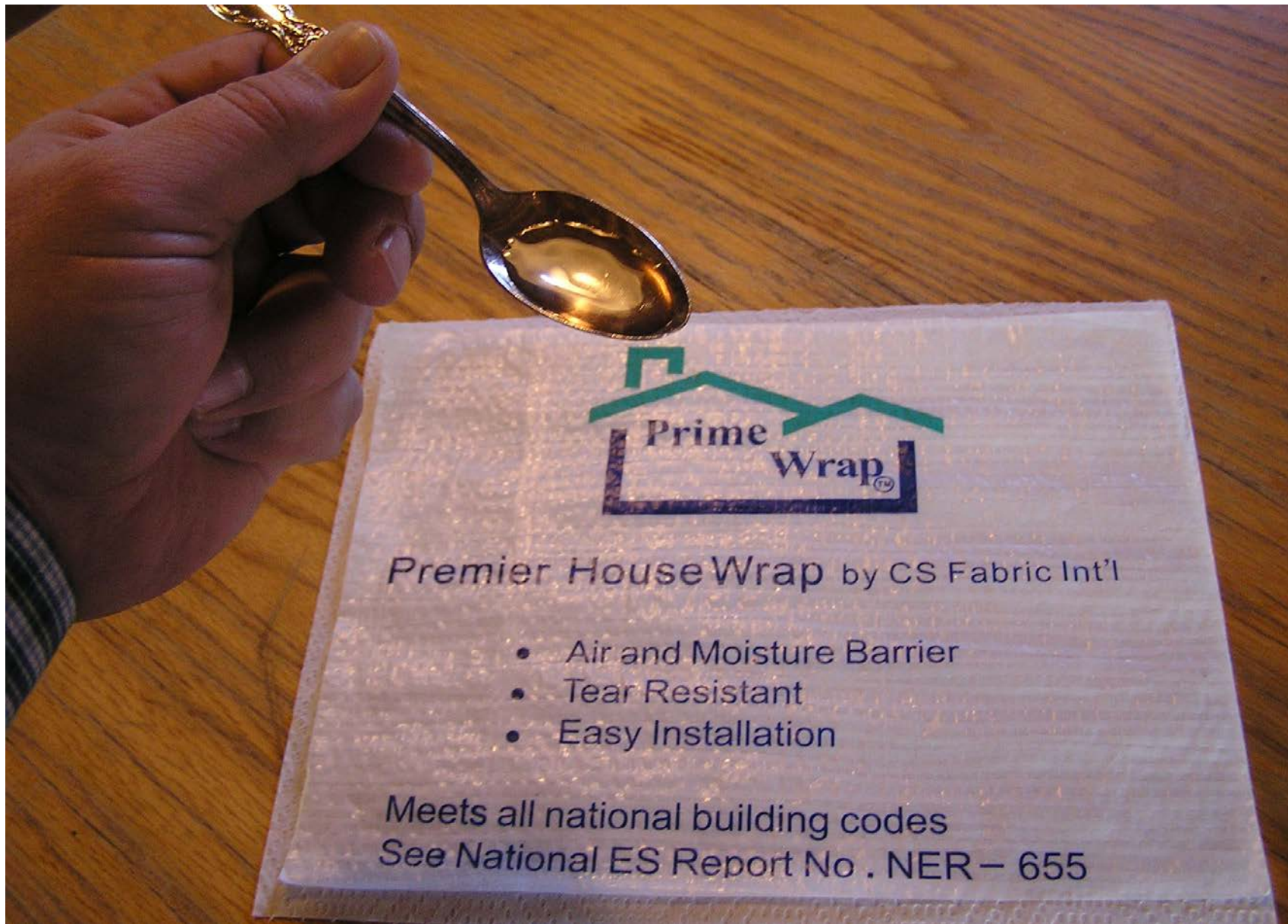










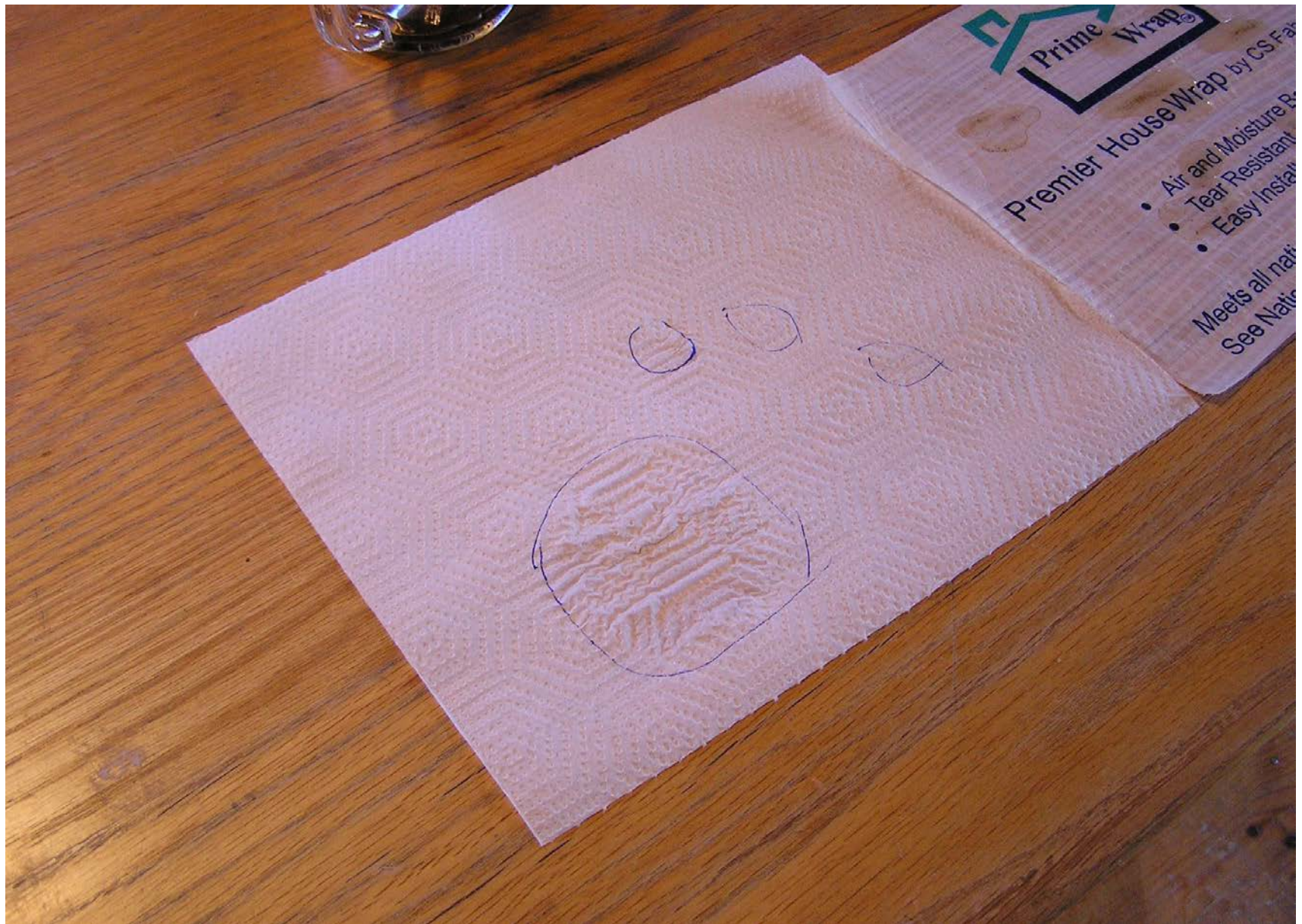














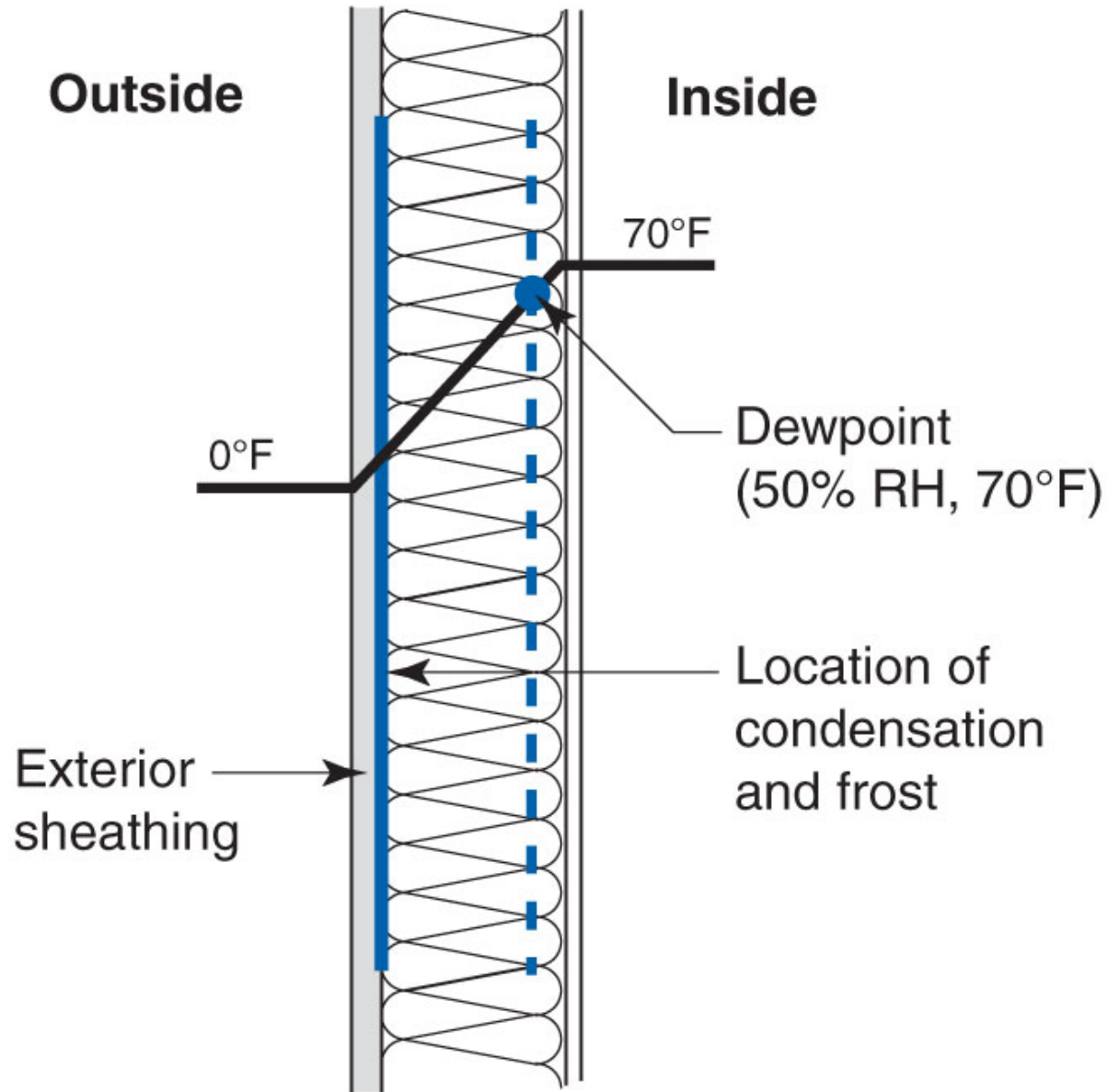




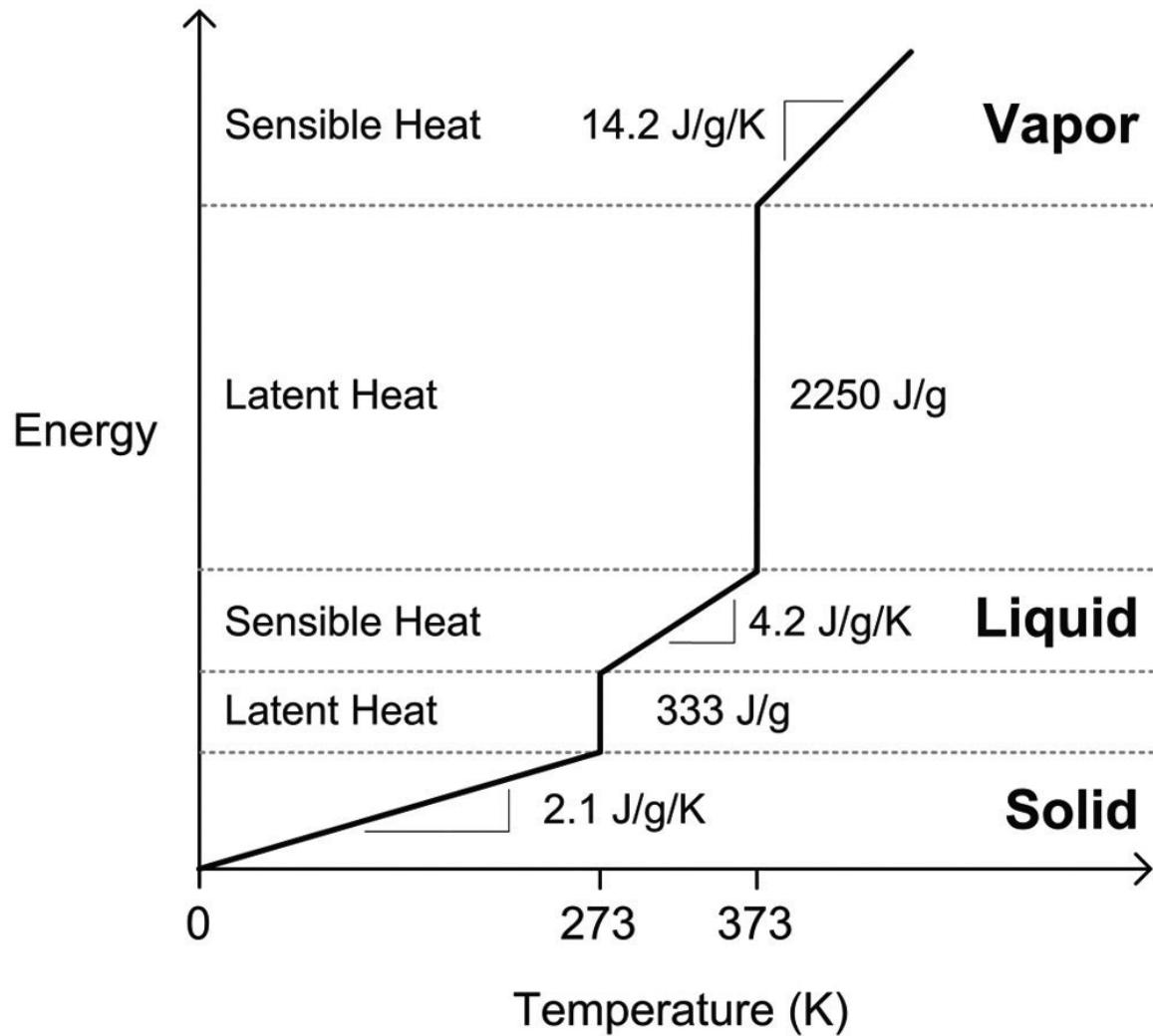
Surface Energy

Water (20 C)	73 dynes/cm
Water (100 C)	59 dynes/cm
Epoxy	46 dynes/cm
Polyethylene	31 dynes/cm
Soapy water	30 dynes/cm
Paraffin wax	25 dynes/cm
Silicone	24 dynes/cm
Teflon	18 dynes/cm

When Phases Change





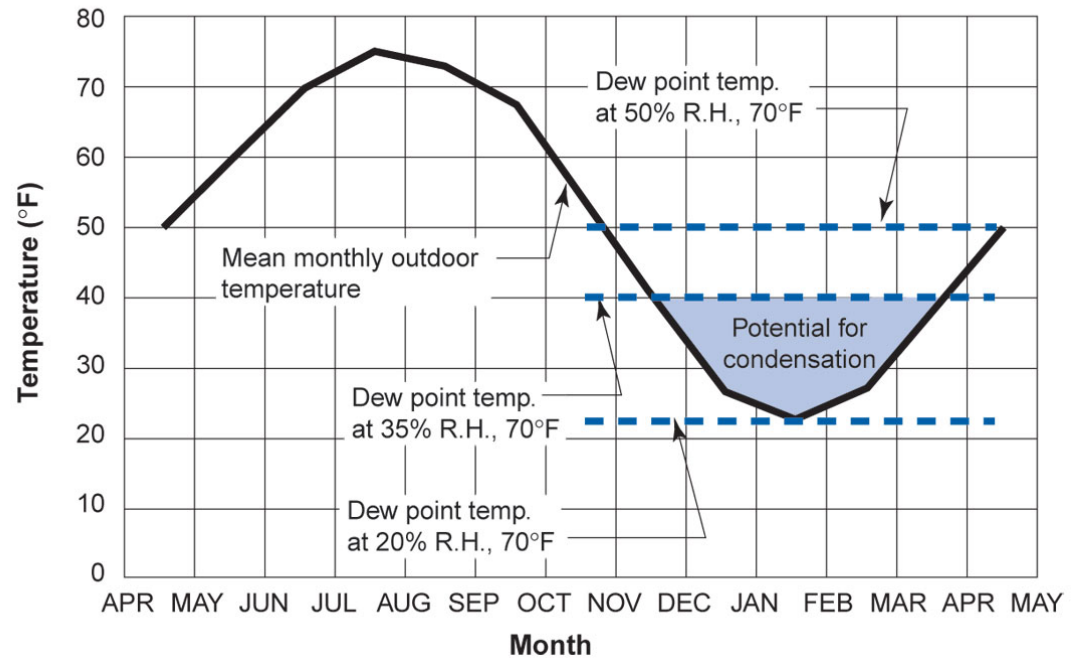
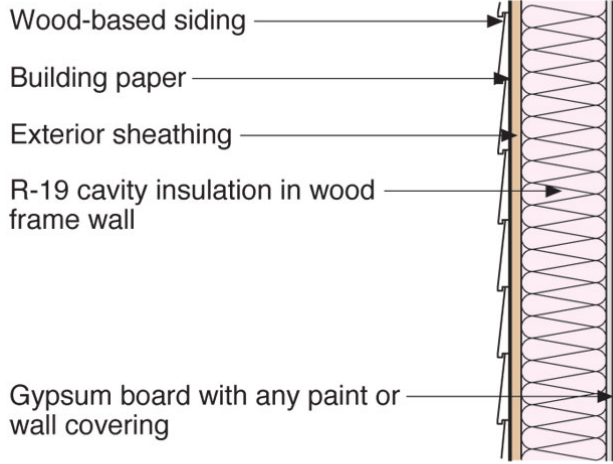


Simple linearized energy-temperature relation for water

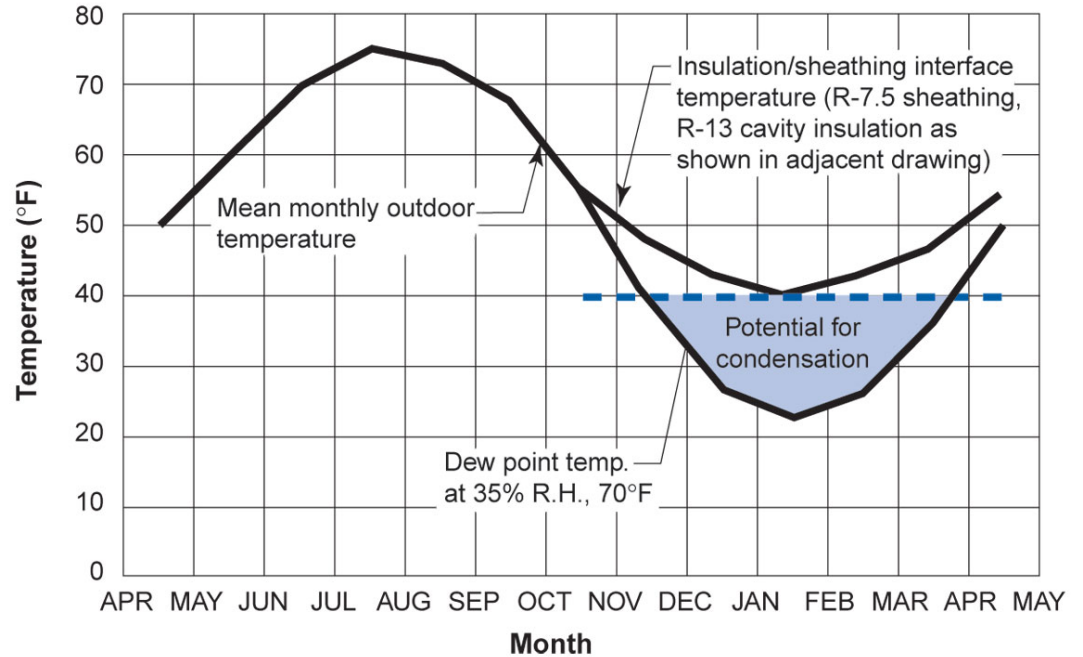
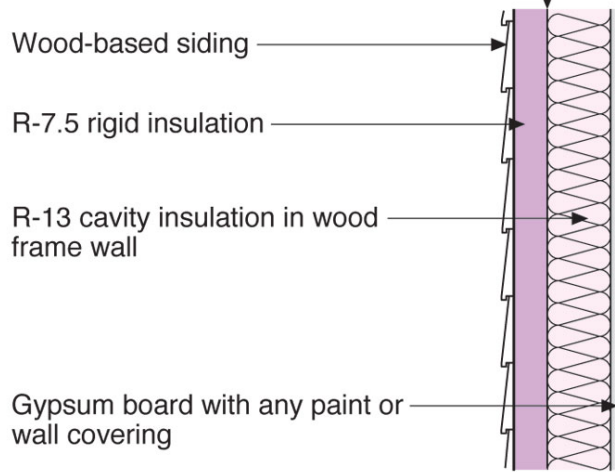
From Straube & Burnett, 2005



The inside face of the exterior sheathing is the condensing surface of interest



The inside face of the insulating sheathing is the condensing surface of interest



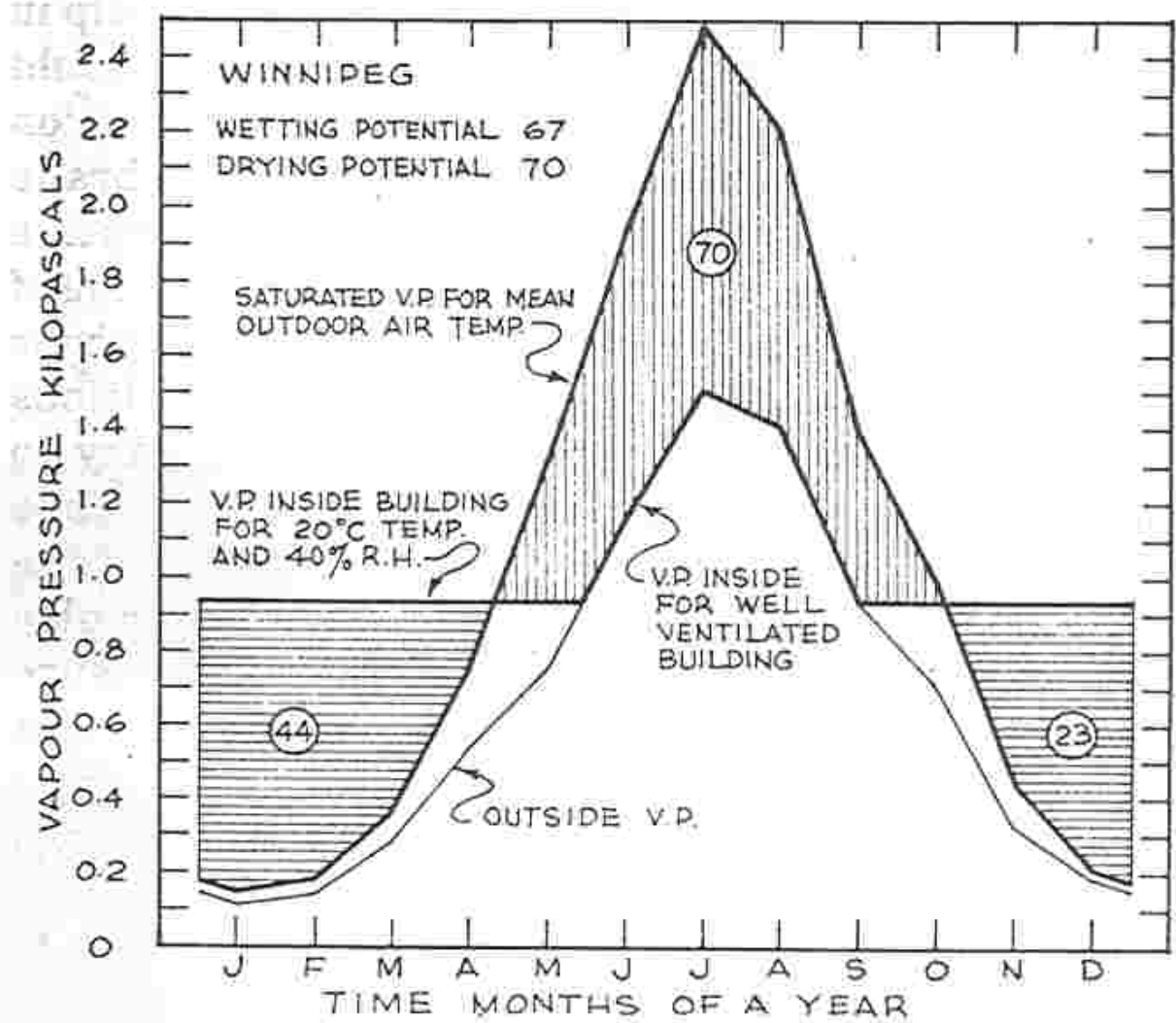
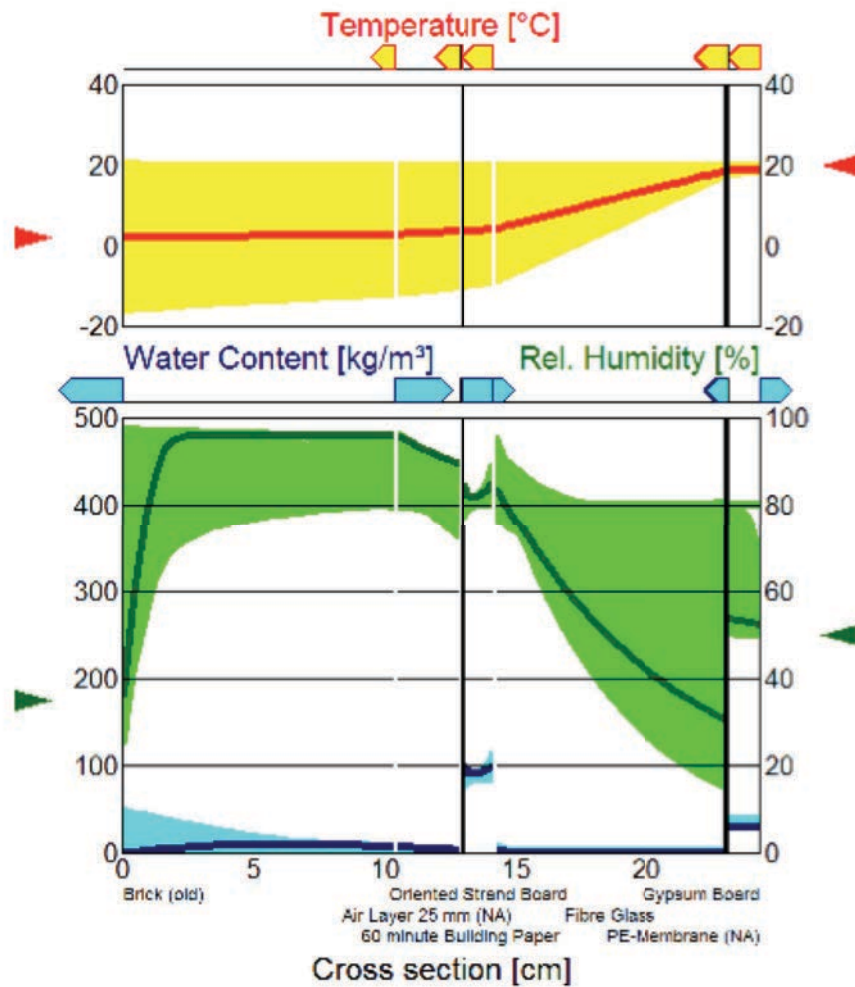


Figure 8-7. Outside vapour pressure, saturated vapour pressure and inside vapour pressure for Winnipeg.



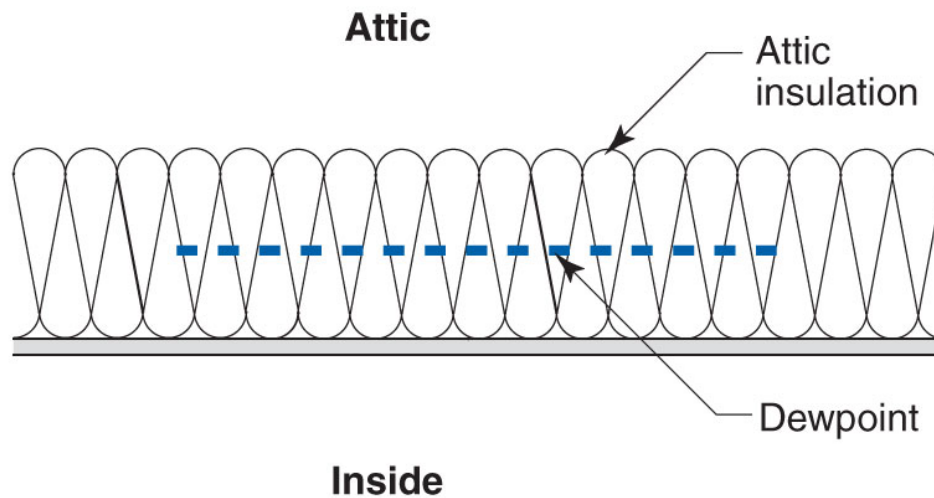
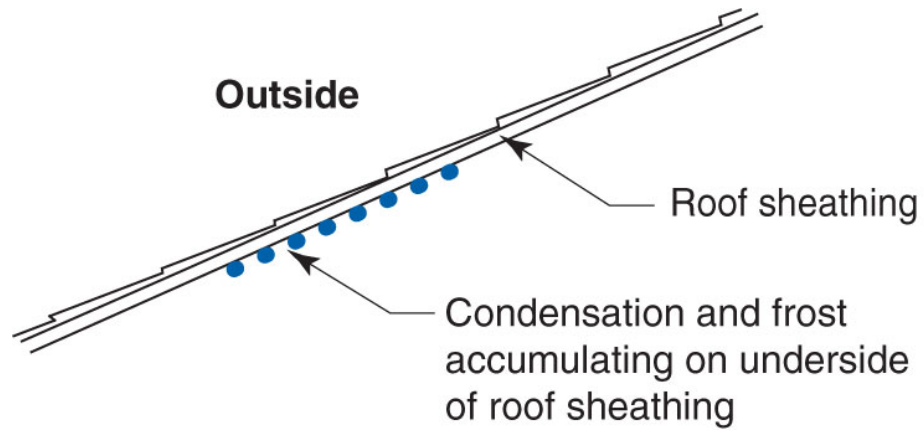
WUFI® 3.3 Pro. IBP
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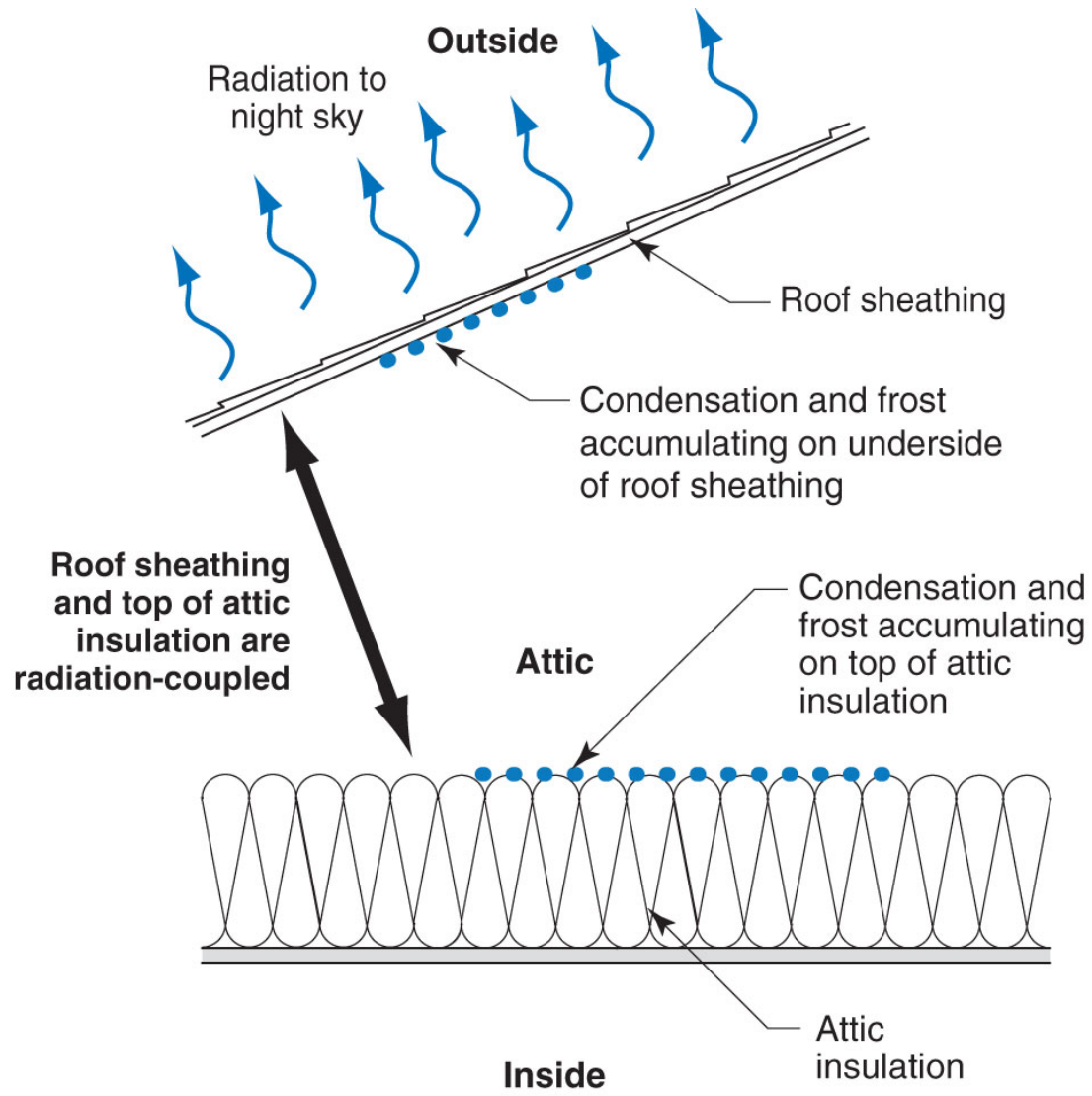
0% 100%

0% 100%

? <- > - > <- >





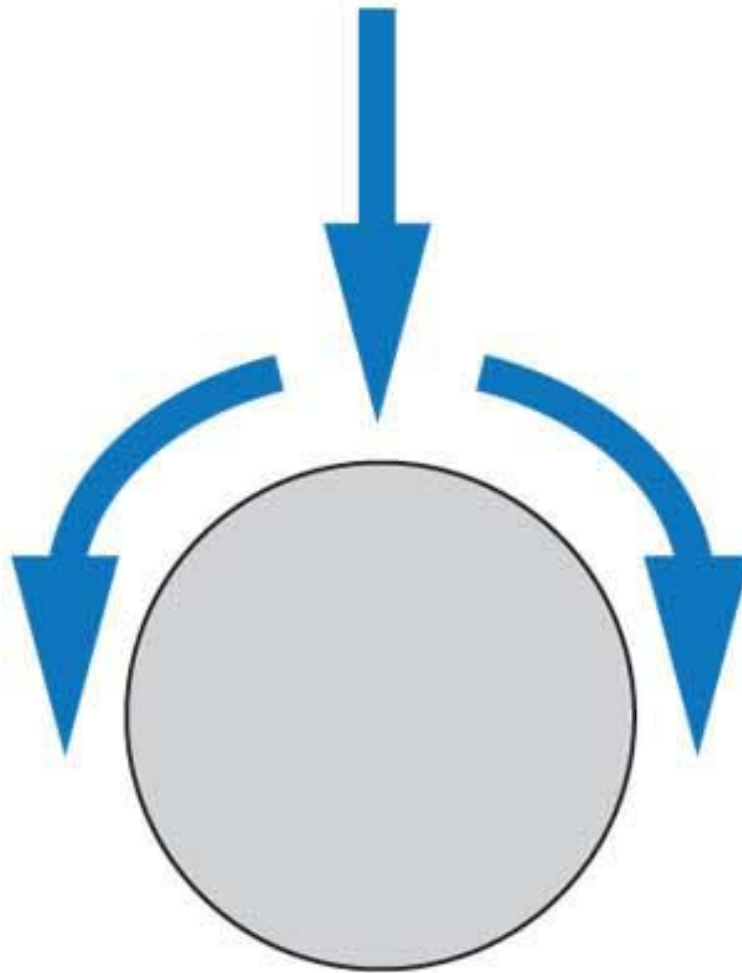


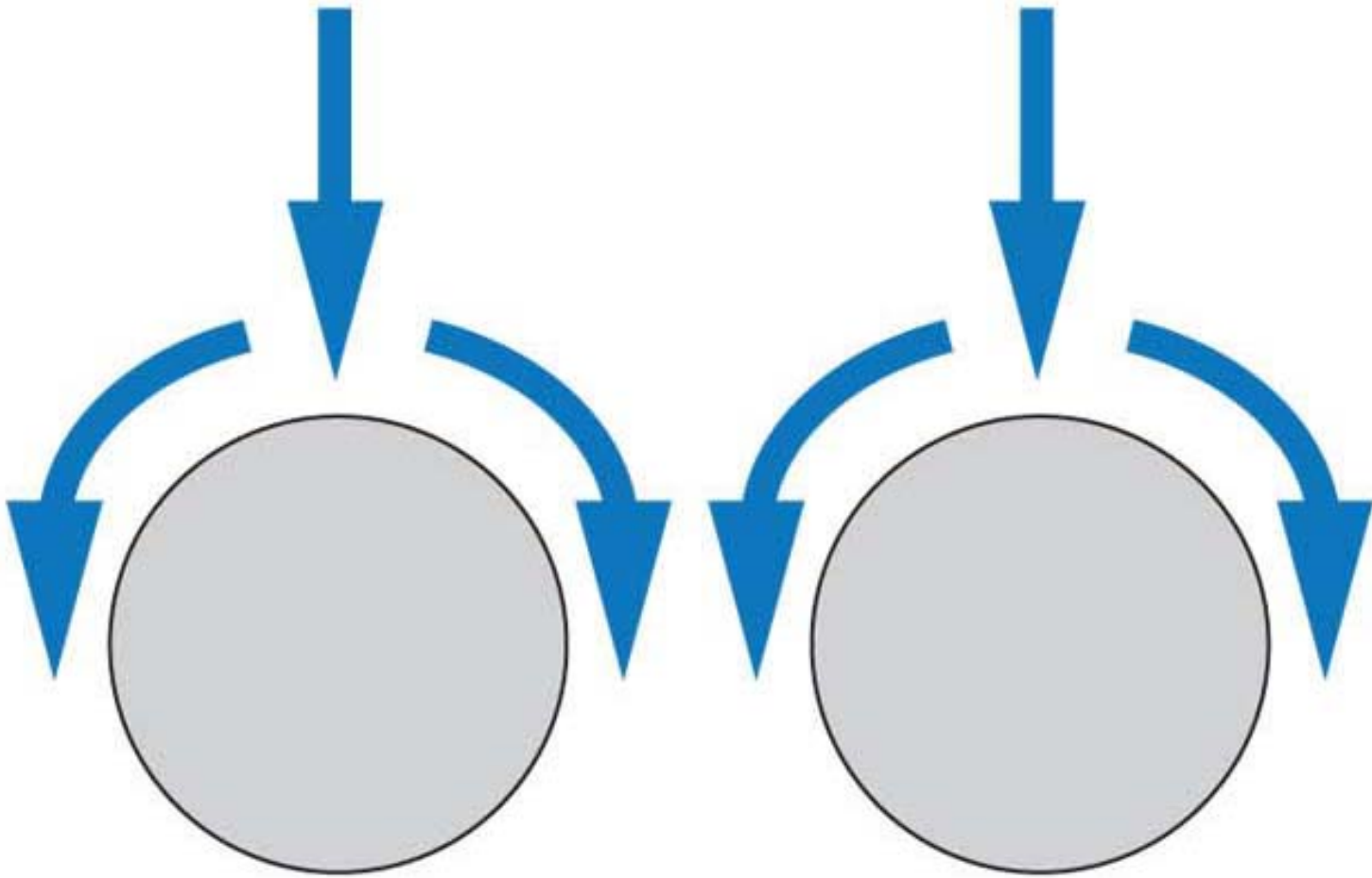


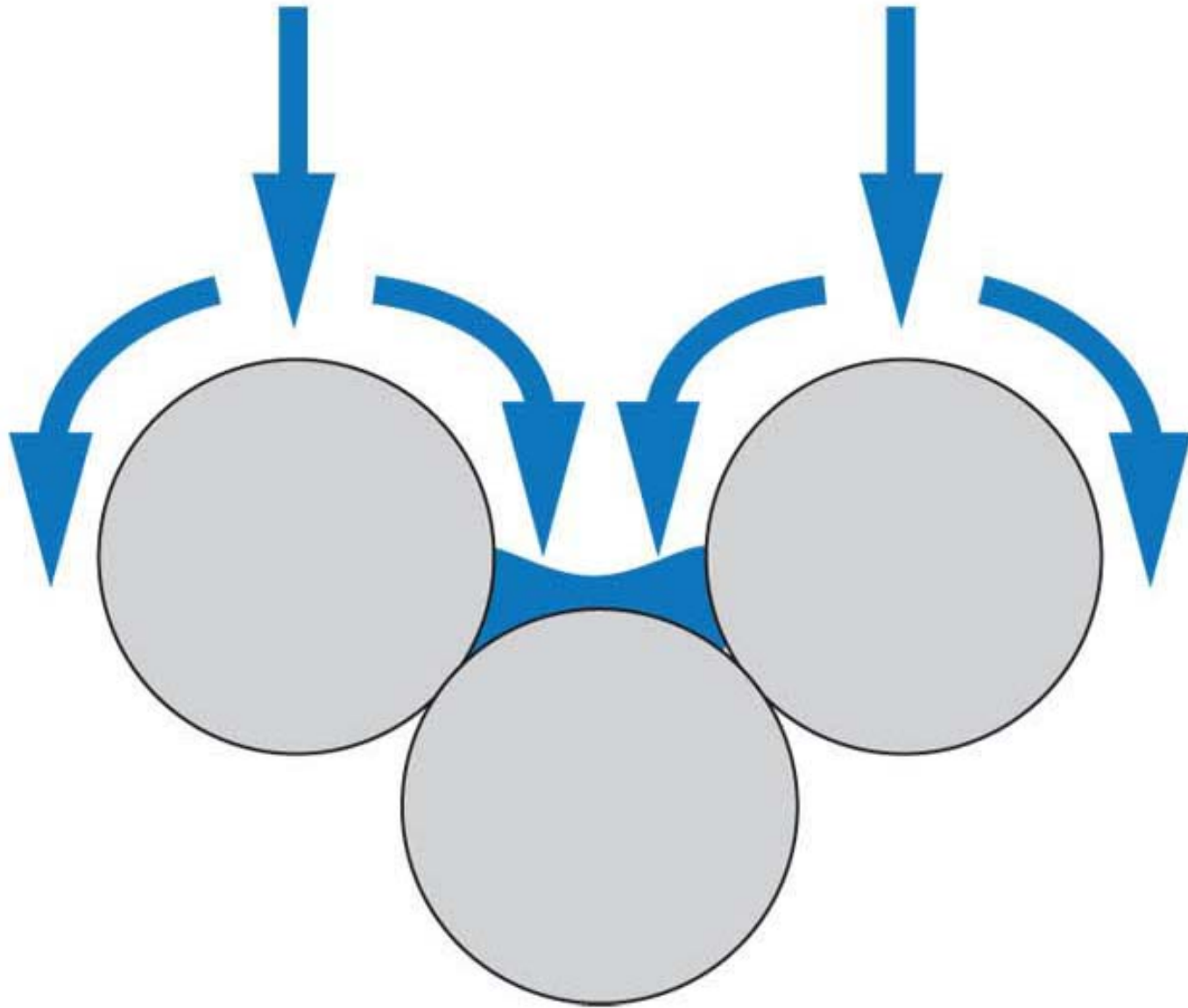


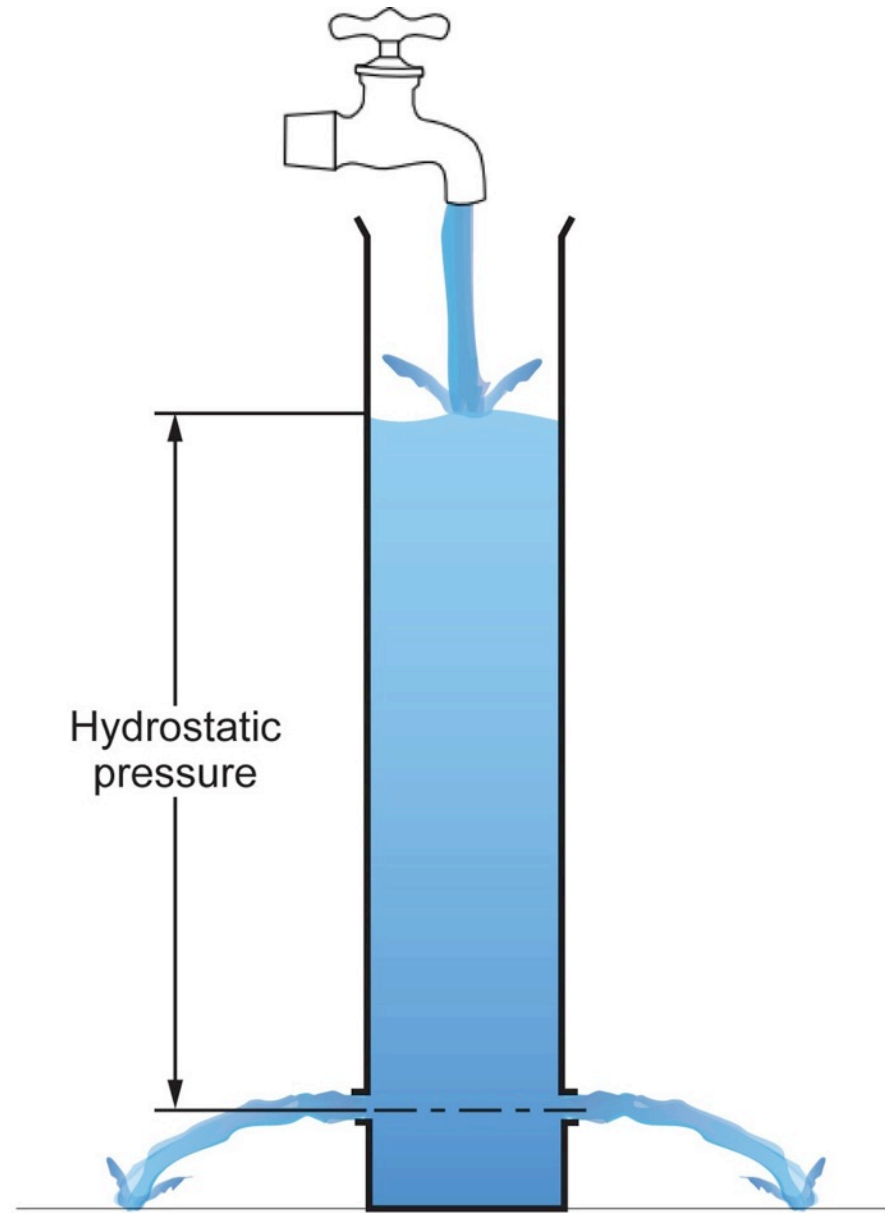
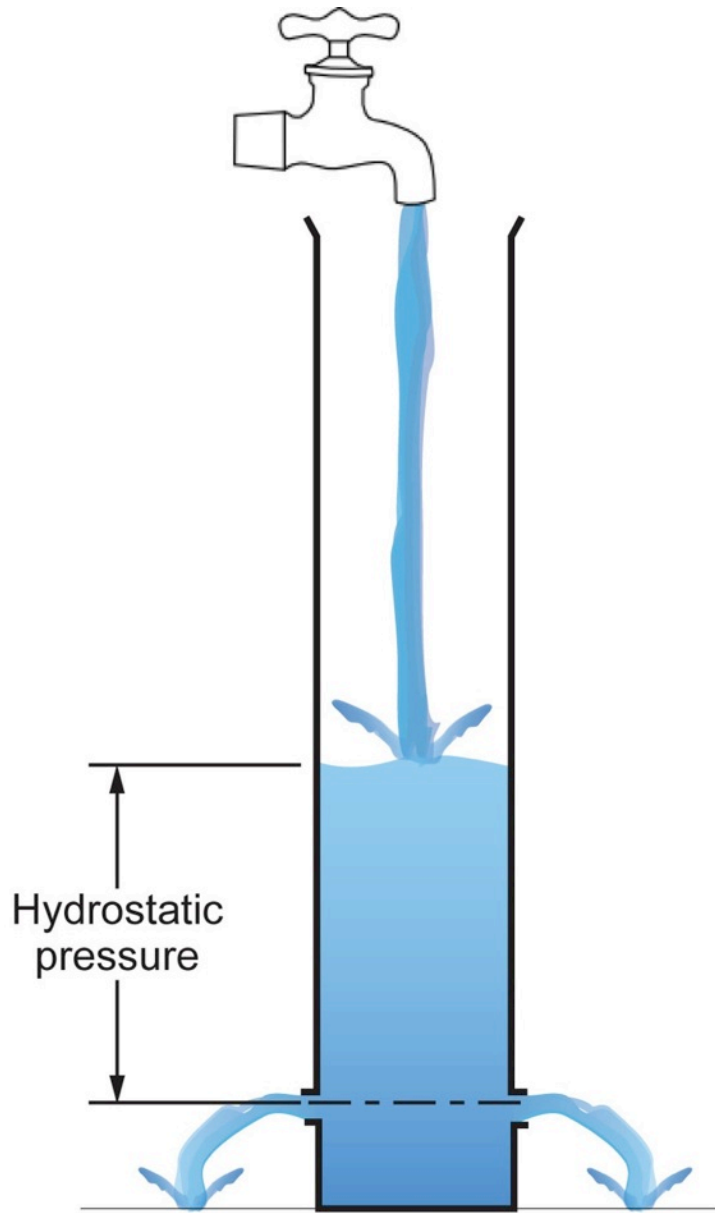
The Best For The Last....

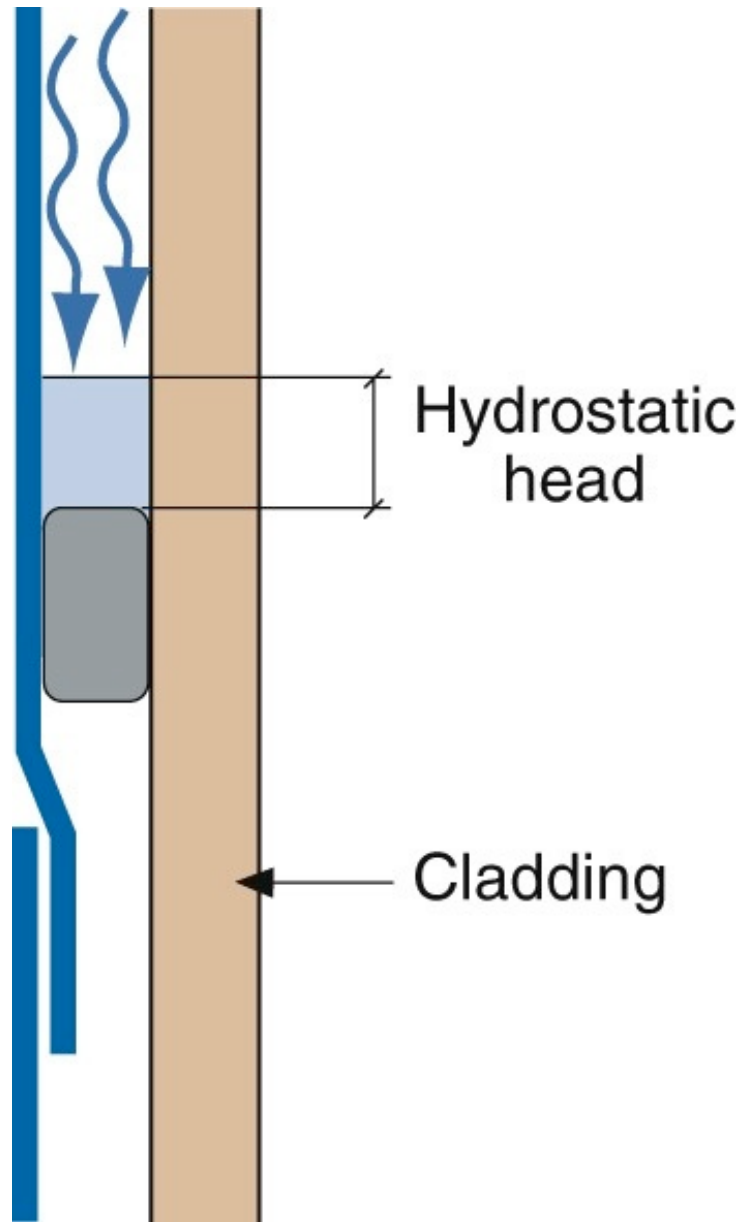


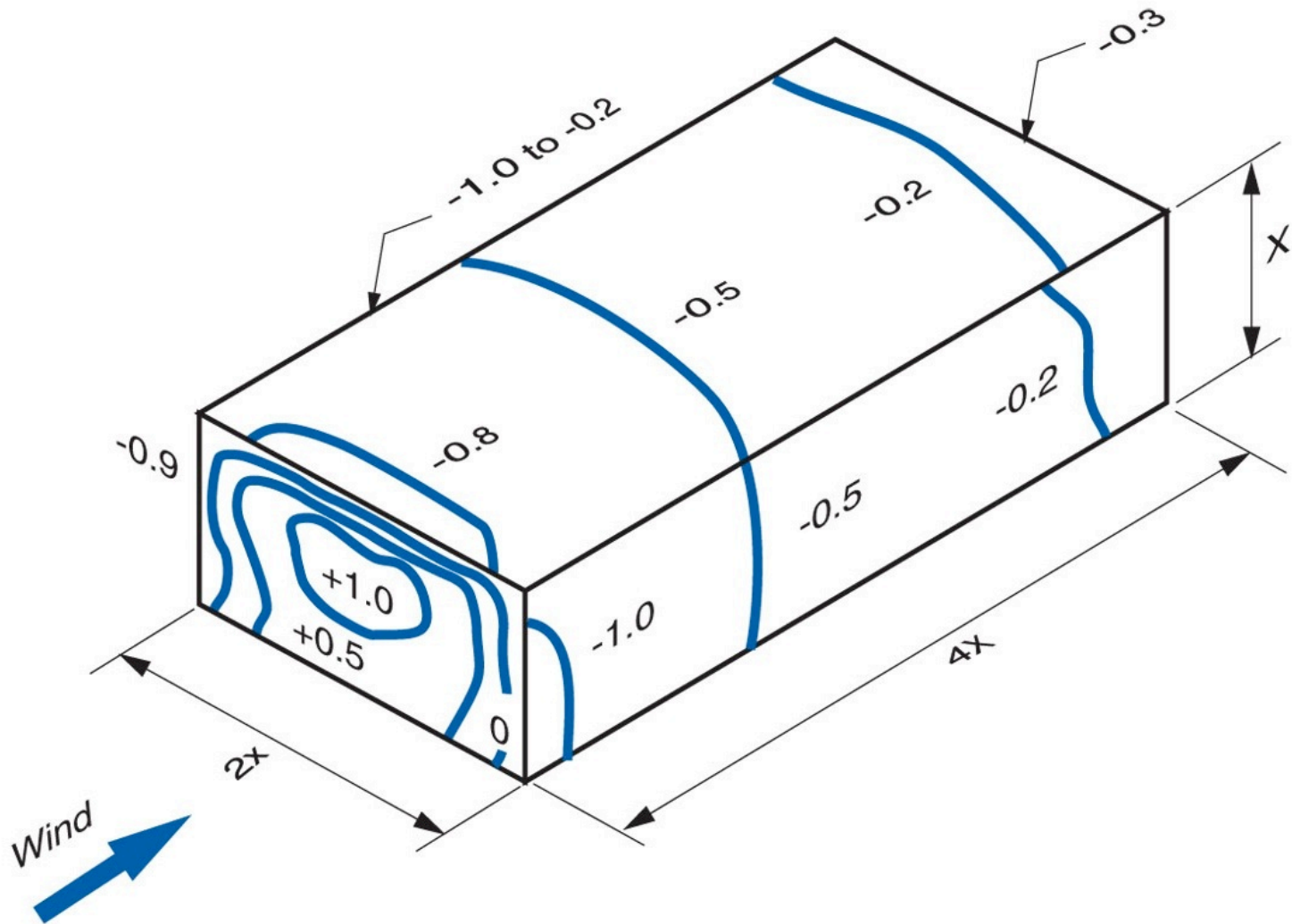






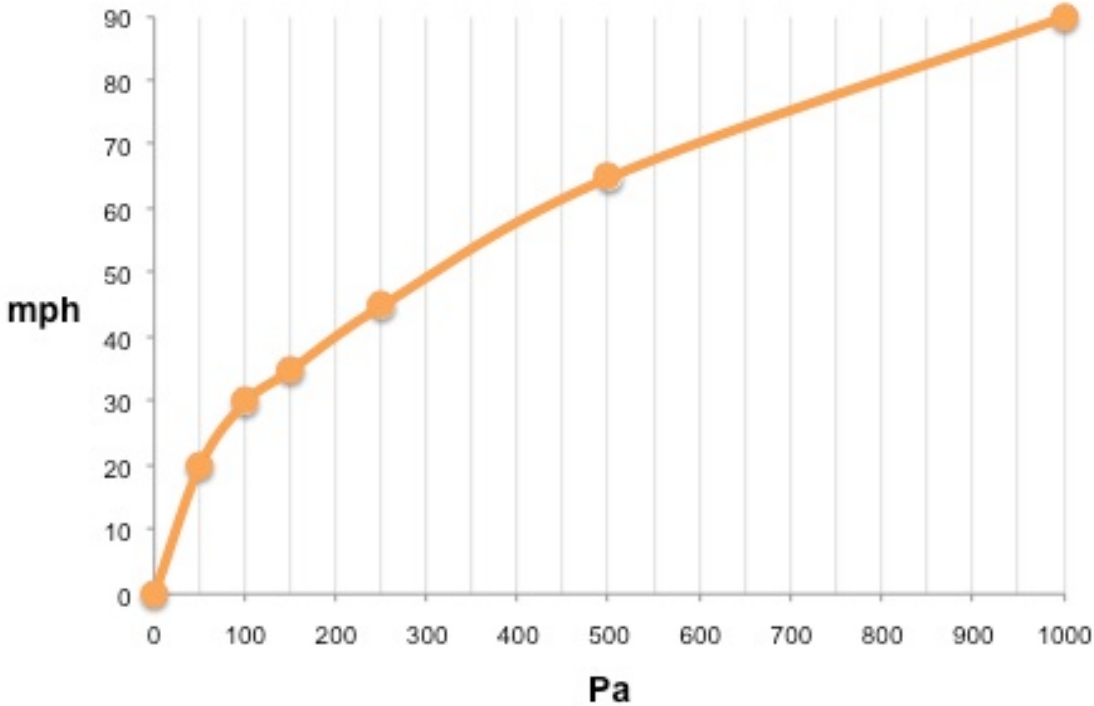






Pascals	mph
50	Pa = 20 mph
100	Pa = 30 mph
150	Pa = 35 mph
250	Pa = 45 mph
500	Pa = 65 mph
1,000	Pa = 90 mph

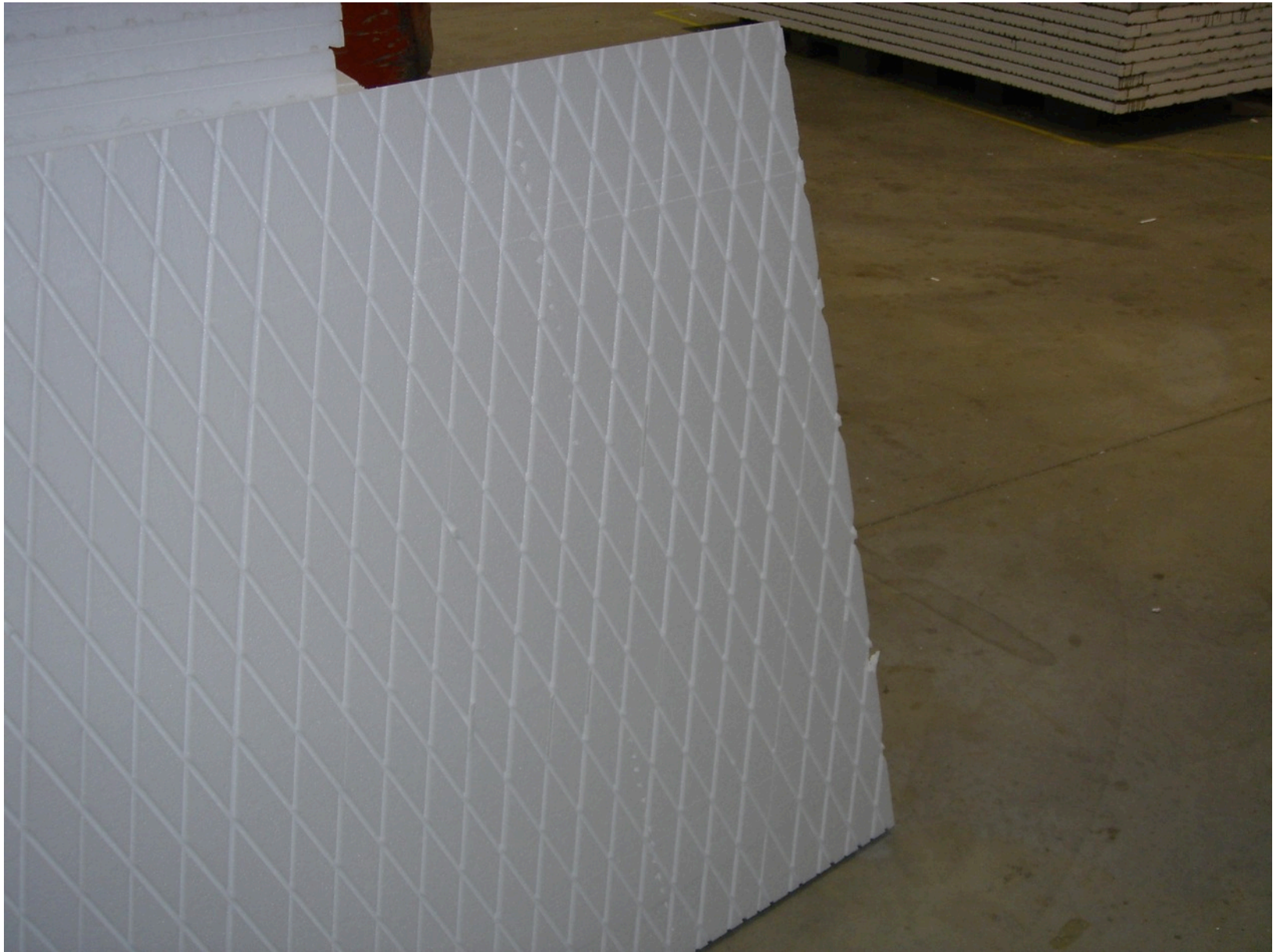
Wind Speed (mph) vs. Stagnation Pressure (Pa)



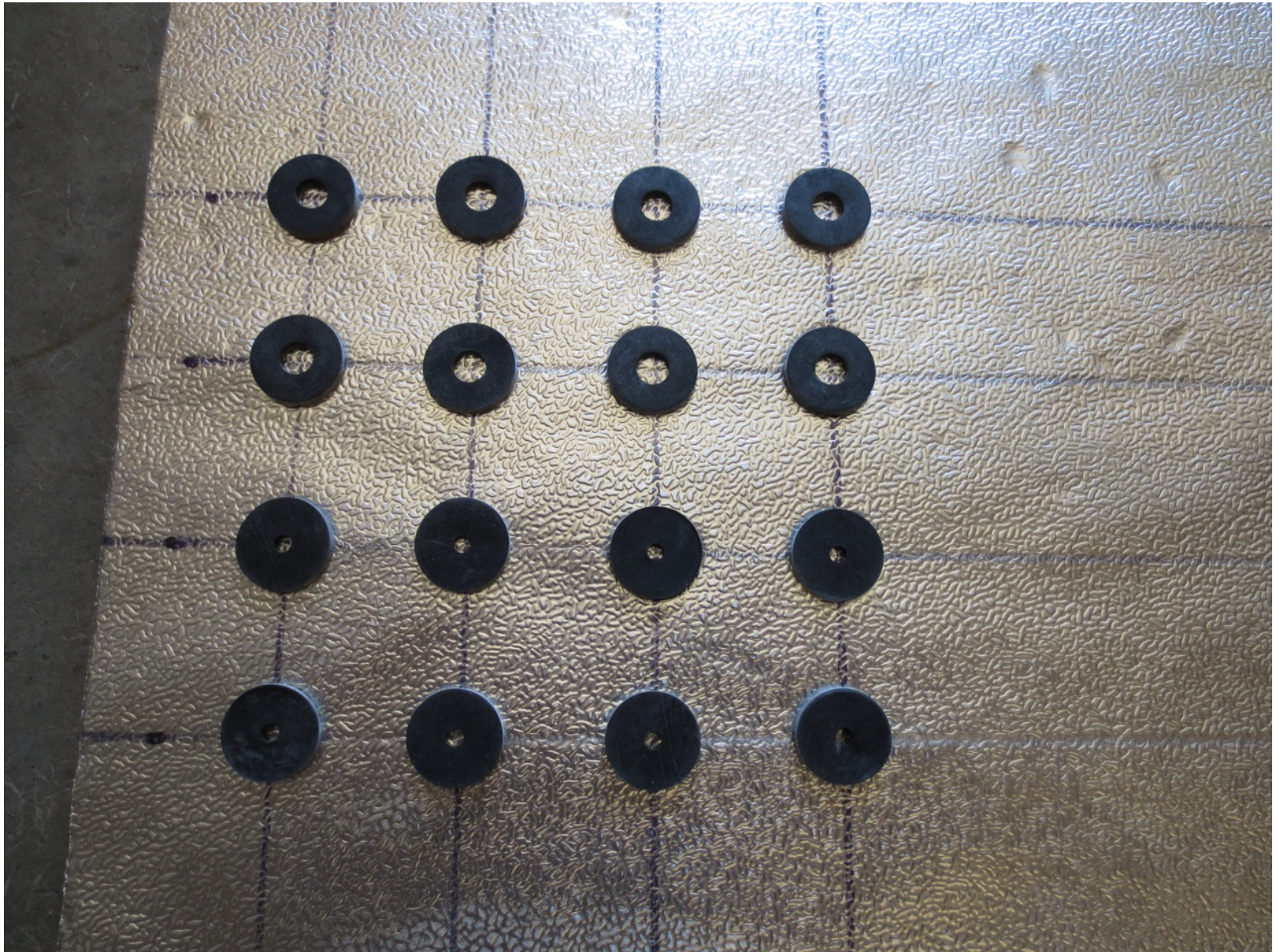








Rain Screen



Beer Screen?

